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# Development of passive bioneutralization technology





## What is passive bionutralization

- Technology must meet definition of “passive”
- Must be capable of treating acidic mine water with pH lower than 4.5, i.e. extend the range of passive sulphate reduction technology downwards
- Suitable as pretreatment ahead of sulphate reduction or as stand-alone technology for acidity and metal removal
- Strictly biological process – no use of neutralizing chemicals



# Background to bioneutralization

- ❖ Standard sulphate reducing systems require  $\text{pH} > 4.5$
- ❖ For mine effluents with  $\text{pH} < 4.5$ , some form of pretreatment is required
- ❖ PHD commenced research in 2003 to develop biological system as redox-reducing reactor upstream of ALD
- ❖ Discovered that redox reduction and bioneutralization could be effected in single stage DPBR with right biota
- ❖ Technology has been developed:
  - o Sustainably treat water with  $\text{pH} < 3$
  - o Also treat water with metals  $> 1000\text{mg/l}$



# PHD Column Laboratory



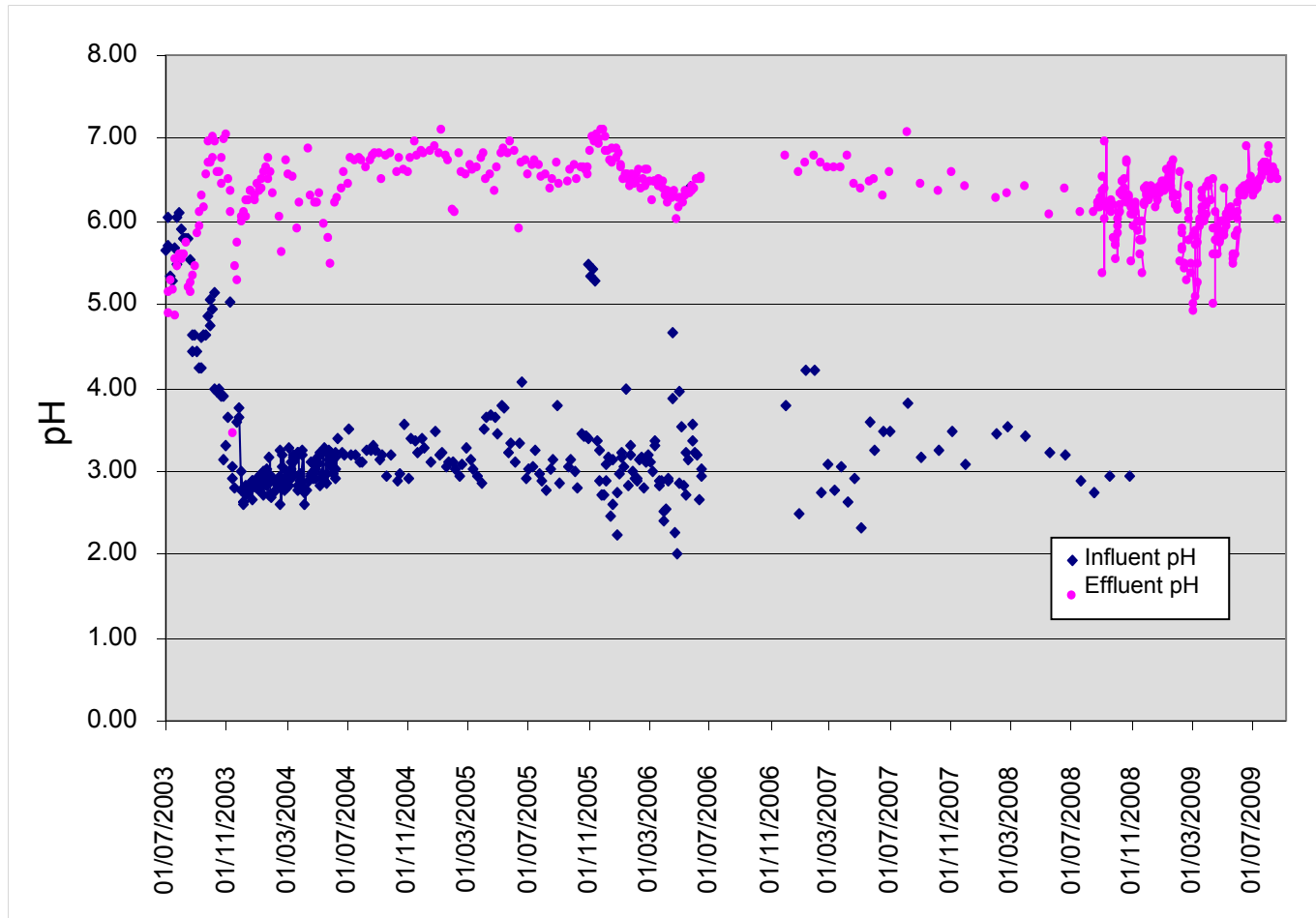


# Key aspects of early research

- Tried numerous approaches before isolating bacterial cultures capable of bioneutralization
- Undertook long-term studies coupled with depth profile studies & microbial ecology
- Developed basic descriptive model
- Believe that there is a complex consortium of bacteria involved in bioneutralization:
  - Population 1 – remove oxygen and reduce redox to -250 to -350 mV with high H<sup>+</sup>
  - Population 2 – anaerobic degradation of LC to simple charge-neutral carbon compounds
  - Population 3 – sulphate reduction
  - All 3 groups in closely-linked feedback
- Bioneutralization happening in first 300 mm of reactor
- Long-term performance validated – viable technology for simple acidic waters



# Bio-neutralization studies - results



- ❖ Have developed and operated a technology that can achieve the following:
  - Operate as a shallow DBPR with 2-3 day retention
  - Increase pH from  $<3$  to  $>6.5$  & produce water with net alkalinity of 400 - 500 mg/l
  - Reduce redox potential to  $< -300\text{mV}$
  - Pre-treat water for use in DPBR
  - Metal accumulation  $>1\text{kg}/\text{m}^3$
  - Sulphate removal up to 1000 mg/l
  - Bacterial cultures can be successfully seeded into new reactors



## Treating highly acidic waters

- ❖ Whereas a standard sulphate reducing system will produce 1.05 mol alkalinity per mol sulphate reduced, the DPBR produces 1.5 mol.
- ❖ DPBR technology can consistently produce 3500 mMol/m<sup>3</sup>/day alkalinity (as CaCO<sub>3</sub>)
- ❖ Technology packages being developed to use DPBR as high rate alkalinity producing system for neutralization and metal removal





**NEW CHALLENGE IN EARLY 2006**

**TREAT HIGHLY ACIDIC WATER WITH  
VERY HIGH METAL ACIDITY -  
LANDAU**



## Treating high acid & metal waters

- ❖ High metals present as high metal acidity in addition to proton acidity (pH<3; Fe 800-1200 mg/l; Al 100-150 mg/l; Mn 80-120 mg/l)
- ❖ Formation of metal hydroxides and sulphides consumes alkalinity
- ❖ Metal precipitates have hydraulic and thermodynamic/physical effects on bacterial consortia
- ❖ Requires specific design approaches to deal with metal acidity
- ❖ May require acceptance of lower efficiency as pretreatment to DPBR

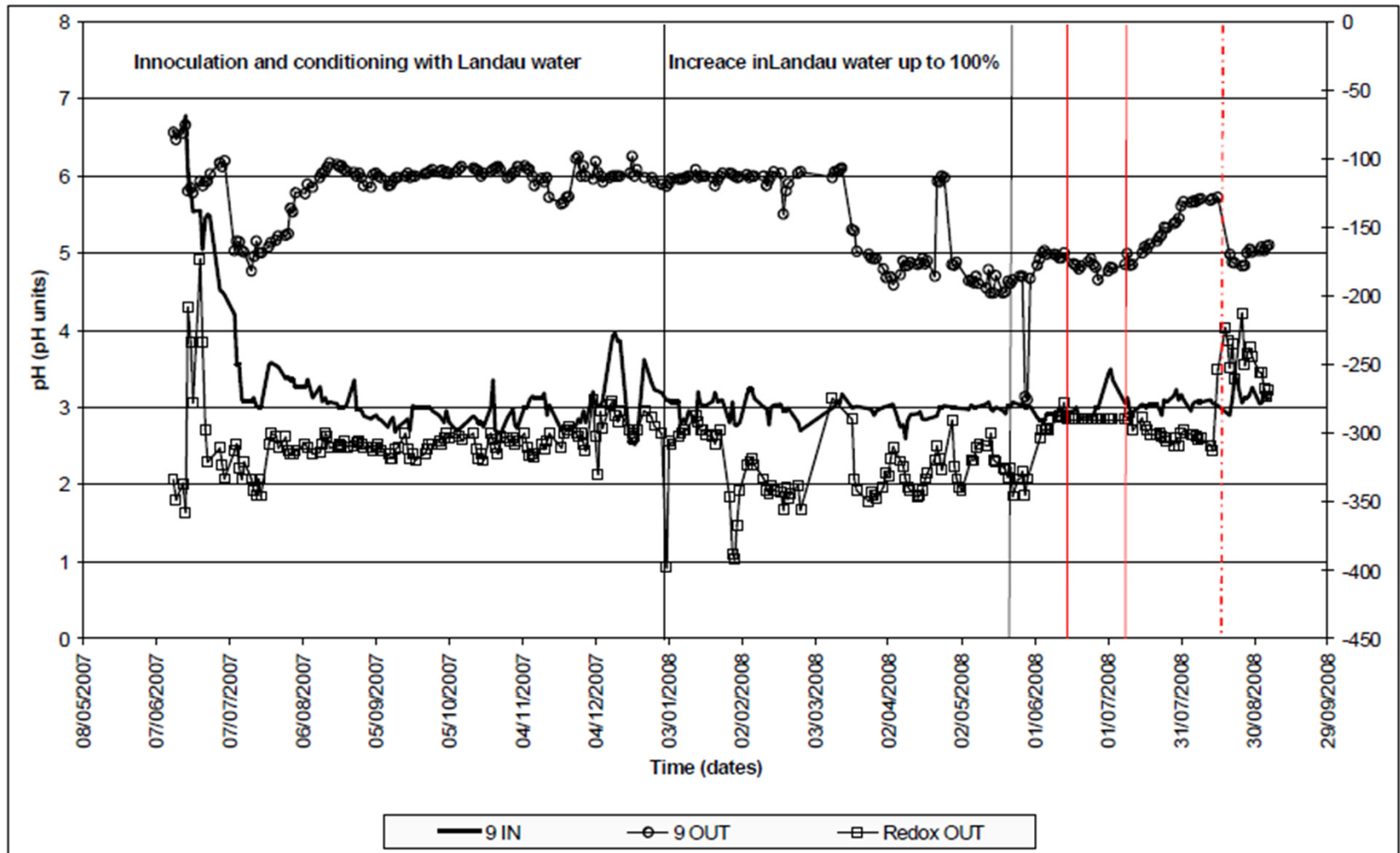


# History of Landau bioneutralization

- **Phase 1:** Upflow DPBRs achieved 30% blend of Landau water then failed – post mortem showed metal fouling
- **Phase 2:** Various components:
  - Bench scale studies to remove maximum metals before bioneutralisation with alkali addition (result = oxidation + settling)
  - Sacrificial downflow metal removal columns – slowly acclimatised to Landau
  - In-situ metal backflushing (result = 89% Fe recovery & 21% Al recovery)
- Had developed good understanding of these reactors by mid-2008 and response to various upsets such as draining of reactor and low temperatures and response to various remediation strategies (retention time, pH, blending, etc.)

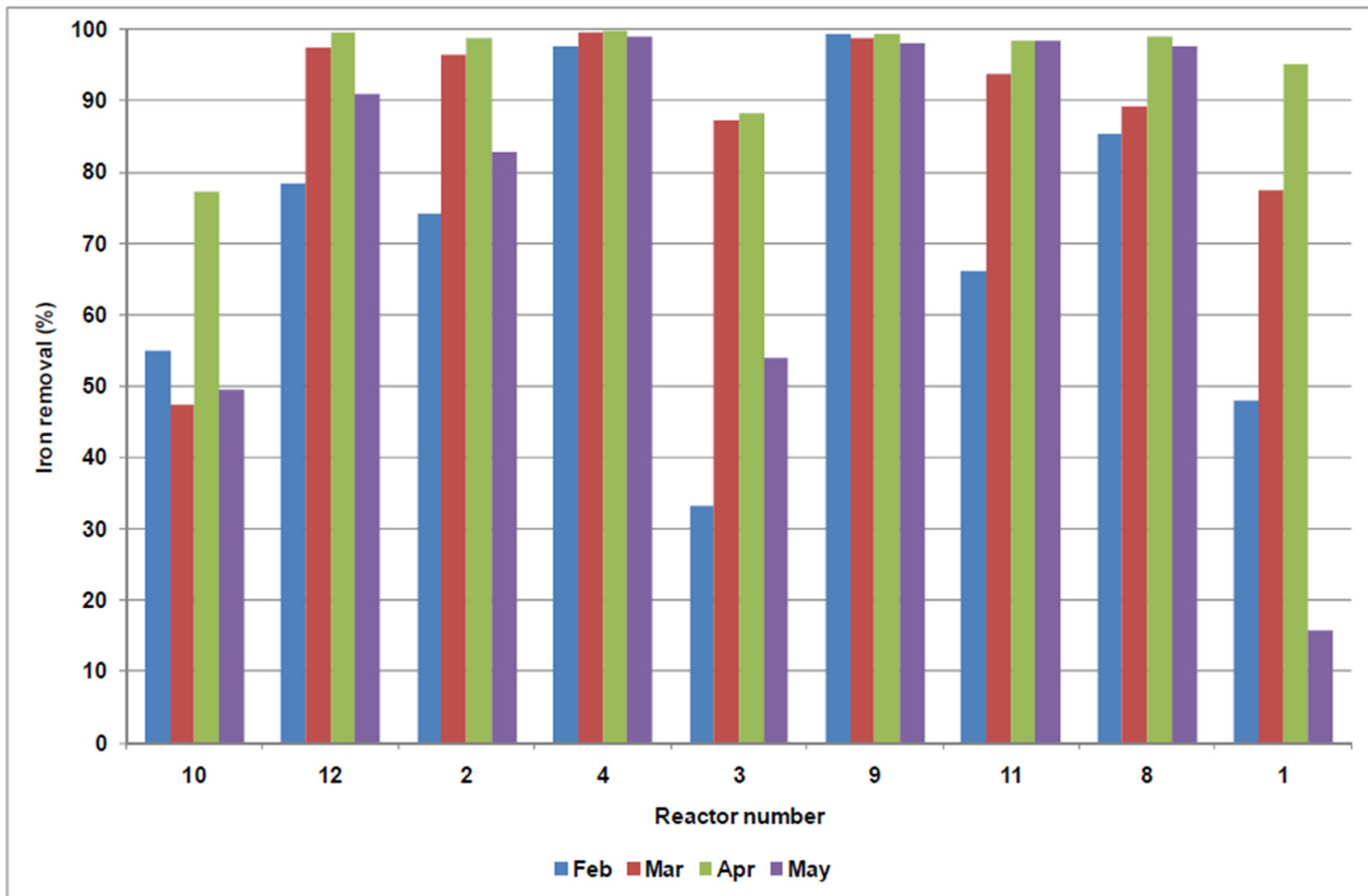


# Typical bioneutralization column results



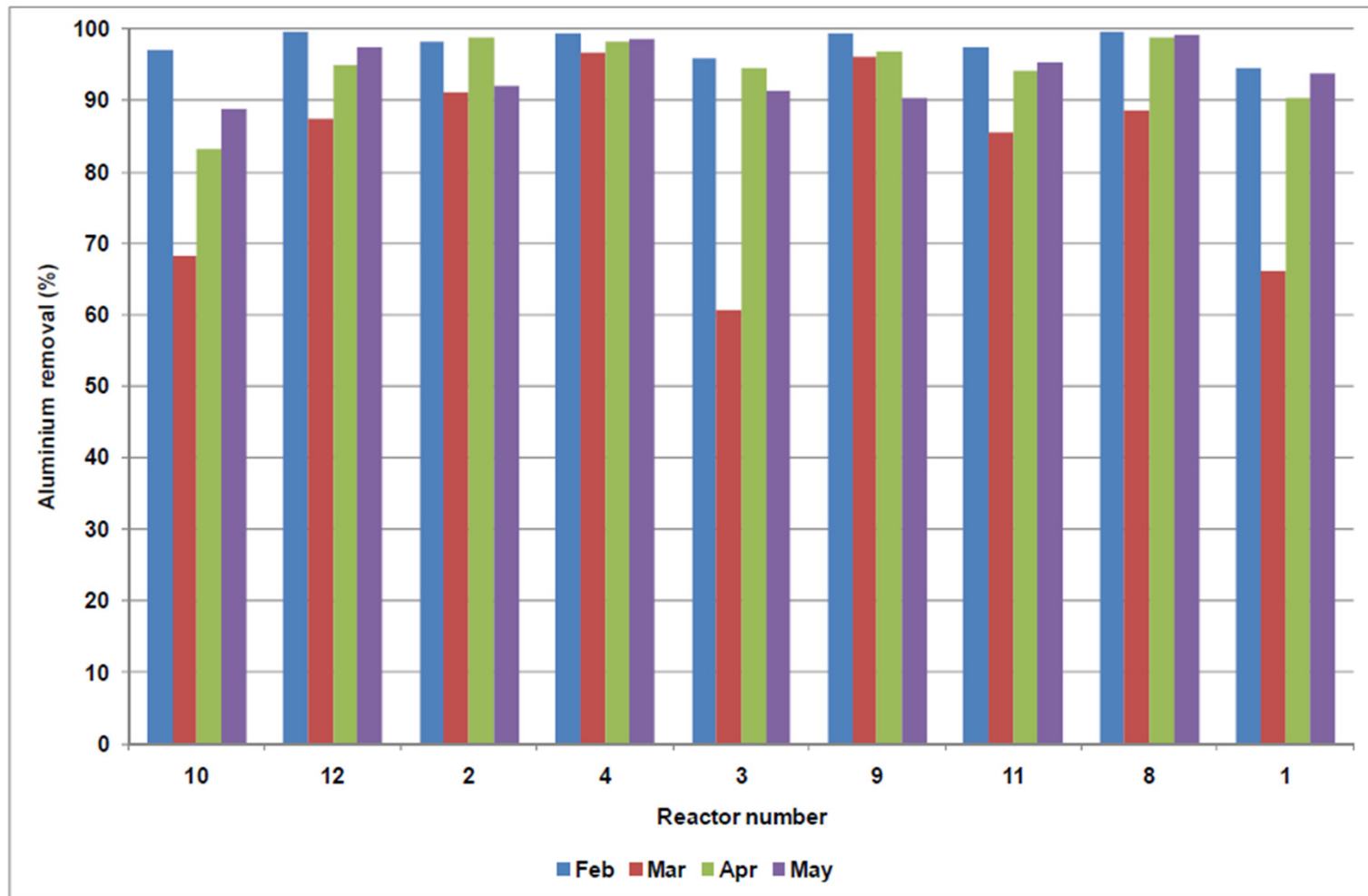


# Iron removal efficiency





# Aluminium removal efficiency





## History of Landau bionutralization - 2

- **Phase 3:** Prepared detailed proposal in October 2008 to take research to next logical step:
  - combine pre-treatment metal removal reactor with bionutralisation DPBR
  - integrate metal removal, SOBR & DPBR
  - further refine and optimise metal removal strategies
  - post-mortem studies (microbial & mineralogical) on metal removal reactor
- **DECISION:** revert to care & maintenance programme with no active research component – this is still the current status.



## Current Status

- Bioneutralisation reactors have been operating around 30 months on 100% Landau water producing pH>5 and around 90% metals removal
- Have set up and evaluated integrated bioneutralization with DPBR since Jan 2010
- Need to address issues relating to removal of metals from reactor and optimisation of process operating conditions
- Research programme in dead end – continued care and maintenance programme will not advance the technology





## Conclusions

- ❖ Robust bioneutralization technology has been developed for treating mine water with pH down to 2.5 and total metals in range of <200 mg/l.
- ❖ Such reactors have been operating continuously in lab-scale for 8 years and incorporate key elements of DPBR technology
- ❖ Treatment of acidic water with very high levels of metals is more problematic but has been shown to be capable of raising pH above 4.5 required for standard DPBR
- ❖ Research programme on acidic high-metal water is currently stalled due to lack of funding and reactors are in care & maintenance mode