

Performance of the Pit Lake at Ridgeway Gold Mine, South Carolina, USA

Oscar Flite and Roy Duckett



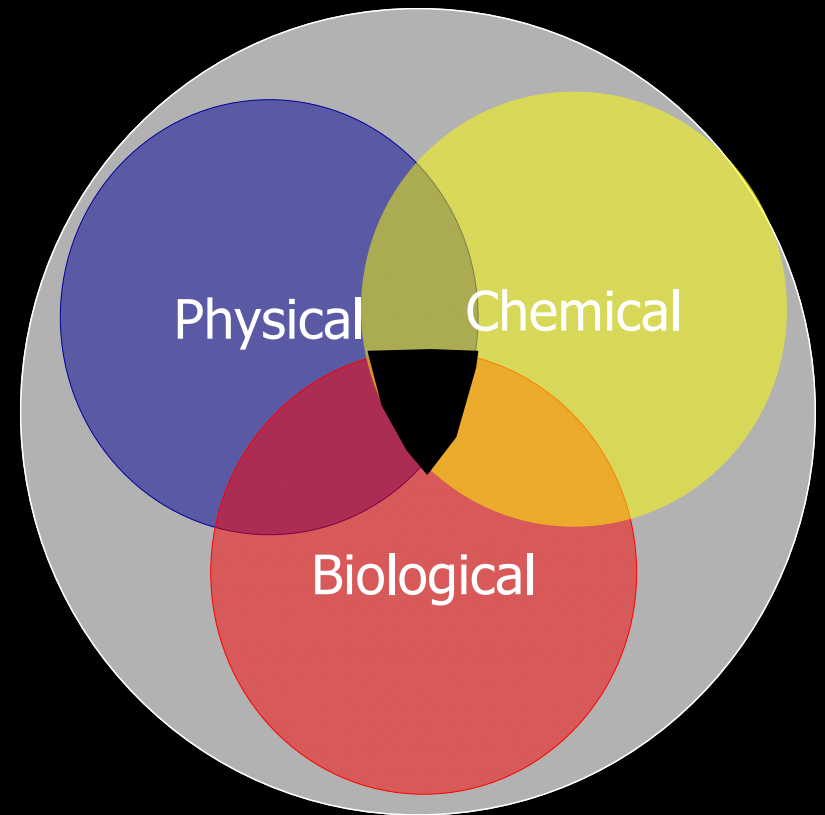
RioTinto

Reframing the big picture...

- Short-term goals
 - Increasing lake pH
 - Waste rock disposal
 - Removing metals from pit lake
 - Resource extraction
- Long-term goals
 - Titrating lake into carbonate buffer system
 - Strategic placement to optimize lake sustainability
 - Developing a sustainable ecosystem
 - Sustainable mining

What is the big picture for pit lakes? What should they look like?

- Many approaches to pit lake remediation
- Not all take a holistic approach
- Disregarding one or more domains often ends short of intended goal



Rio Tinto-Ridgeway Mining Company

Ridgeway, South Carolina

- Mining from 1988-1999
 - Closure from 1996-2001 (included liming)
 - Clay cover on tailings facility
 - Site vegetation
 - Wetlands construction
 - Development of two fresh water lakes
 - Extensive Monitoring Program
 - Physical
 - Chemical
 - Biological
- 3.5 years of weekly data
3 years of monthly data
4 years of biannual-annual data



Pit lake monitoring program

Continuous: thermistors
weather stations

Weekly: Hydrolab profile
Licor profile
phytoplankton
zooplankton
chlorophyll a

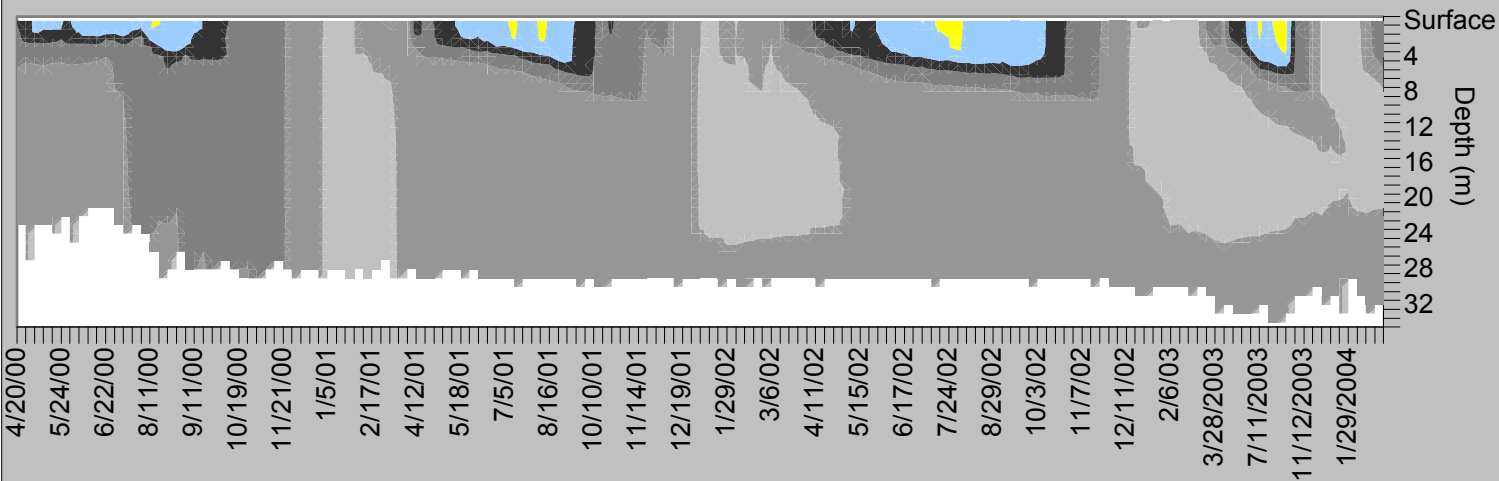
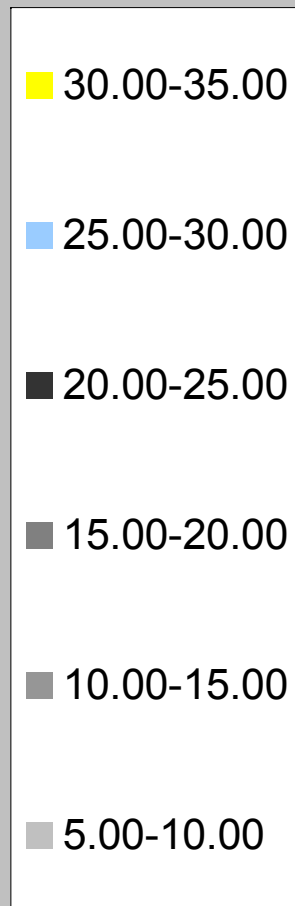
Monthly: primary productivity
macroinvertebrates

Quarterly: chemistry profiles



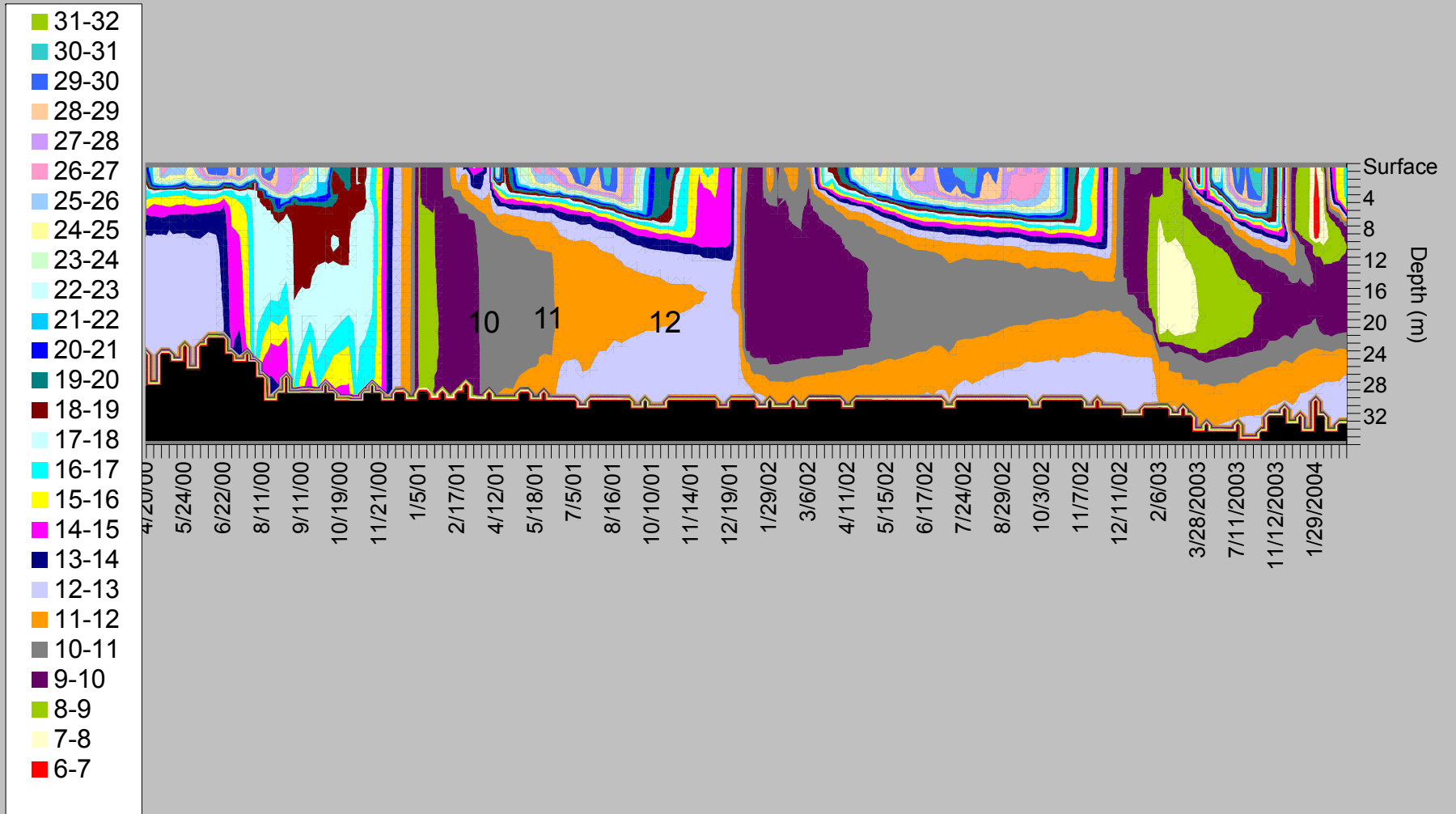
Temperature (5° C resolution)

Temperature (°C)



Temperature (1° C resolution)

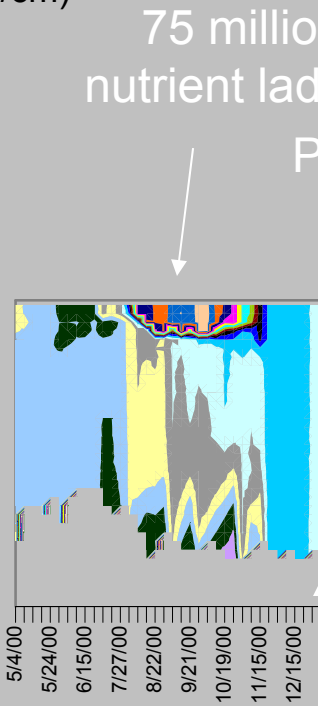
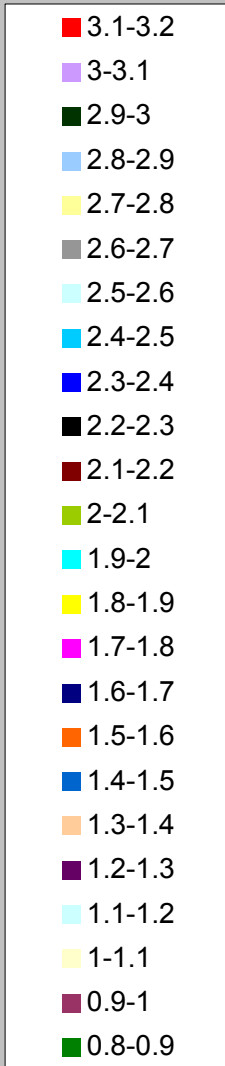
Temperature (°C)



Specific cond

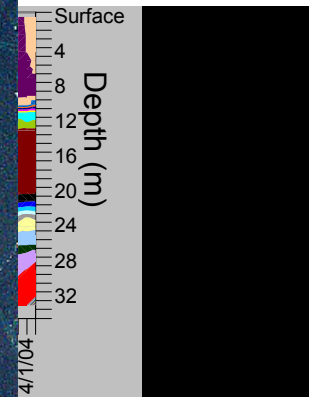
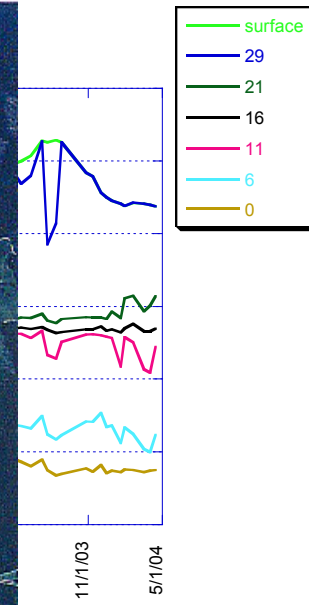
Stochastic events (perturbations)

Conductivity (mS/cm)



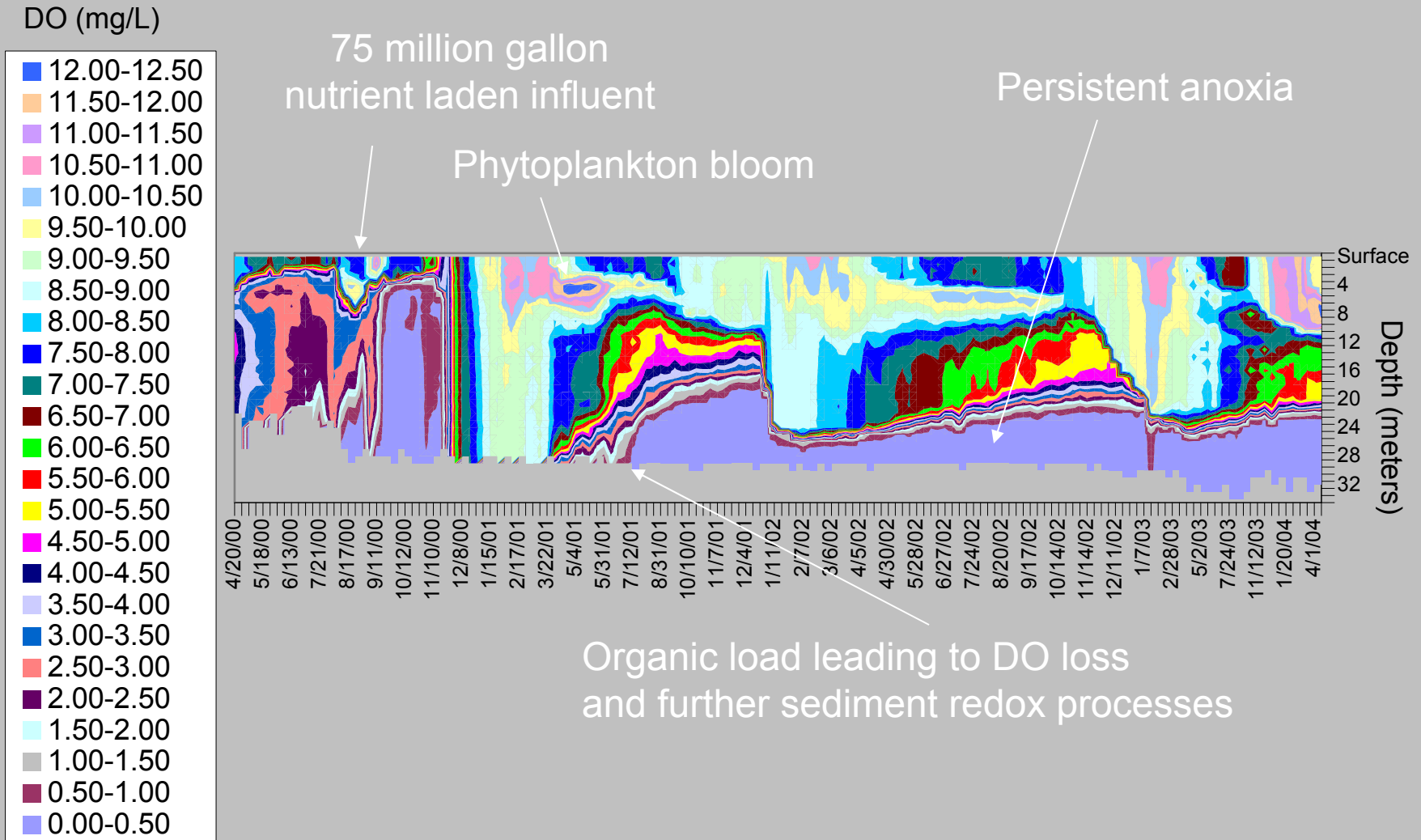
Winte

Fa
up

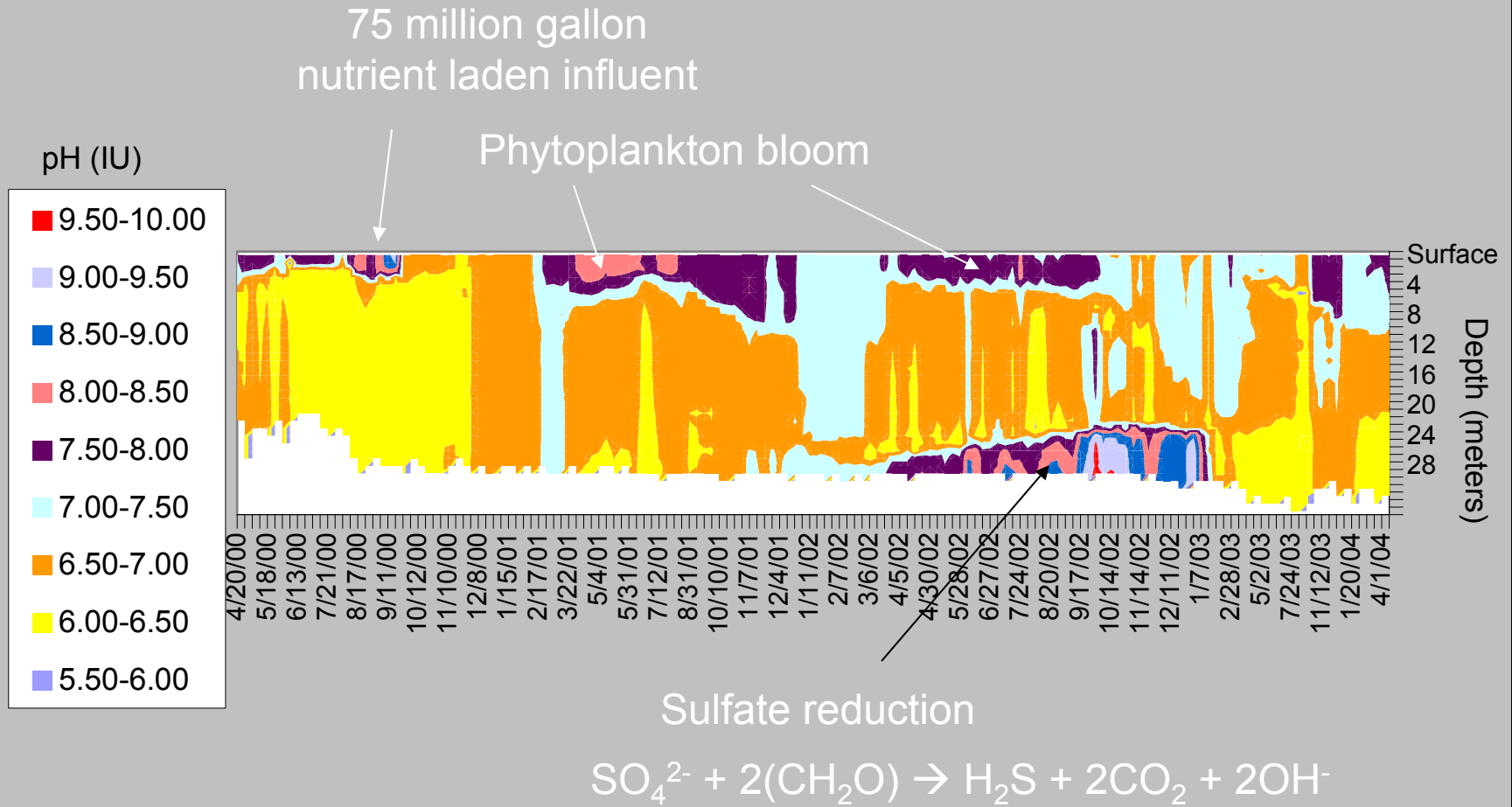


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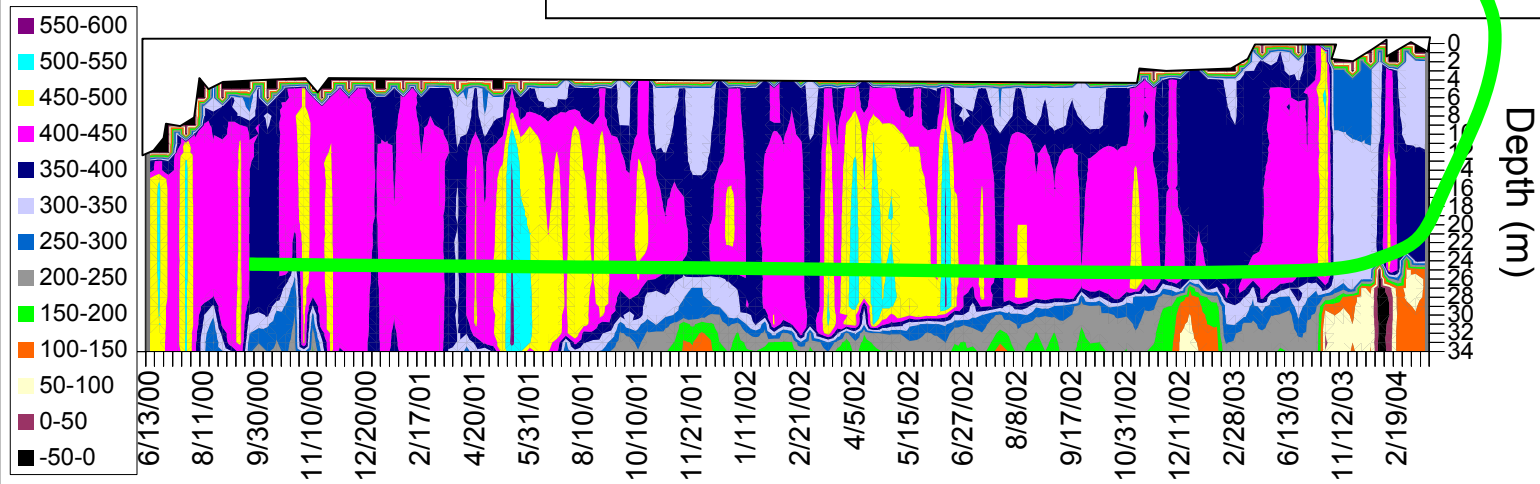
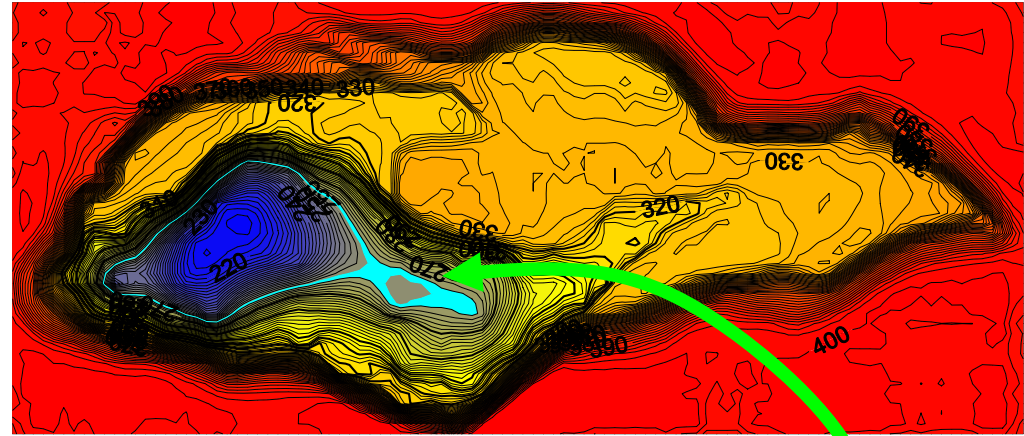
Dissolved oxygen



pH trends



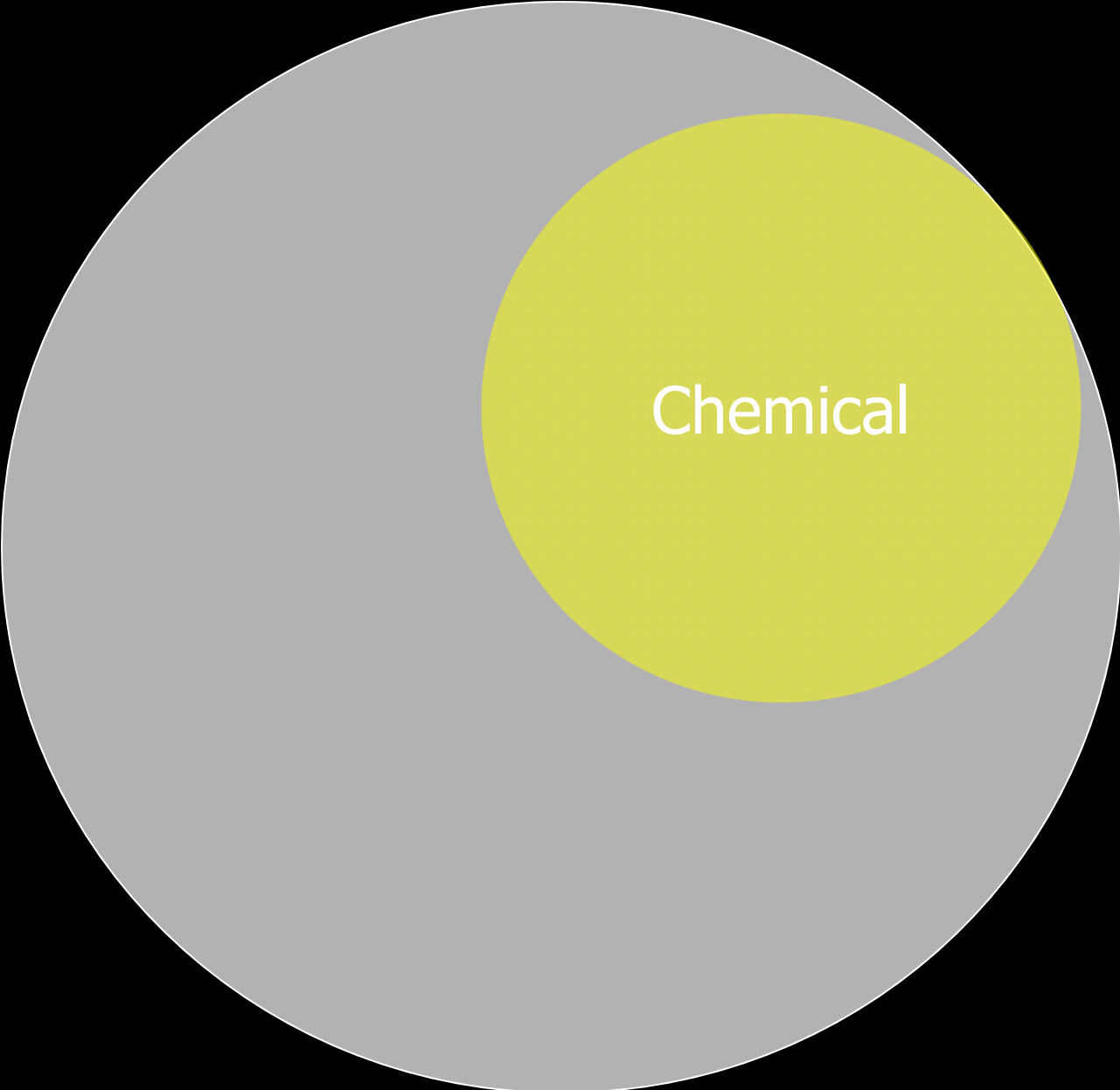
Redox trends



How important is the lake bathymetry for producing and/or
MAINTAINING meromixis? These systems are manmade!

Development of meromixis in South Pit Lake

- 1) Biogeochemical
- 2) Dilution



Chemical

More than just dilution

2009 data

depth	Na
0	39
15	38
30	64
1.641026	

Chemistry profile-bottom water density dynamics

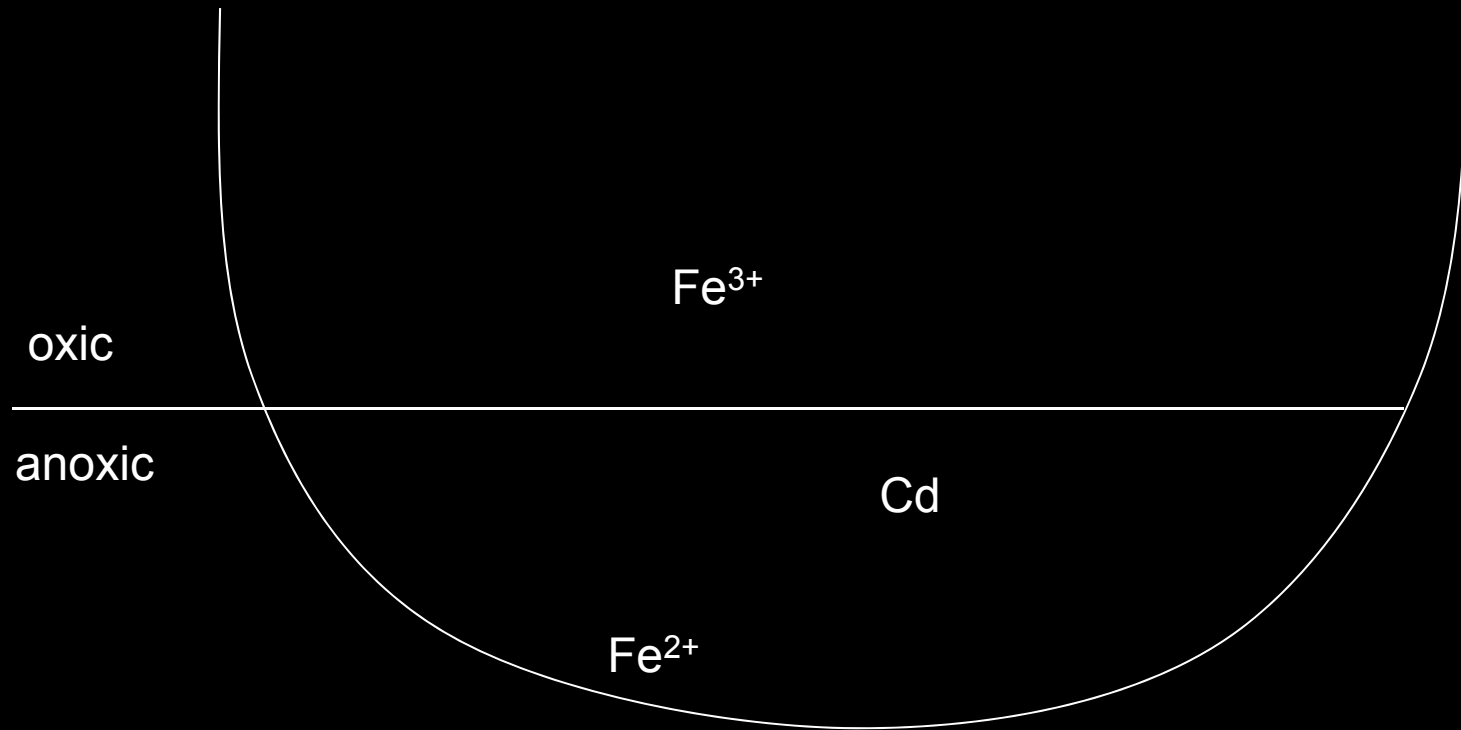
Constituent (mg/L)	South pit lake depth (m) - 2005							
	0	5	10	15	20	25	30	34
Aluminum	0	0	0	0	0	0	0	0
Ammonia - N (phenate)	0.11	0	0	0.12	0.14	1	1.6	1.5
Antimony	0	0	0	0	0	0	0	0
Arsenic	0	0	0	0	0	0	0	0
Barium	0	0	0	0	0	0	0	0
Beryllium	0	0	0	0	0	0	0	0
Boron	0	0	0	0	0	0	0	0
Cadmium	0	0	0	0	0	0	0	0
Calcium	180	200	220	330	310	500	570	620
Chloride	5	5.4	5.6	6.2	6.7	7	7.5	7.5
Chromium	0	0	0	0	0	0	0	0
Cobalt	0	0	0	0	0	0	0.32	0.35
Copper	0	0	0	0	0	0	0	0
DOC	3.2	4	3.3	2.3	2.2	2.1	1.9	2.1
Iron	0	0	0	0	0	14	68	75
Lead	0	0	0	0	0	0	0	0
Magnesium	39	42	46	60	53	89	140	150
Manganese	0	0	0	1.3	1.7	24	24	26
Molybdenum	0	0	0	0	0	0	0	0
Nickel	0	0	0	0	0	0	0	0.2
Nitrate - N	0	0	0	0.081	0.12	0.058	0	0
Ortho-phosphorus	0	0	0	0	0	0.054	0.091	0.049
Phosphorus	0.014	0	0	0	0	0.012	0.032	0.041
Potassium	0	0	0	0	0	0	29	32
Selenium	0	0	0	0	0	0	0	0
Silica (calculation)	0	0	0	0	0	0	12	15
Silicon	0	0	0	0	0	0	5.9	6.9
Silver	0	0	0	0	0	0	0	0
Sodium	26	30	32	46	44	51	66	73
Sulfate	730	640	740	1100	1300	1800	2000	2000
Thallium	0.091	0.052	0	0.056	0	0.1	0.064	0.071
Tin	0	0	0	0	0	0	0	0
Titanium	0	0	0	0	0	0	0	0
TKN	0	0	0.54	0.82	0	2.7	2.8	2
Vanadium	0	0	0	0	0	0	0	0
Zinc	0	0	0	0	0	0	0	0

Dilution factors

	5m	30m	30m/5m
Cl	5.4	7.5	1.39
Ca	200	570	2.85
Mg	42	140	3.33
SO4	640	2000	3.13
Na	30	66	2.20
Fe	0	68	
Mn	0	24	
Alk(2003)	30	239	7.97

Analysis does not answer how the system is stabilized!

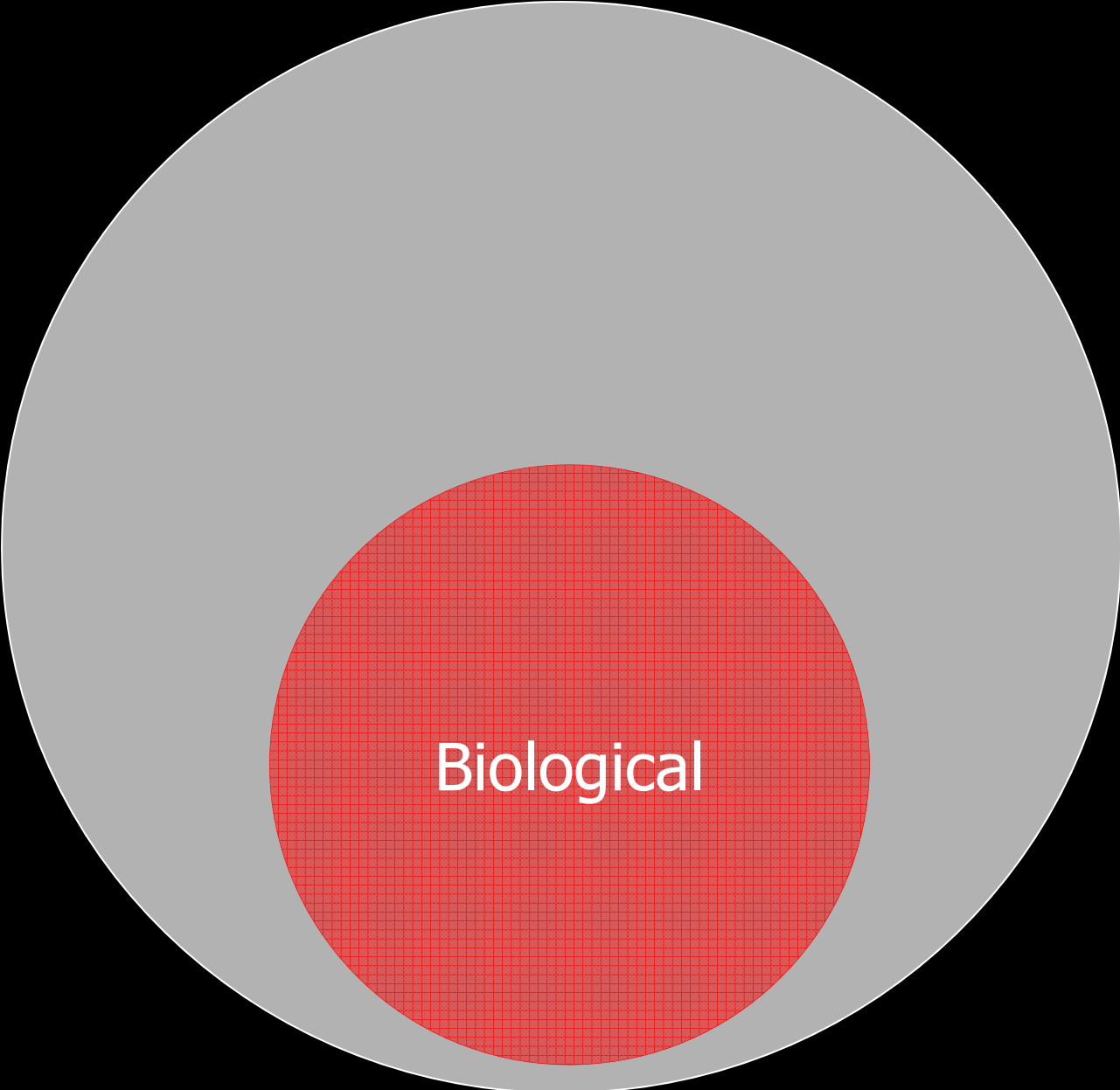
Controlling metals through the 2-layered system



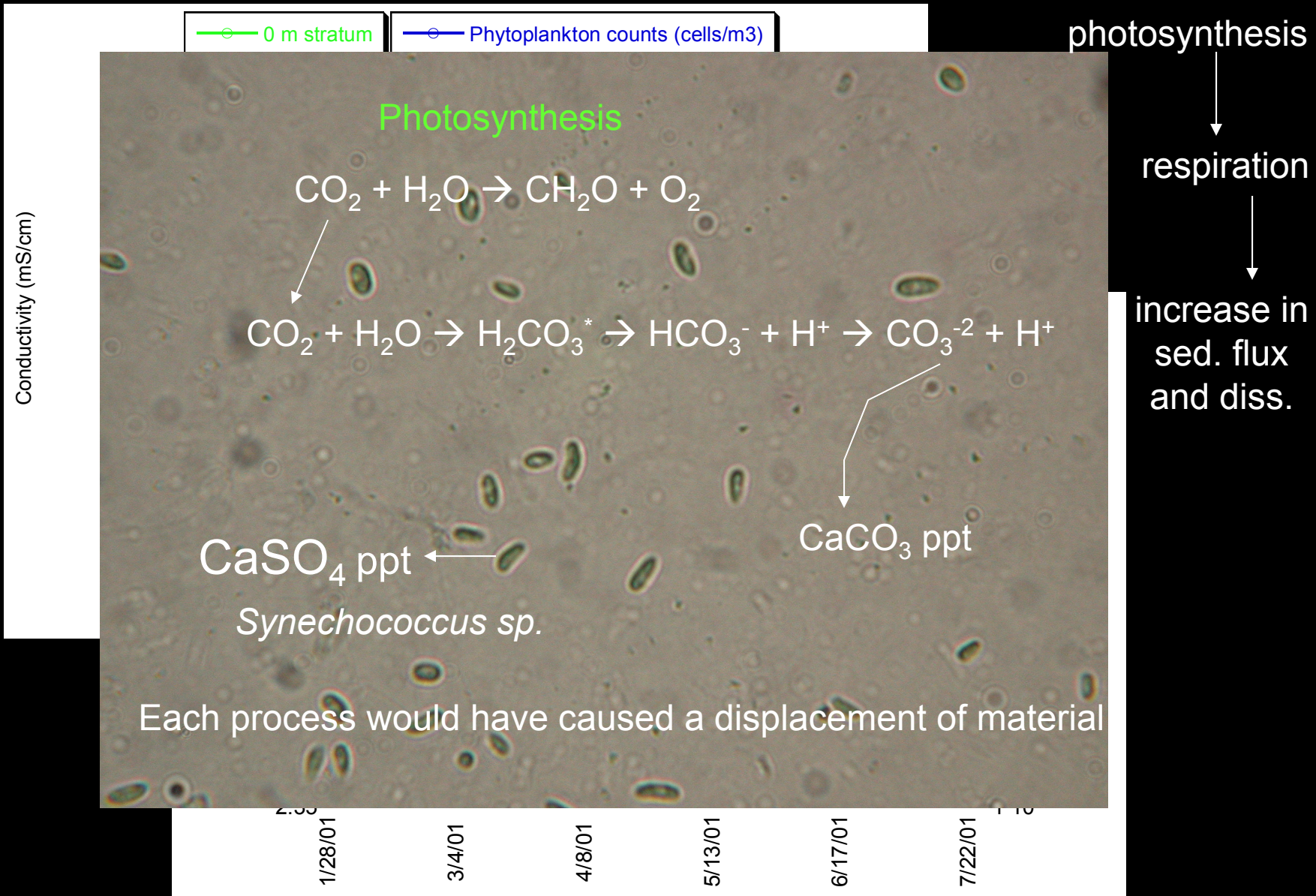
In-lake reaction chamber

Development of meromixis in South Pit Lake

- 1) Photosynthesis
- 2) Respiration

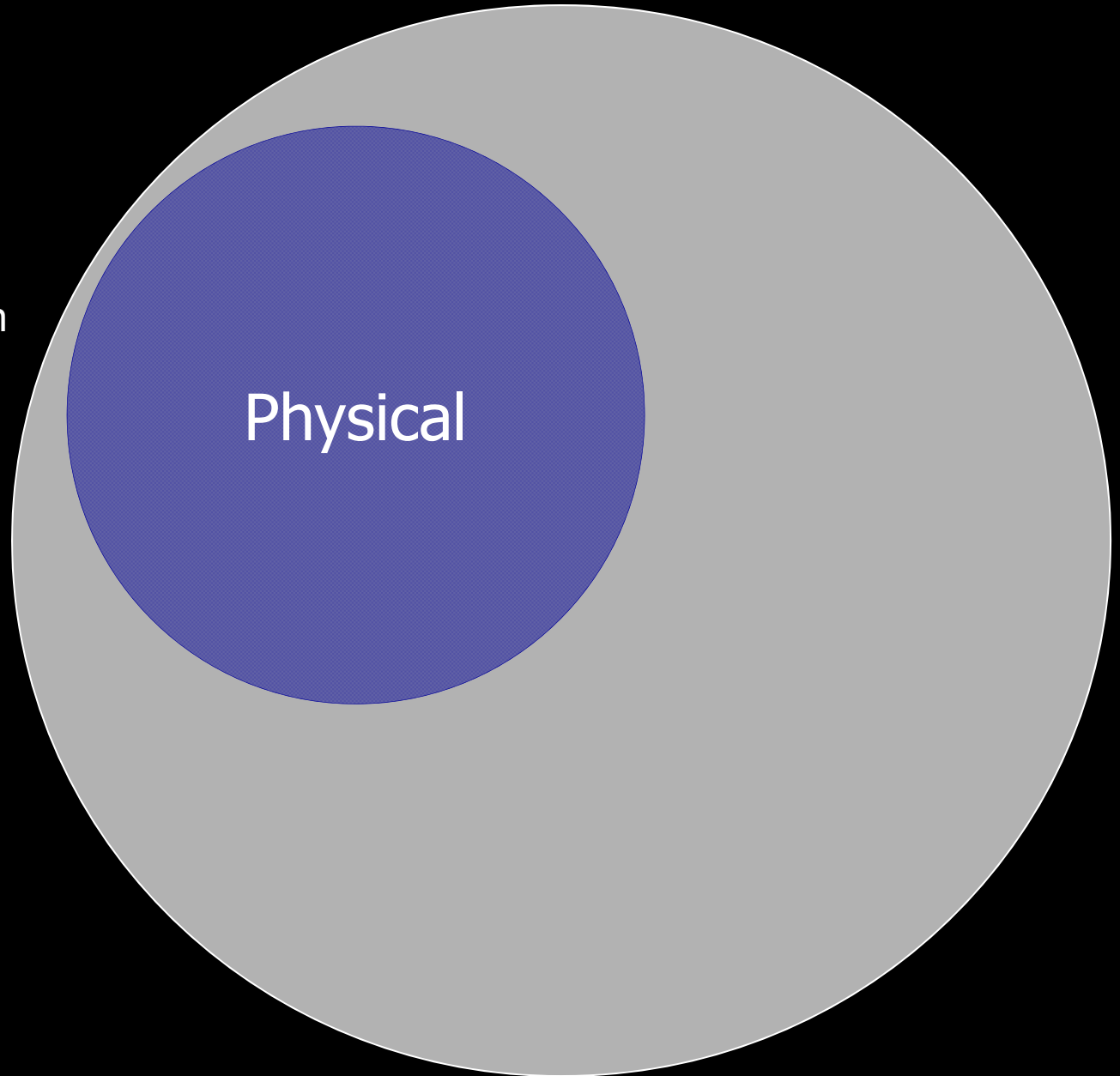


Evidence of phytoplankton facilitated meromixis



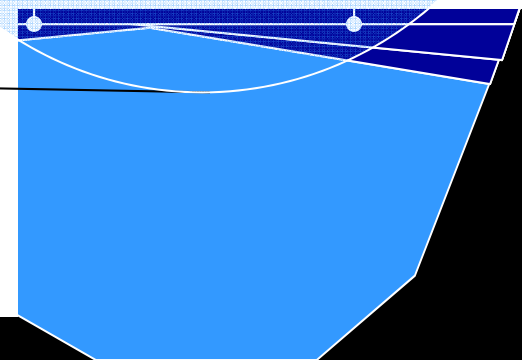
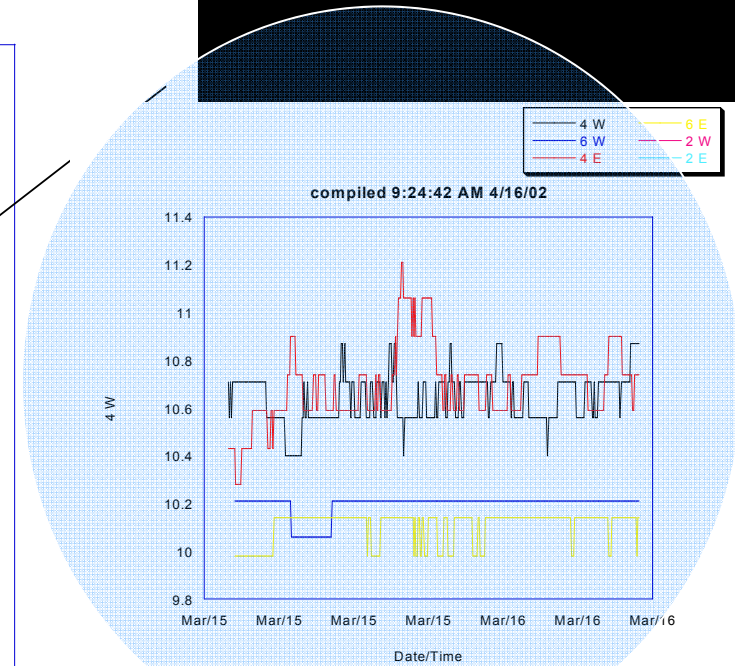
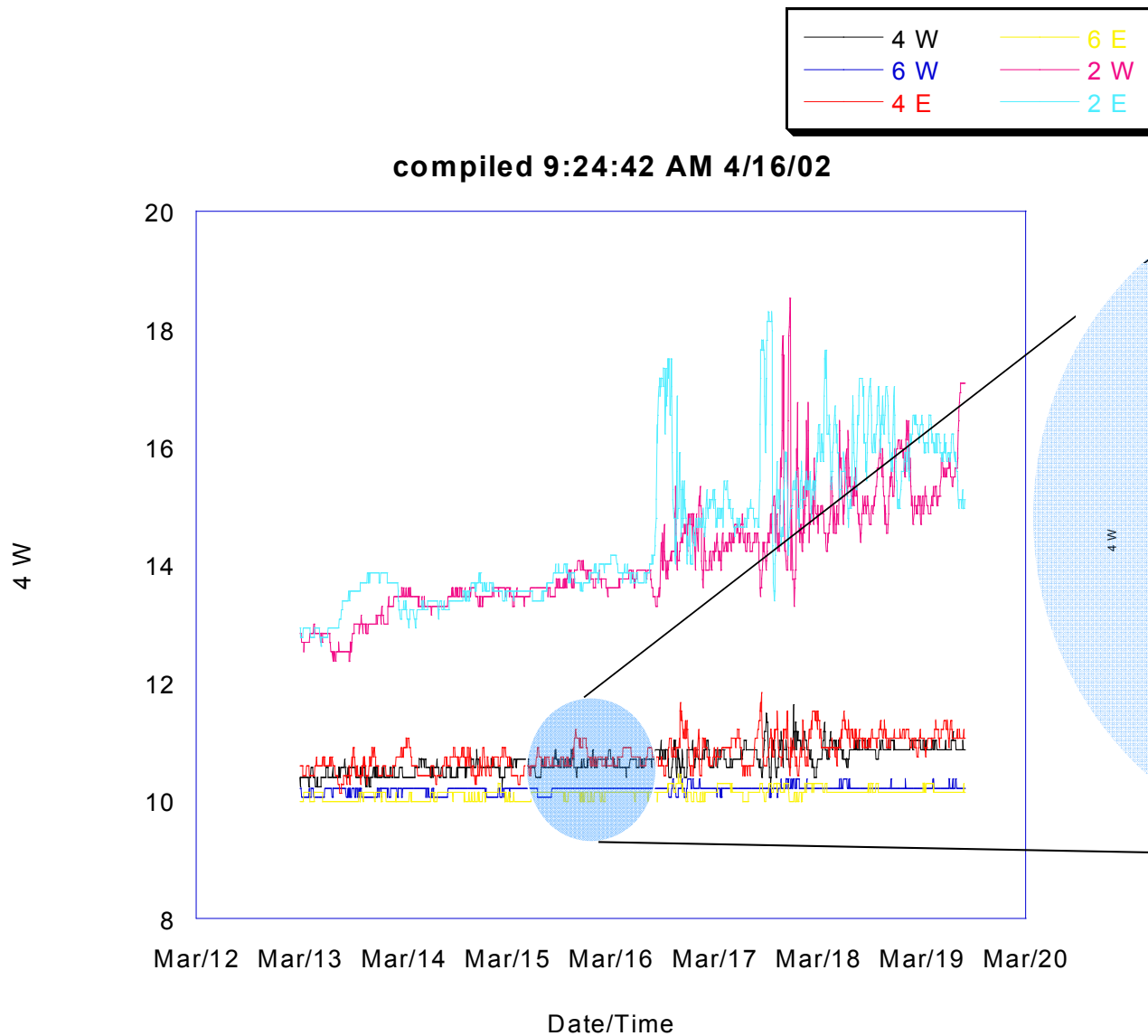
Development of meromixis in South Pit Lake

- 1) Lake bathymetry
- 2) Wind mixing
- 3) Energy dissipation



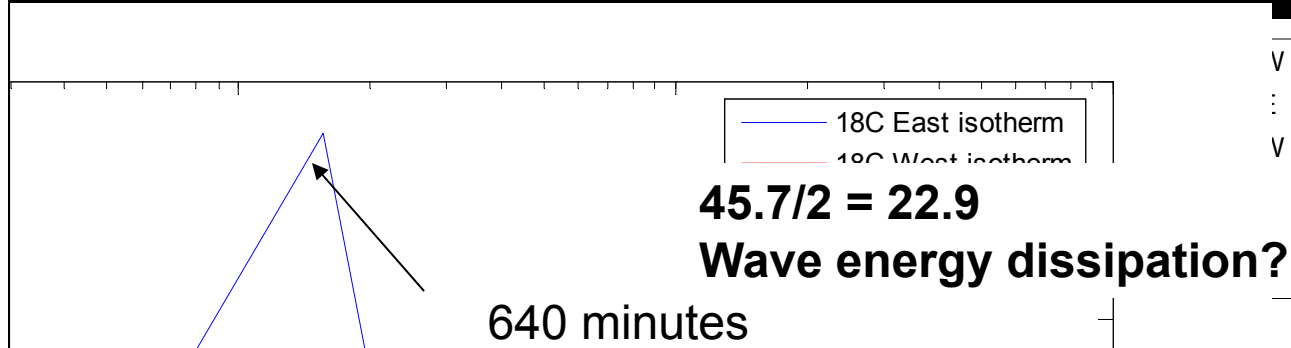
Physical

Using lake signatures to understand physical dynamics

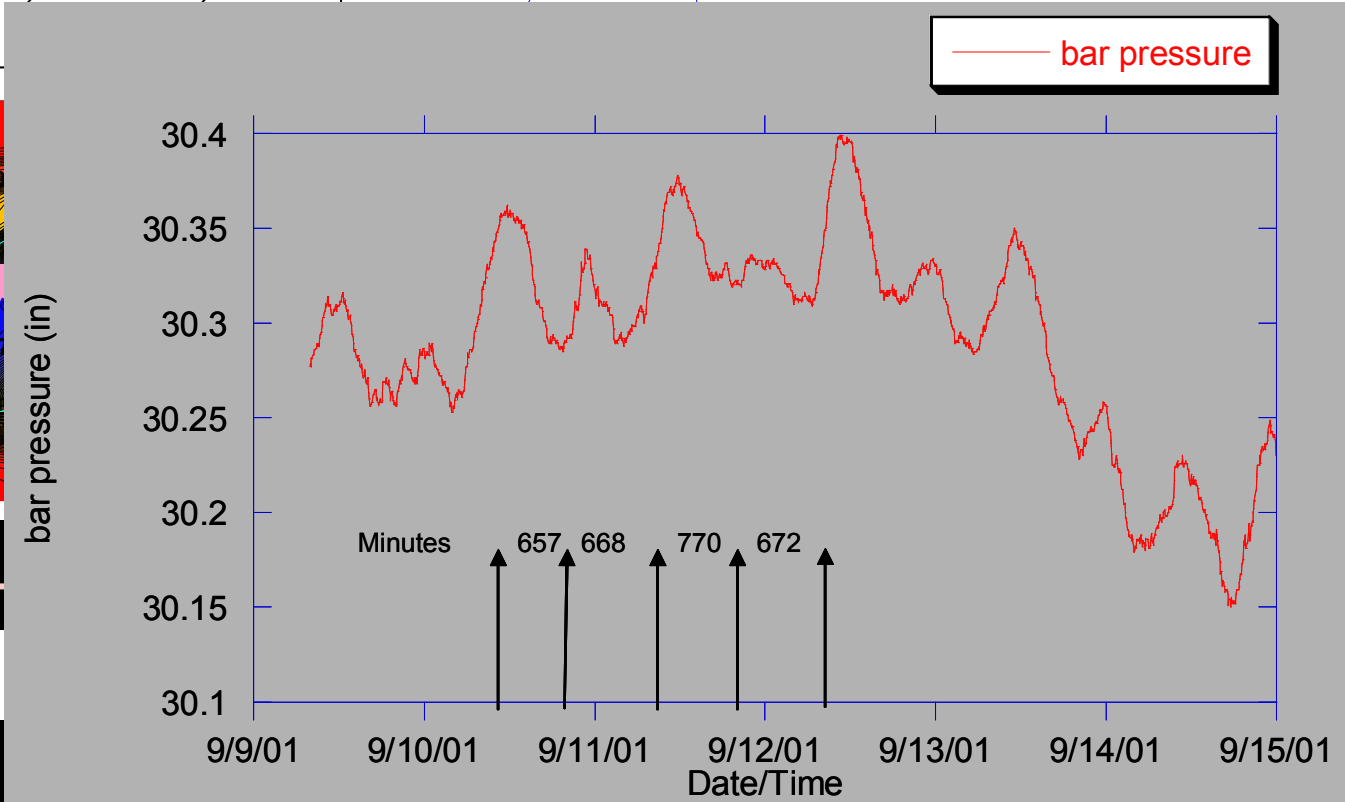


Did the bathymetry of SP help to dissipate the wind energy? Two approaches...

$$t = \frac{2l}{\sqrt{\frac{g(\rho_h - \rho_e)}{\frac{\rho_h}{7} + \frac{\rho_e}{7}}}}$$



44



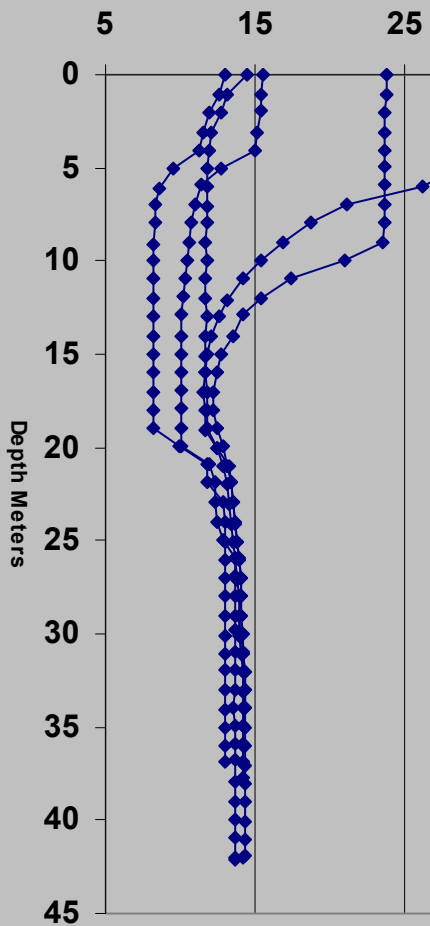
minutes

2.9 minutes

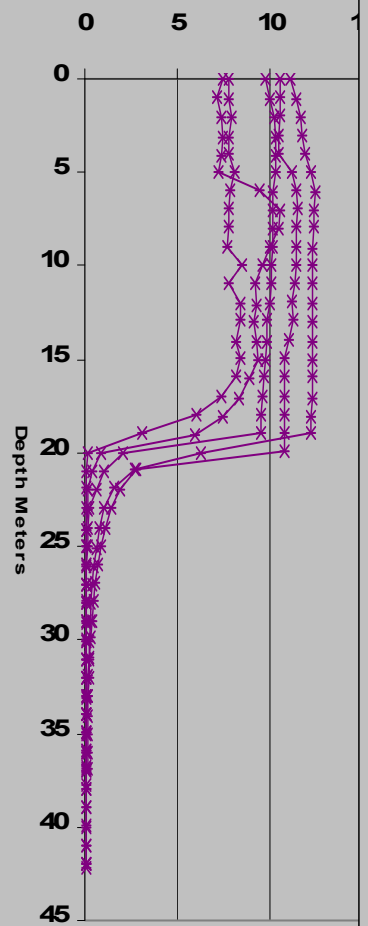
10⁻¹

2008-2010

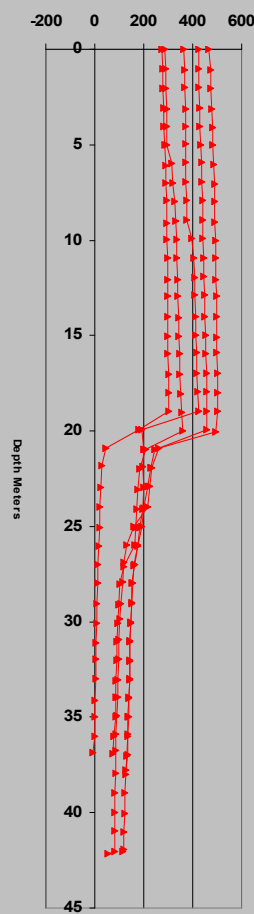
Temperature (C)



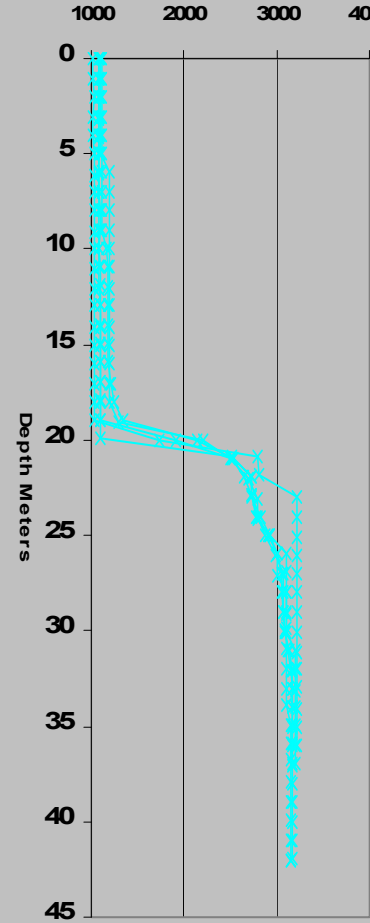
Dissolved oxygen (mg/L)



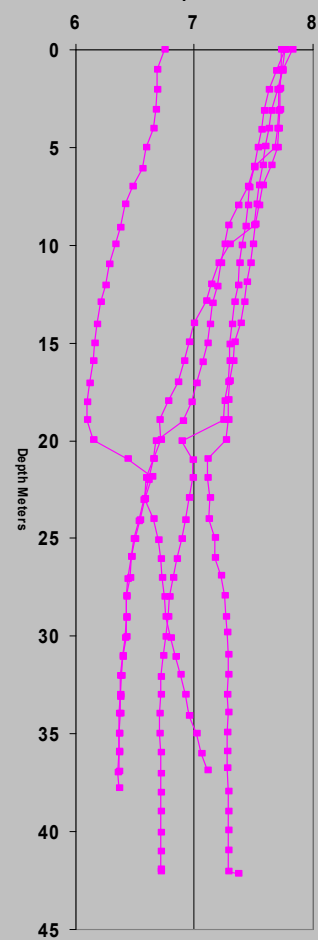
redox (mV)



conductivity (uS)



pH



Further research

- physical
 - determine extent of turbulent kinetic energy dissipation on sloping boundaries
- chemical
 - determine dynamics surrounding meromictic stability (is the chemical stability stabilized by either top-down or bottom-up flux?)
- biological
 - determine role of the biota in maintaining anoxia and meromixis and in sediment flux rates (algal biomass production in upper water and bacterial respiration rates in lower water)

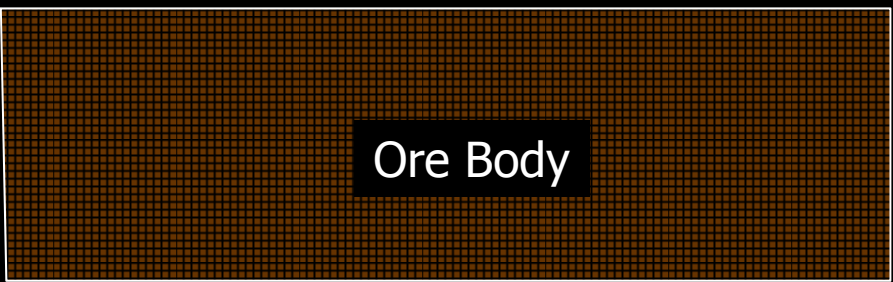
Further research

- This method has been proven in two pit lake systems in South Carolina, can it be done:
 - In colder climates?
 - For coal mining?
- Are there more applications for this type of treatment system?
 - Can the reaction chamber be done as a pass-through treatment system?

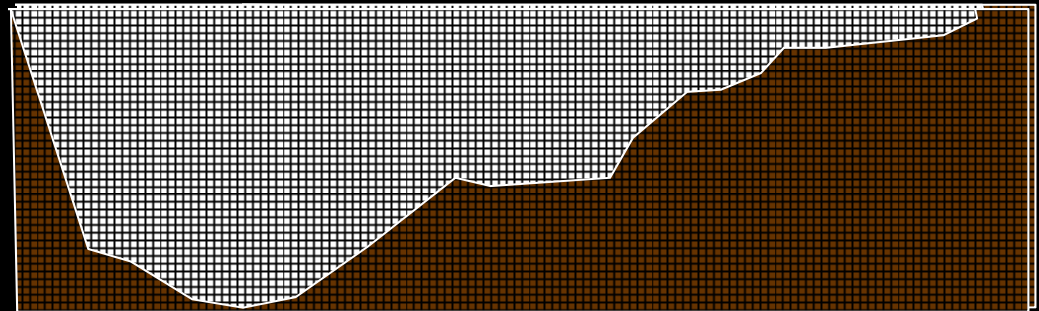
Final summation

- physical
 - lake shape helped minimize impact of wind by dissipating wind energy at boundaries
- chemical
 - system was in the carbonate system
 - precipitation and resolubilization of gypsum, carbonate, manganese, and iron established meromixis
 - iron and manganese oxides coprecipitated heavy metals from the water column
- biological
 - phytoplankton induced an environment conducive for gypsum, iron, manganese, and carbonate precipitation from the upper water to lower water
 - benthic bacteria induced an environment conducive for resolubilization and sediment flux in the lower water

- Prospecting results in ore body model
- Model partitions ore body into blocks
- Removal of blocks based upon:
 - costs of handling, processing, and recovery of that block
 - ore grade within that block
 - values of adjacent or lower blocks
 - current market price



•Ultimately, pit lake shape is determined by mining economics

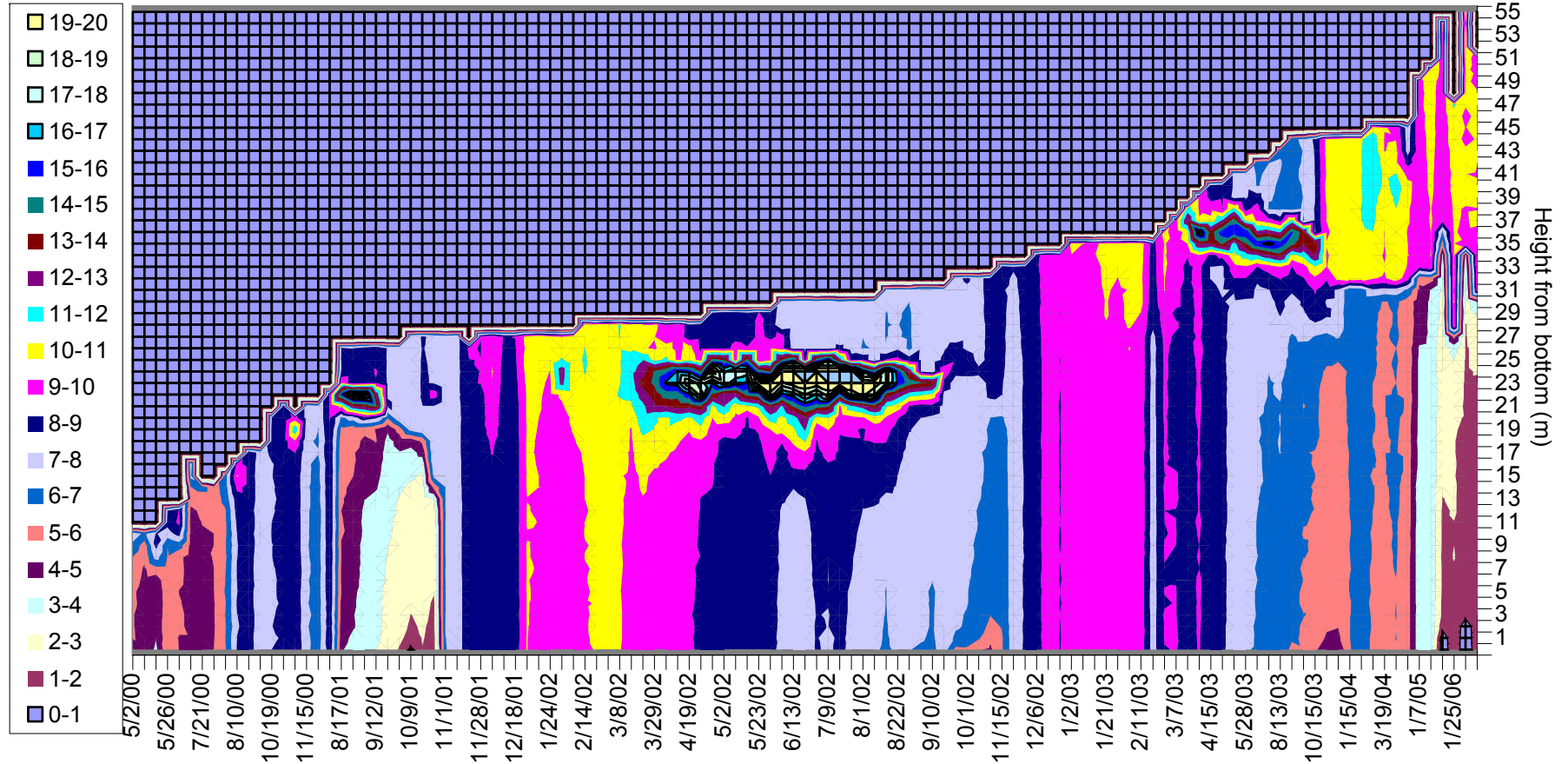


- Predominant wind direction
- Pit lake orientation optimized
- Placement of backfill optimized
- Alkalinity titration

Maximize pit lake sustainability

North Pit Lake

DO (mg/L)



pH

- 10.5-11
- 10-10.5
- 9.5-10
- 9-9.5
- 8.5-9
- 8-8.5
- 7.5-8
- 7-7.5
- 6.5-7
- 6-6.5
- 5.5-6
- 5-5.5
- 4.5-5
- 4-4.5
- 3.5-4
- 3-3.5

