

**3.5. MODELING ACIDIC DRAINAGE FROM
WASTE ROCK PILES**

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**MODELING ACIDIC DRAINAGE
FROM WASTE ROCK PILES**

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Objectives

*“critical review and discussion of mathematical models
of acidic drainage from waste rock piles ...”*

“emphasis on engineering models ...”

“reference to ... physical and conceptual modelling”



Topics covered in Report (and Presentation)

Terminology

Process - specific models

Engineering models

Recommendations



Terminology

“ ... discussions about modeling are often confused by imprecise or inconsistent terminology ”

conceptual vs. physical vs. mathematical models

empirical vs. mechanistic

deterministic vs. stochastic

comprehensive vs. process specific

engineering models

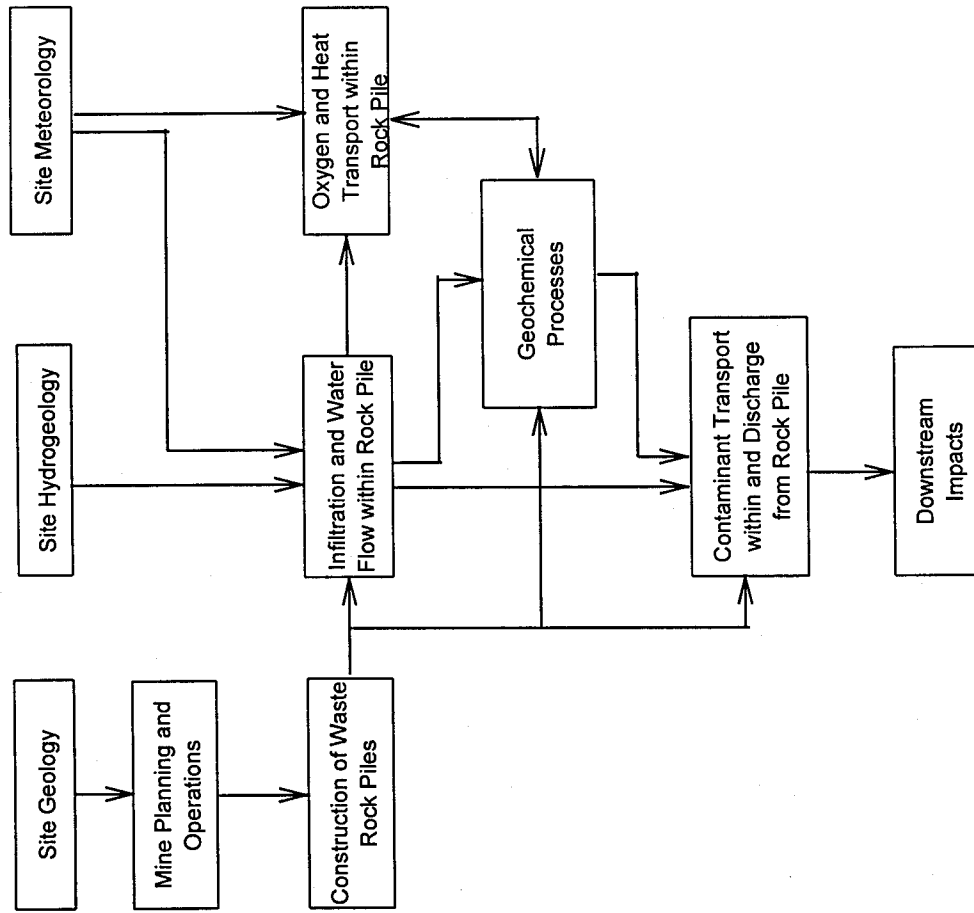


Process - Specific Models

*“models of individual processes,
or groups of coupled processes,
that contribute to, influence or are influenced by
acidic drainage in waste rock”*



Process - Specific Models



Process - Specific Models

External processes and variables

- *site geology & mine planning*
- *site hydrogeology*
- *site meteorology*

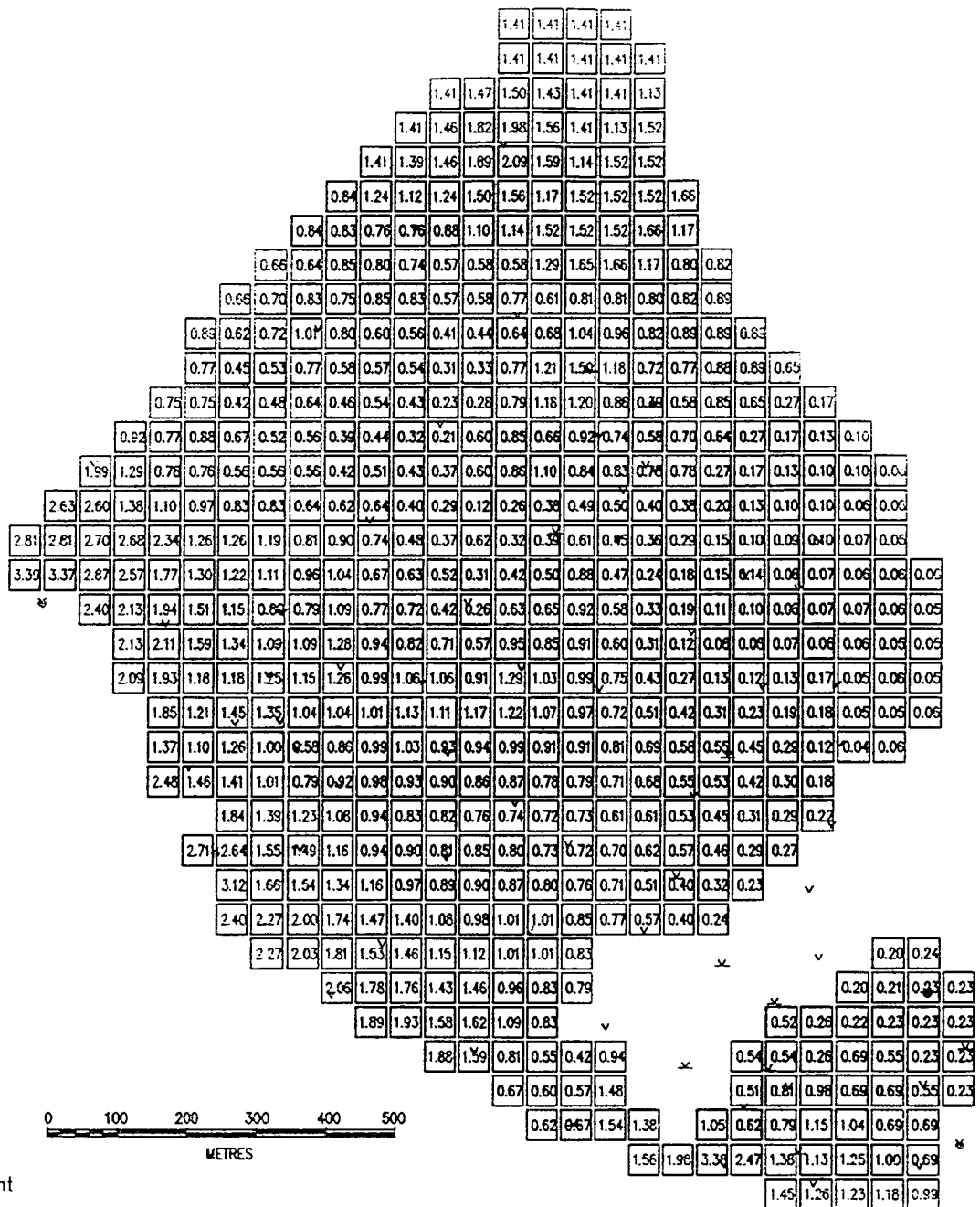


Process - Specific Models

Construction of waste rock piles

- *pile geometry*
- *distribution of material*
- *mixing of material during construction*
- *effects on physical characteristics*





Level number 11 320

Item 1 : probability

0.00 : 60.00

60.00 : 80.00

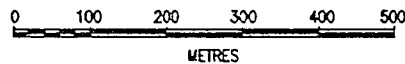
80.00 : 999.00

Item 2 : NP/SP

0.31 0.00 : 1.00

1.55 1.00 : 3.00

3.55 3.00 : 99.00



Drill hole pierce point

* Collar within bench

✓ hole enters and exits bench

x hole ends within bench



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NP:AP RATIOS FROM ORDINARY
KRIGING-ABSETZERHALDE LEVEL 300

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FIGURE

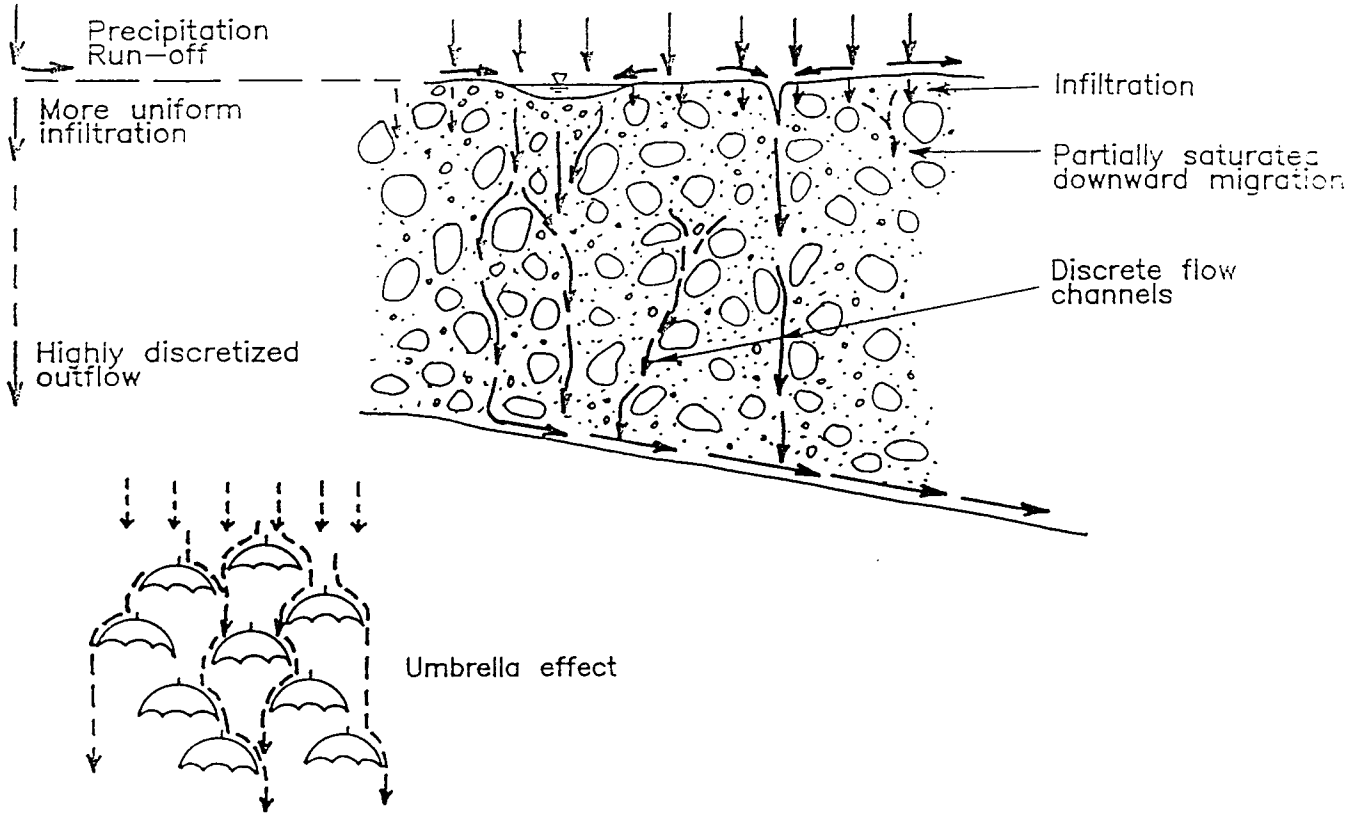
3.4

Process - Specific Models

Infiltration and water flow within waste rock piles

- *infiltration*
- *unsaturated flow*
- *channel flow*
- *saturated flow*





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MECHANISMS OF CHANNELING
IN WASTE ROCK

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FIGURE

3.7

Process - Specific Models

Oxygen and heat transport

- *advection*
- *thermal convection*
- *air phase diffusion*



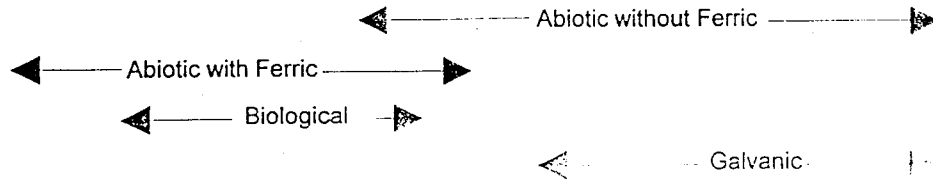
Process - Specific Models

Geochemical processes

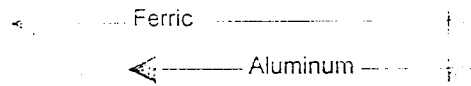
- *oxidation of sulphide minerals*
- *dissolution of carbonates, hydroxides, silicates*
- *precipitation of oxy-hydroxides*
- *precipitation and dissolution of sulphates*
- *co-precipitation, ion exchange, sorption*



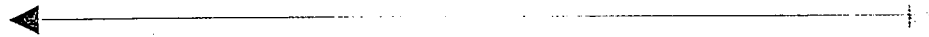
Oxidation of Sulphides



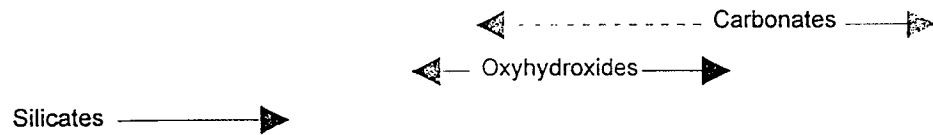
Precipitation of Oxyhydroxides



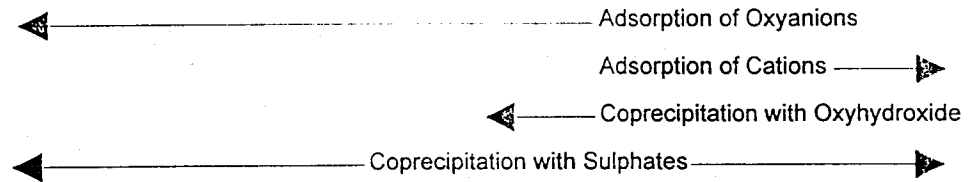
Dissolution/Precipitation of Sulphate Minerals



Dissolution of Neutralizing Minerals



Trace Metal Sorption



pH 1 2 3 4 5 6 7 8



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GEOCHEMICAL PROCESSES vs. pH

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FIGURE

3.8

Process - Specific Models

Contaminant transport within waste rock piles

Downstream impacts

Remediation measures

- *covers*

- *geochemical controls*

- *collection and treatment*



Engineering Models

*“models that are intended to support practical decisions
about real systems”*



Presented at the Fourth Annual B.C. ARD Symposium, Vancouver B.C., November 7-8, 1996

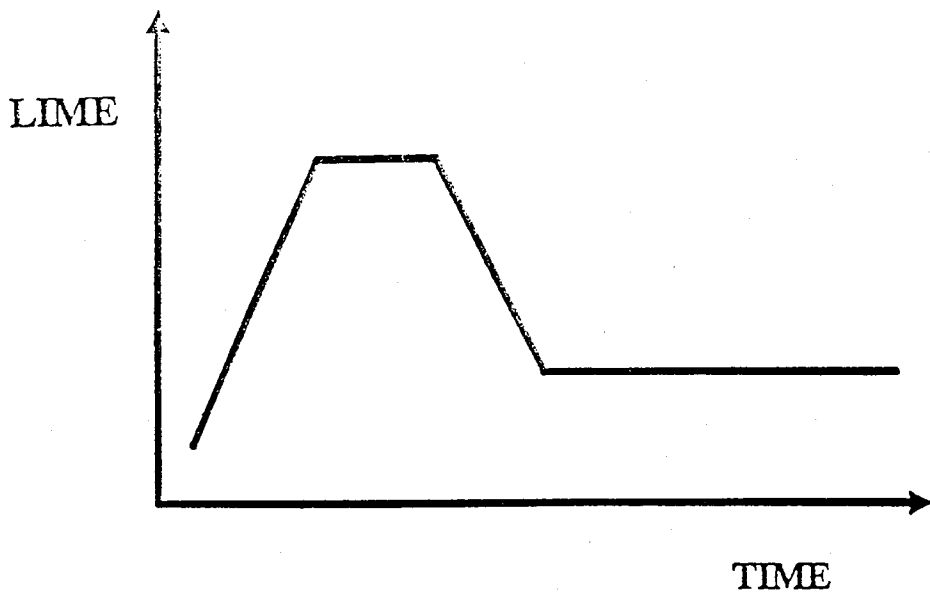
Engineering Models

Empirical models

Equity Silver Technical Committee (1991)

AMD Time - Ziemkiewicz (1994)





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**THE LIME CONSUMPTION CURVE USED
BY THE TECHNICAL COMMITTEE MODEL**

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| | | | |
|------------------------|--------------------|----------|---------------|
| PROJECT NO. S1202P9 | DATE SEPT.,1995 | APPROVED | FIGURE 4.1 |
|------------------------|--------------------|----------|---------------|

Engineering Models

Semi-empirical models

Morin & Hutt (1994)

QROCK - SRK (1993)

ACIDROCK - Senes (1991)

FIDHELM - ANSTO (1994)

TOUGH AMD - (Gelinias et al. 1994)



Input Parameters for FIDHELM Simulations

| Parameter | Values | Type * |
|---|-------------------------------|--------|
| ambient temperature (°C) | 3 | M |
| intrinsic air density (kg/m ³) | 1.2 | C |
| acceleration due to gravity (m/s ²) | 9.8 | C |
| air permeability (m ²) | $2.9 \times 10^{-9}, 10^{-8}$ | M |
| viscosity of air (kg/(m.s)) | 1.9×10^{-3} | C |
| thermal coefficient of volume expansion (K ⁻¹) | 3.47×10^{-3} | C |
| oxygen diffusion coefficient of oxygen in air (m ² /s) | 2.26×10^{-3} | C |
| parameter in particle oxidation model (m ³ /(kg.s)) | 1.354×10^{-9} | E |
| bulk density of solid (kg/m ³) | 2200 | M |
| specific heat of solid (m ² /(s ² .K)) | 866 | E |
| specific heat of air (m ² /(s ² .K)) | 1.06×10^3 | C |
| thermal diffusivity (m ² /s) | 5.0×10^{-7} | M |
| heat of oxidation reacton per mass of reactant oxidised (J/kg) | 2.2×10^7 | C |



| Parameter | Values | Type * |
|--|--|--------|
| mass fraction of oxygen in air | 0.22 | C |
| mass of oxygen used per mass of reactant in oxidation reaction | 1.746 | C |
| solid volume fraction | 0.763 | M |
| density of reactant (kg/m ³) | 118 | M |
| maximum intrinsic oxidation rate (kg (O ₂)/m ³ /s) | 10 ⁻⁹ , 10 ⁻⁸ , 10 ⁻⁷ | E |
| infiltration rate (m/y) | 1.134 | M |
| viscosity of water (kg/m s) | 0.001 | C |
| intrinsic water density (kg/m ³) | 1000 | C |
| specific heat of water (m ² /(s ² .K)) | 4.184 x 10 ³ | C |
| temperature at which micro-organisms cease to be effective as catalysts (°C) | 50 | E |
| temperature above which the micro-organism catalytic activity diminishes | 40 | E |
| tortuosity factor | 0.59 | M |
| radius of pile (m) | 34 | M |
| height of pile (m) | 6.7 | M |
| side slope angle of heap (radians) | 0.245 | M |



Conclusions

Process - Specific Models

- *surprising number & diversity*
- *some processes well understood & modeled*
- *other (important) processes neglected*
- *current approach is "bottom up"*



Conclusions

Engineering Models

- *all are at least partly empirical*
- *empirical models are more transparent*
 - *poor documentation*
- *lack of consideration of uncertainty*
 - *lack of validation*
 - *“bottom up”*



Recommendations

- *limitations of ARD models need to be better communicated to industry and regulators*
- *strong “bottom up” work should be complimented by a “top down” approach*
- *appropriate combination of mathematical and physical models needs more attention*

