

# Characterization of ARD Neutralization Sludge: Links between ARD Influent Chemistry, Treatment Process and Sludge Composition.

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Alan J. Martin, Lorax Environmental Services Ltd.

Skya Fawcett, Lorax Environmental Services Ltd.

Andrew Rollo, Canadian Environmental Assessment Agency

Diana Loomer, University of New Brunswick

Tom Al, University of New Brunswick



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- Goldcorp Canada – Equity Mine (BC)
- Xstrata Zinc– Geco Mine (ON) and Brunswick Mine (NB)
- EPCOR - Britannia Mine (BC)
- Inmet - Samatosum Mine (BC)
- Teck - Sullivan Mine (BC)
- Hudson Bay Mining and Smelting – Chisel North Mine (MB)

## Reviews:

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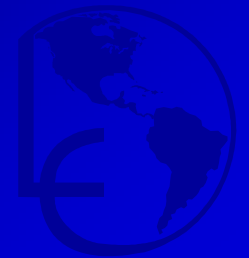


# PREFACE

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Neutralization Sludges:

*They're not what you think they are.....*



# BACKGROUND

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- At many mine sites the management of ARD involves neutralization with lime to reduce levels of acidity and trace elements prior to discharge to the environment.
- In general, treatment is achieved by raising the pH to values greater than 8.5 and separating the resulting precipitates (sludge) from the treated water prior to discharge.
- Within Canada alone, it is estimated that approximately 7 million m<sup>3</sup> of neutralization sludge are produced on an annual basis from mining/metallurgical operations (MEND, 1997).



## BACKGROUND cont.

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- In general, trace-element associations (i.e., metal-bearing phases) within lime neutralization sludge are not well understood.
- Due to the extremely fine-grained and often amorphous (*i.e.*, non crystalline) character of sludge solids, metal associations have been difficult to elucidate.
- Given that neutralization sludges represent large inventories of metal-rich material, understanding sludge composition is prerequisite for assessing the long-term chemical stability in various depositional settings (pH and Eh dependent solubility).



# OBJECTIVES

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- To elucidate the trace metal associations (phase relationships) in sludge materials using high resolution microscopy methods; and
- To assess the links between sludge composition, ARD influent chemistry and treatment process;



Provide basis from which to assess long-term chemical stability.



# METHODS - Sludge Samples

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## Sludges from 7 mines across Canada

1. Equity Mine (B.C.) - Goldcorp Canada
2. Geco Mine (Ontario) – Xstrata Zinc
3. Britannia Mine (BC) - EPCOR
4. Brunswick Mine (New Brunswick) – Xstrata Zinc
5. Samatosum Mine (BC) - Inmet
6. Sullivan Mine (BC) - Teck
7. Chisel North Mine (Manitoba)- Hudson Bay Mining and Smelting



# METHODS - Sludge Samples

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- } Focus of today's presentation





# Geology and Treatment Process

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|                   | Equity                            | Geco   | Britannia           |
|-------------------|-----------------------------------|--|---------------------|
| Commodities       | Au-Ag-Cu                          | Cu-Zn  | Cu (Zn-Pb-Ag)       |
| Deposit Type      | Intrusion-related<br>hydrothermal | Metamorphosed<br>Volcanogenic<br>Massive Sulfide | Massive Sulfide     |
| Treatment Process | High density sludge               | High density sludge                              | High density sludge |

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HDS: typically 20-30% solids



# METHODS – SLUDGE ANALYSIS

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## Solid Phase:

- Elemental Abundance
- X-Ray Diffraction
- High-resolution microscopy
  - Optical Microscopy
  - Scanning Electron Microscopy (SEM)
  - (Scanning) Transmission Electron Microscopy ((S)TEM).
  - X-ray absorption spectroscopy (XAS): synchrotron accelerator at Canadian Light Source – Saskatoon)

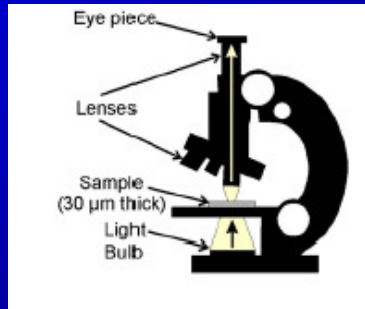
## Aqueous Phase:

- Influent and effluent chemistry



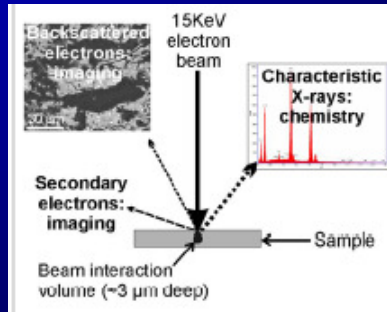
# MICROSCOPY METHODS

Optical



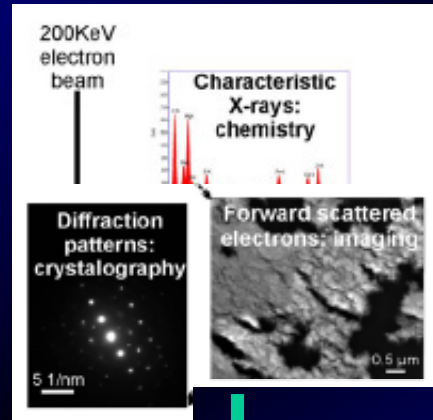
0.1 mm

SEM



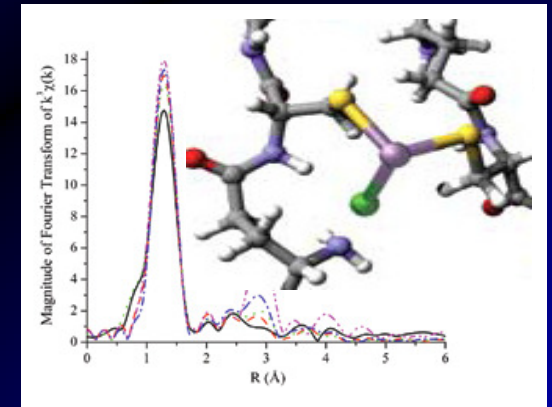
Micrometres

STEM



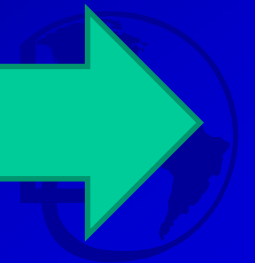
Nanometres

XAS



Atomic

Increasing Resolution



# Results: ARD Influent (FEED) Chemistry

| Parameter       | Equity | Geco  | Britannia |
|-----------------|--------|-------|-----------|
| pH              | 2.7    | 3.0   | 4.05      |
| SO <sub>4</sub> | 9,900  | 4,600 | 1320      |
| Al              | 776    | 10    | 19        |
| Fe              | 920    | 1,100 | 0.6       |
| Mg              | 919    | 170   | 64        |
| Ca              | 330    | 338   | 411       |
| As              | 6      | <0.01 | <0.0005   |
| Cd              | 1.1    | 0.04  | 0.09      |
| Zn              | 123    | 12    | 18        |
| Cu              | 60     | 0.31  | 18        |
| Mn              | 137    | 7.0   | 4.2       |
| Ni              | 8      | 0.02  | 0.04      |
| Pb              | <0.3   | 0.01  | 0.05      |

All values in mg/L

## Equity:

- High sulphate, Al, Fe, Mg
- Priority trace elements: As, Cd, Cu, Ni, Zn

## Geco:

- High sulphate, Fe
- Low Al, Mg
- Priority trace elements: Cu, Zn

## Britannia:

- Low SO<sub>4</sub>, Fe, Al, Mg
- High Cu, Zn
- Priority trace elements: Cu, Zn



# Results – Sludge Elemental Abundance

| Element   | Unit | Equity | Geco  | Britannia |
|-----------|------|--------|-------|-----------|
| Aluminum  | %    | 4.0    | 0.3   | 5.6       |
| Calcium   | %    | 14     | 10    | 13        |
| Iron      | %    | 4.3    | 26.2  | 0.7       |
| Magnesium | %    | 2.5    | 0.3   | 3.5       |
| Sulphur   | %    | 11.6   | 7.4   | 1.1       |
| Arsenic   | ppm  | 267    | <0.5  | 7         |
| Cadmium   | ppm  | 58     | 7     | 262       |
| Copper    | ppm  | 2,693  | 73    | 50,000    |
| Zinc      | ppm  | 6,317  | 2,743 | 52,000    |
| Lead      | ppm  | 0.3    | 2.6   | 150       |
| Manganese | ppm  | 6,956  | 1,880 | 12,000    |
| Nickel    | ppm  | 368    | 9.2   | 112       |



# EQUITY Mine: Microscopy

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## Mineral Assemblage:

- Dominant phase: Gypsum (only phase detected by XRD)
- Other mineral phases identified with SEM:
  - Al hydroxide,
  - Fe-oxyhydroxide,
  - Calcite
  - Apatite

Dominant trace element-bearing phase in the sludge:

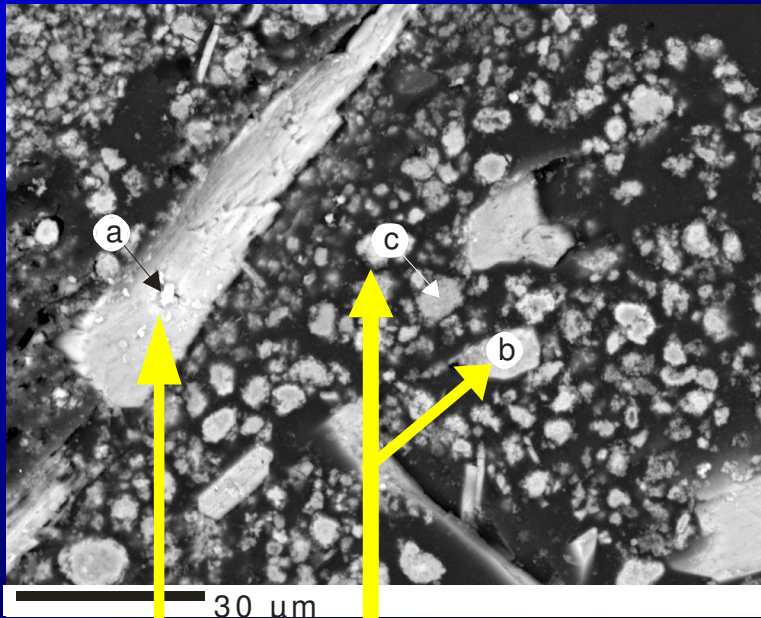


Amorphous Mg-Al-(Fe) hydroxysulphate

*Consistent with high Mg, Al and sulfate in ARD*

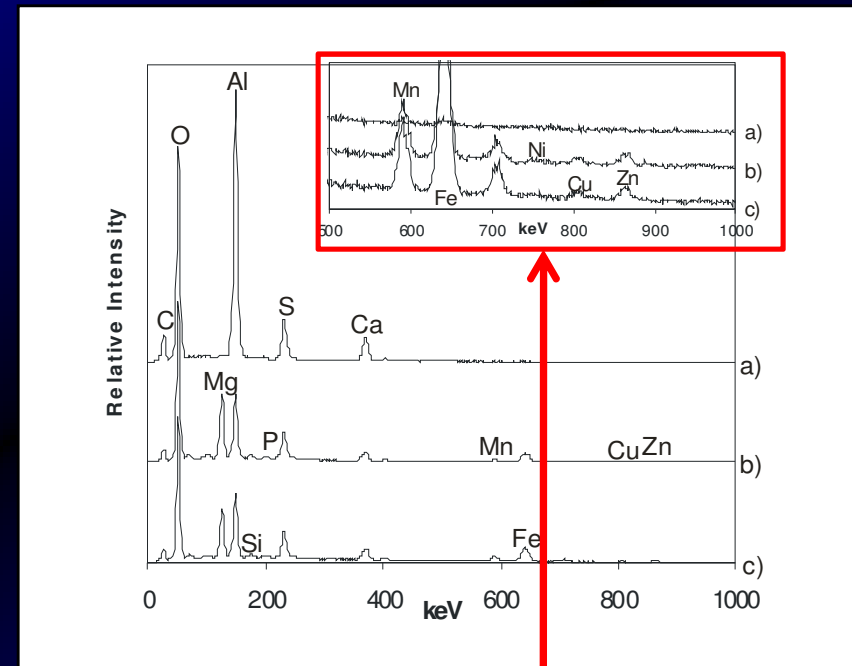


# EQUITY HDS - SEM BSE Images



a) Gypsum + Al-Fe hydroxide

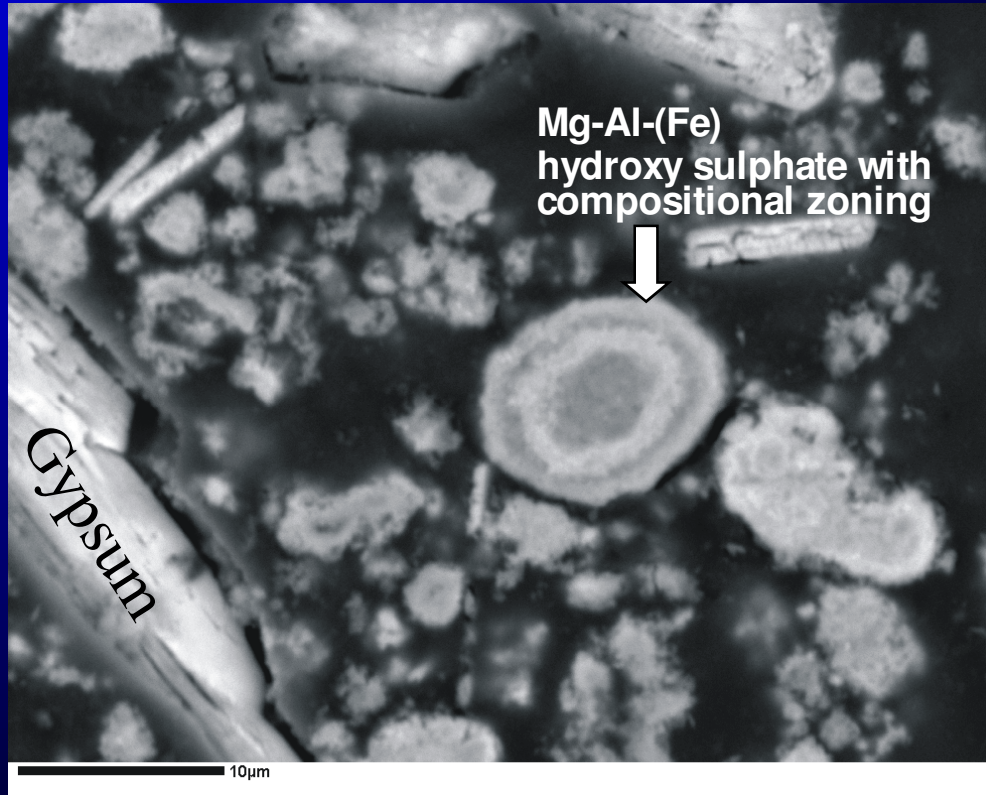
b) & c) Mg-Al-(Fe) hydroxysulphate (<1 to 15 µm)



Metals associated with Mg-Al hydroxysulfate (spectra b and c)



# EQUITY HDS: Mg-Al-(Fe) hydroxysulphate



Concentric layers  
alternating between  
Mg+Al-rich (Fe-poor)  
and Fe-rich

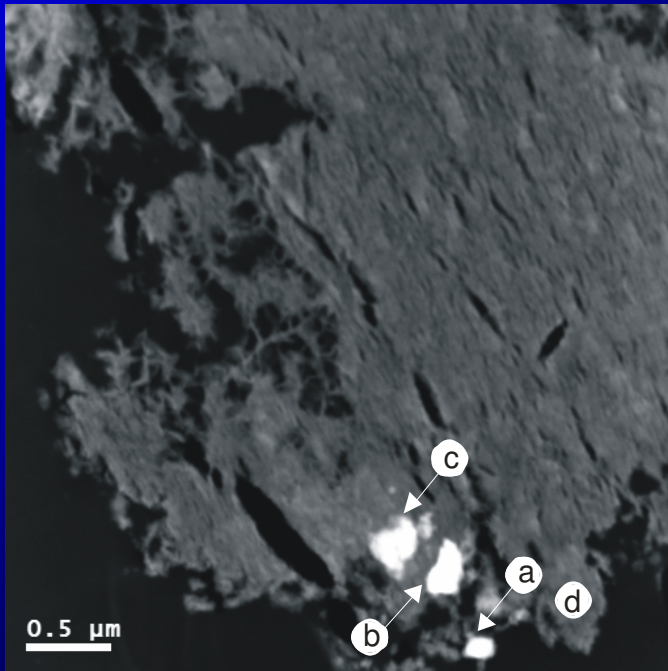
Consistent with the  
dissolution and re-  
precipitation reactions  
predicted to occur during  
sludge recycling in HDS  
process





# EQUITY HDS: (S)TEM Images and EDS spectra

## Mg-Al-(Fe) hydroxysulphate



- Zn is more strongly correlated with Mg/Al than Fe.
- Selected-area electron diffraction (SAED) did not produce any discernable diffraction patterns or rings, suggesting that this phase is amorphous.



# Geco Mine: Microscopy

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## Mineral Assemblage:

- Dominant phase: Gypsum (only phase detected by XRD)
- Other mineral phases identified by SEM:
  - Al hydroxide
  - Calcite

Dominant trace element-bearing phase in the sludge:

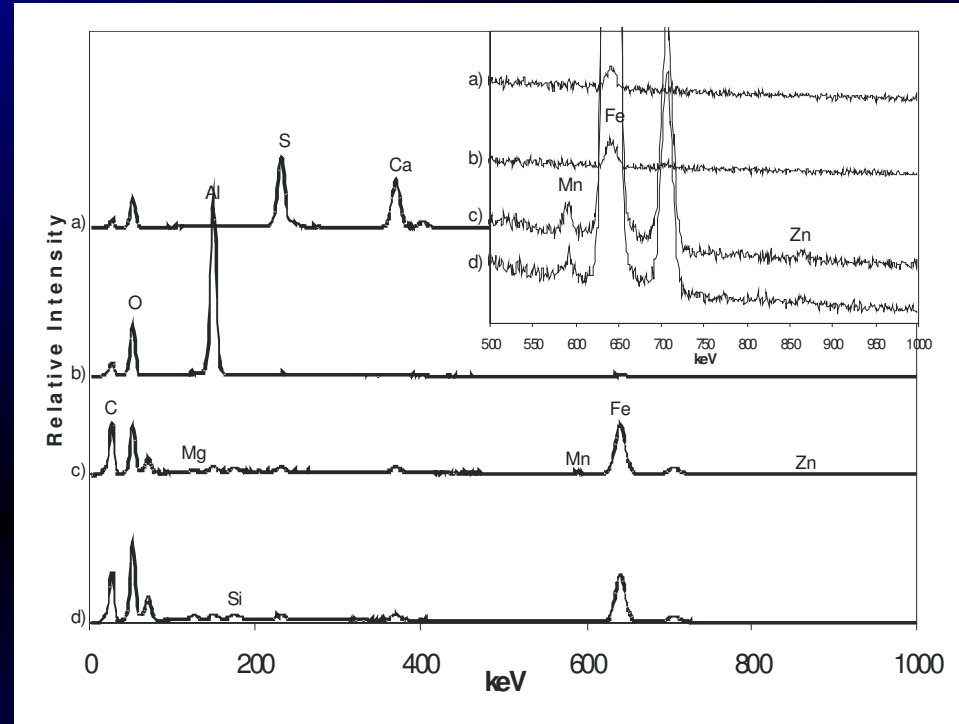
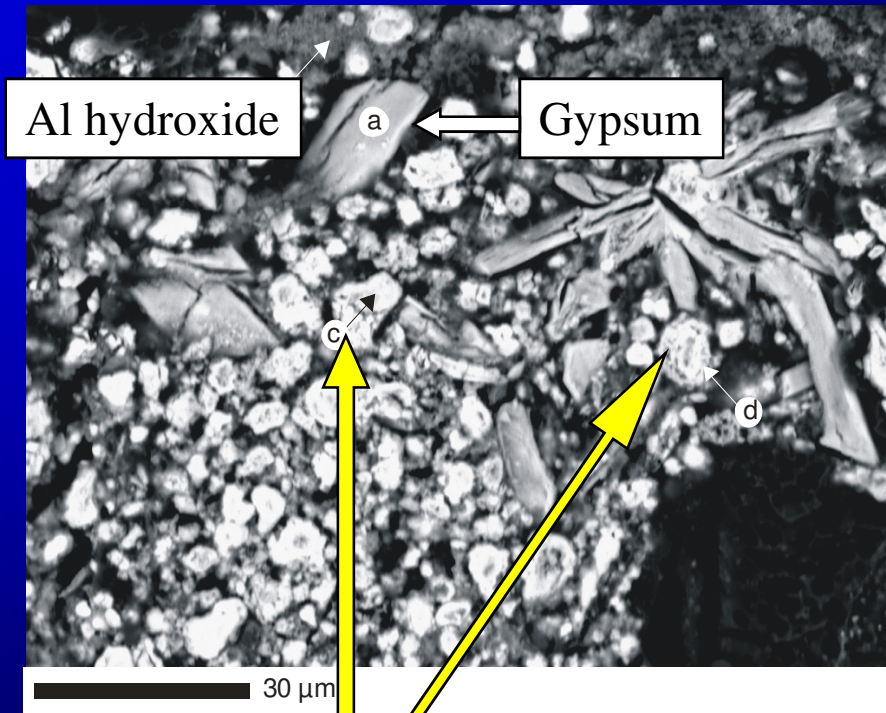


Relatively pure Fe hydroxide

*Consistent with high Fe and low Mg & Al in ARD*



# GECO - SEM BSE Images

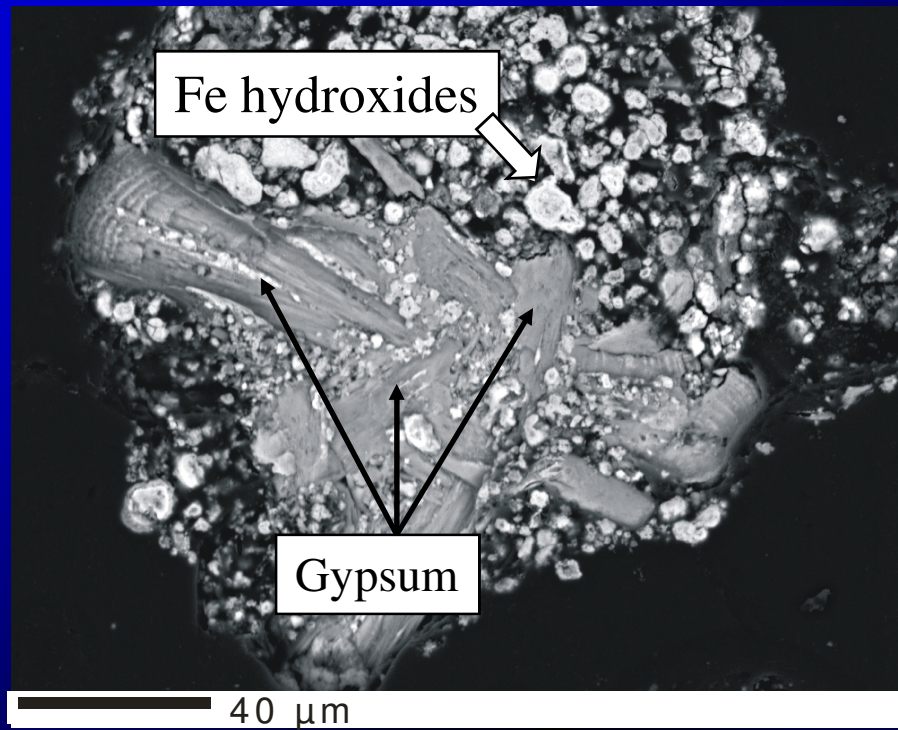


Fe hydroxides: (1-20  $\mu\text{m}$ )  
well rounded grains that  
display compositional  
zoning

Zn and Cu associated with Fe  
hydroxides (spectra c and d) and not  
gypsum (a) or Al hydroxide (b)

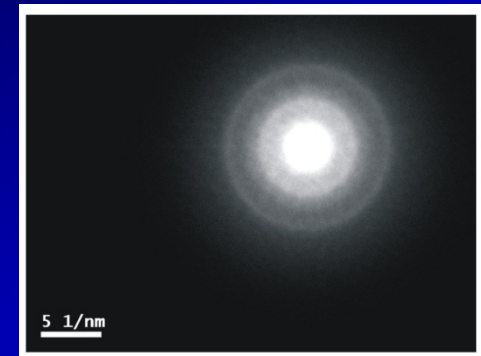


# GECO - SEM BSE Images



- Fe hydroxide (<1 to 20  $\mu\text{m}$  in size) is essentially pure with only minor amounts of other constituents (e.g. Mg, Al and S).

- Selected area electron diffraction (SAED) indicates the presence of broad diffuse rings, suggestive of poorly crystalline to amorphous Fe oxides.



# Britannia Mine: Microscopy

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## Mineral Assemblage:

- Dominant phases by XRD
  - Calcite and/or Mg-calcite
- Other mineral phases identified by SEM:
  - Quartz
  - Fe oxyhydroxide
  - No gypsum

Dominant trace element-bearing phase in the sludge:



Amorphous Cu-Zn hydroxide

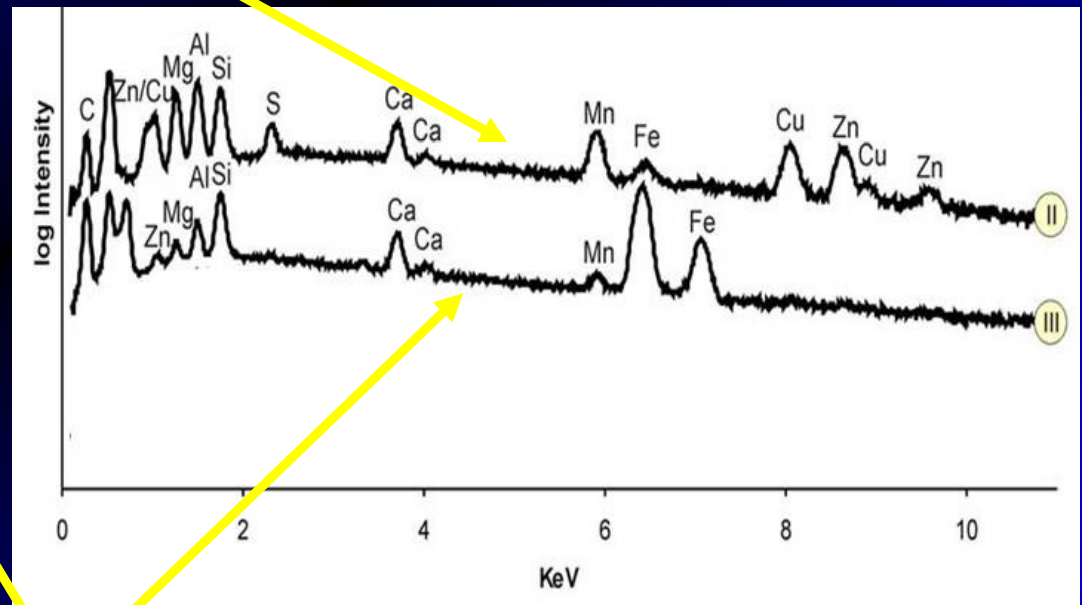
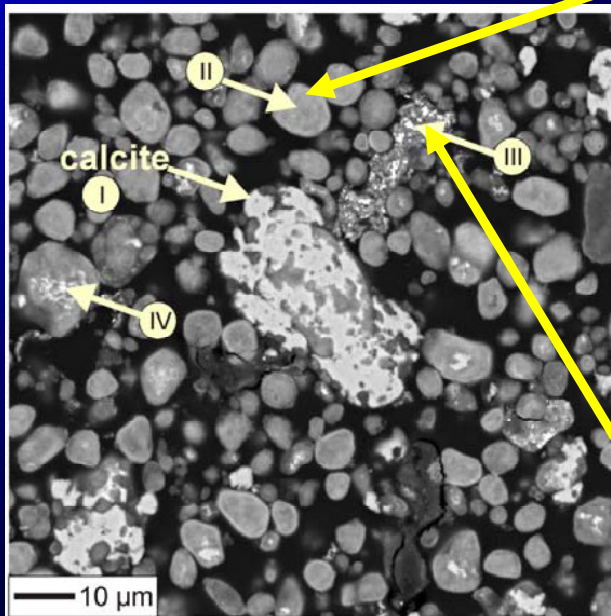
*Consistent with low Fe/Al and high Cu and Zn in ARD*



# Britannia - SEM BSE Images

- Metal bearing phase : fine-grained aggregates of Cu hydroxide
- Contains variable Mg, Al and Si
- Fibrous grains ranging in size from 5 to 100  $\mu\text{m}$ .

Cu-Zn (Mg, Al, Si) hydroxide



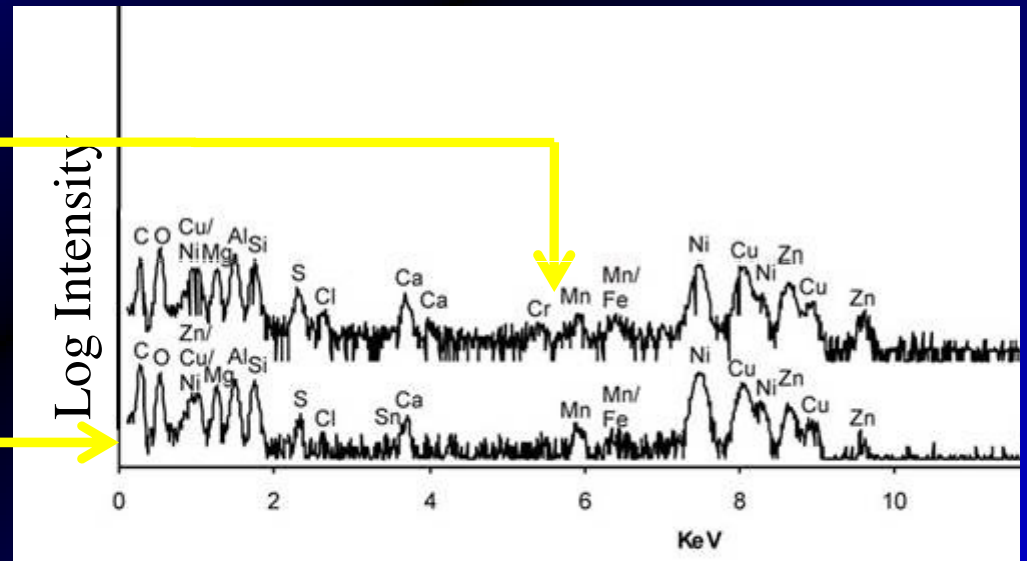
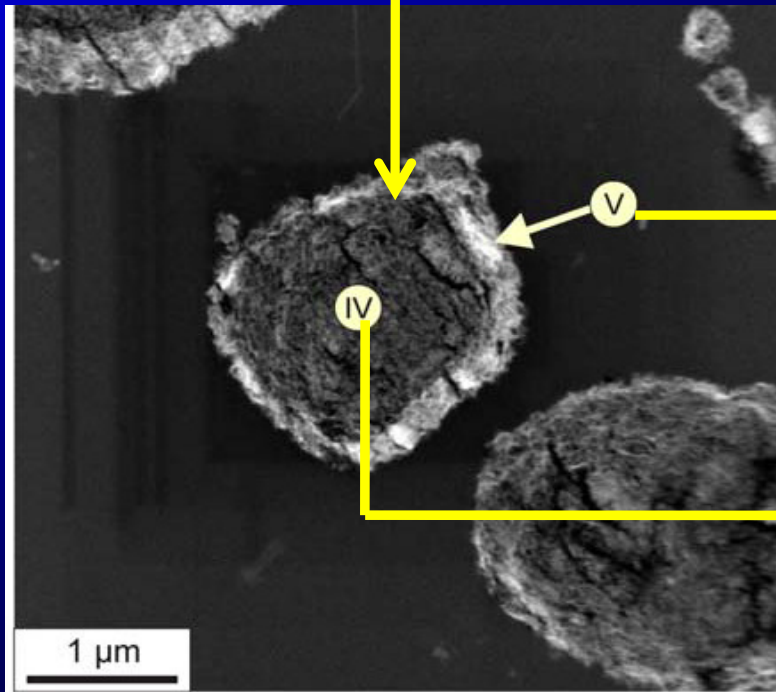
Fe oxyhydroxide



# Britannia – STEM Images

Cu-Zn hydroxide

Fibrous grains display  
compositional zoning



Selected area electron diffraction patterns consistent  
with amorphous or nanocrystalline material



# SUMMARY

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- In all sludges, metal-bearing phases were amorphous (or poorly crystalline) and variable in composition (relatively pure to heterogeneous).
- Metal-bearing phases largely governed by composition of ARD. Proportions of Fe, Mg, Al and S important. Textures and concentric zoning reflective of HDS process.
- Varied spectrum in metal-bearing phases observed, including: 1) heterogeneous Mg-Al-(Fe) hydroxysulphate; 2) relatively-pure Fe hydroxide; and 3) heterogeneous Cu-Zn hydroxide.
- Different metal-hosting phases can be expected to show contrasting chemical stabilities with respect to pH and Eh conditions. Results therefore highlight need to understand controls governing long-term chemical stability of various sludge phases.





# Predictive Tool

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ARD Chemistry



“Sludge type”  
(metal-bearing phases)



In situ chemical stability,  
depending on depositional  
environment

Further investigation  
required



# Thank you

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## Questions?

