

**C.4 Microbial Ecology of Anaerobic Sulfate- and Iron-Reducing Bacteria in Sulphidic Mine Tailings**

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## Microbial ecology of anaerobic sulfate- and iron-reducing bacteria in sulfidic mine tailings

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## What do we know about sulfidic mine tailings?

- They contain large amounts of pyrite and pyrrhotite
- They often oxidize and generate acid-mine drainage
- Fe- and S-oxidizing bacteria play a significant role in the oxidation of sulfides minerals at low pH
- Very little is known about the activity of anaerobic bacteria, such as sulfate-reducing bacteria (SRB) and iron-reducing bacteria (IRB)

## Microbial ecology of SRB

- SRB generally prefer reduced conditions, neutral pH and use simple electron donors, such as acetate, formate, lactate, H<sub>2</sub>, etc.
- However SRB have been identified in acidic mining lakes and they have been recovered in Cu-Zn and Au tailings, but they appear to be only active in Cu-Zn tailings
- Microbial sulfate reduction leads to the formation of metal (Fe) sulfides and generates alkalinity

## Microbial ecology of IRB

- They require reduced conditions and can sustain slightly acidic conditions, even though they prefer neutral pH
- They can use acetate, glucose, H<sub>2</sub>, etc. as electron donors
- They can reduce various forms of Fe(III)-oxides
- IRB have been recovered from mining lakes and streams affected by mine drainage, but not (so far) in sulfidic tailings.
- Microbial iron reduction generates soluble Fe(II) and alkalinity

## Why do we study IRB and SRB?

- They directly affect the mobility (fate) of Fe and S in mine tailings
- Microbial sulfate and iron reduction both generate *in situ* alkalinity
- *In situ* formation of diagenetic Fe-sulfides and reduction of sulfate levels
- They both have an impact on the mineralogical and chemical stability of the tailings

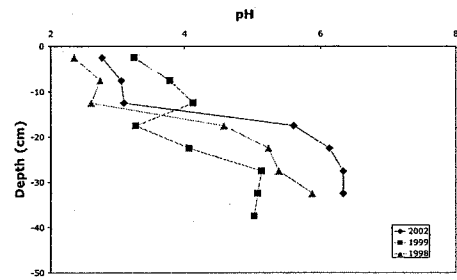
## Objectives

- Assess the presence of SRB and IRB in two types of sulfidic tailings (acidic and alkaline tailings)
- Link their presence to porewater chemistry
- Determine the *in situ* rate of sulfate reduction
- Determine the factors affecting their presence and activity

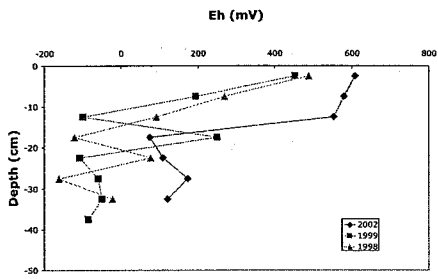
## Methodology

- One acidic and oxidized site in Northern Ontario (PO-01) and one alkaline site near Ottawa (organic-rich) (CA-01)
- Samples taken along a 50-cm deep profile every 5 cm
- IRB and SRB assessed by MPN with growth media adjusted to the *in situ* pH and containing various electron donors
- Porewaters extracted from the tailings and also collected with peepers
- Analysis of dissolved Fe(II), Fe (total), SO<sub>4</sub>, HS<sup>-</sup>, DOC/DIC, pH, Eh, etc.
- *In situ* rates of sulfate reduction (<sup>35</sup>S core-injection technique)

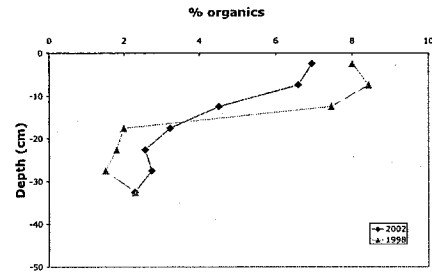
## PO-01



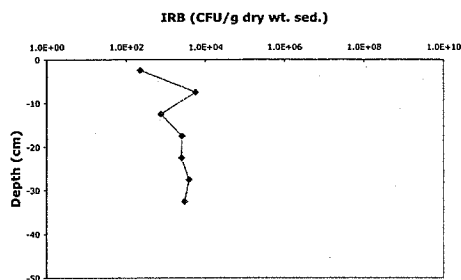
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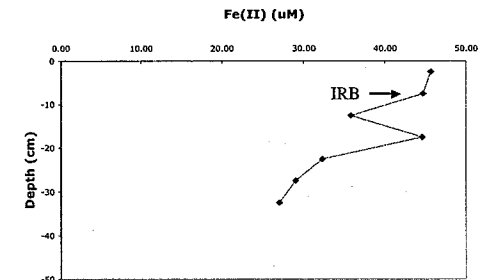
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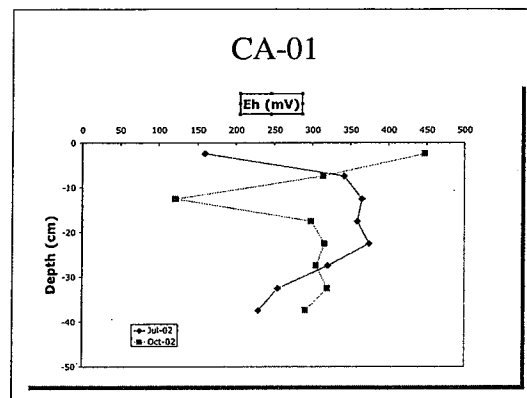
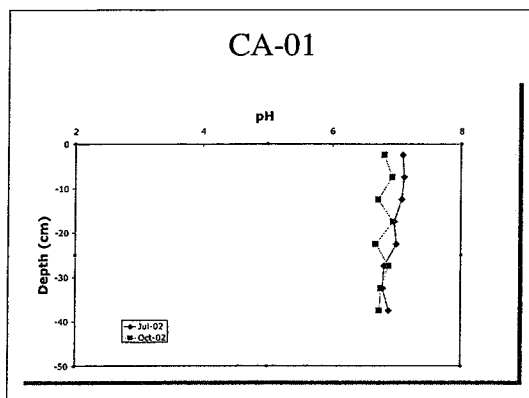
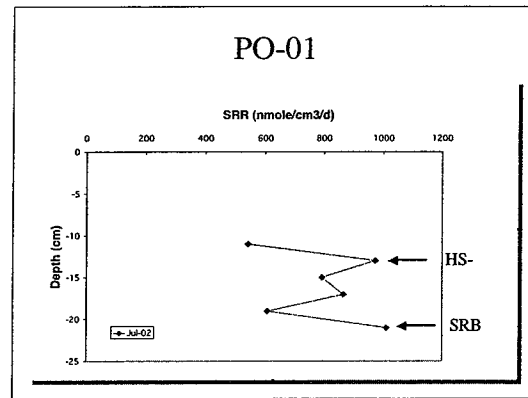
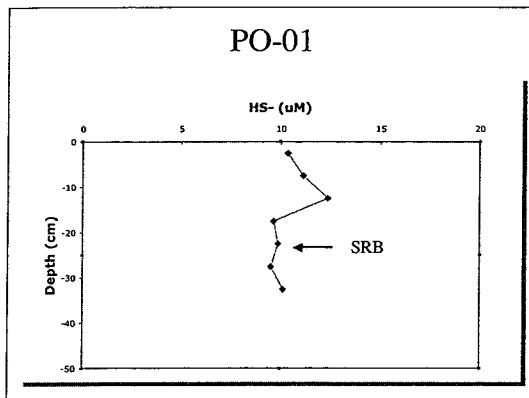
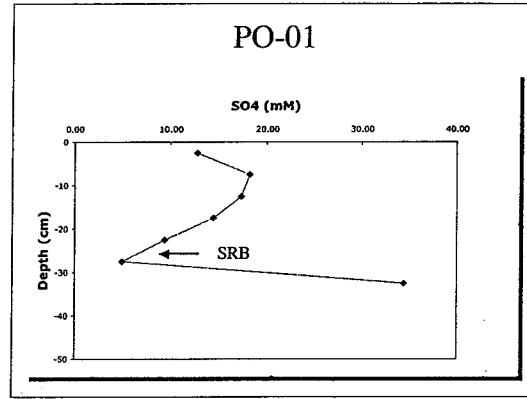
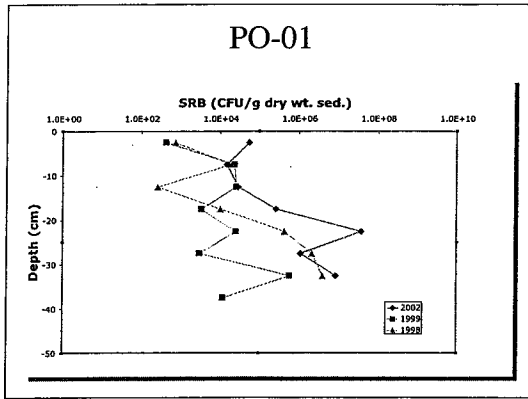


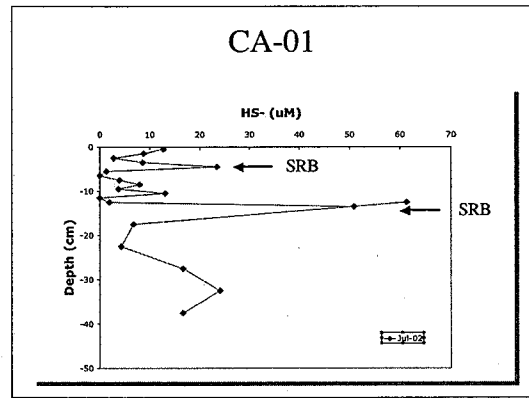
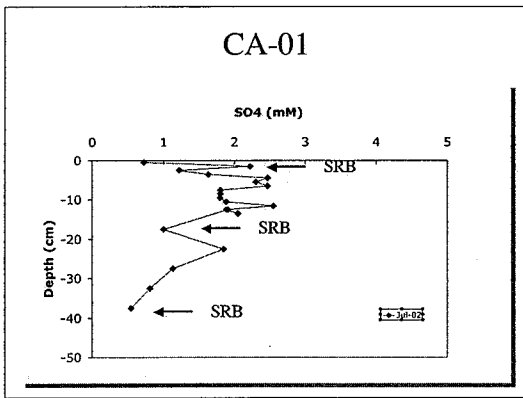
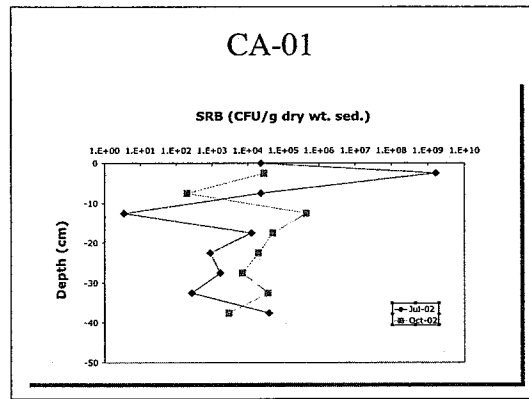
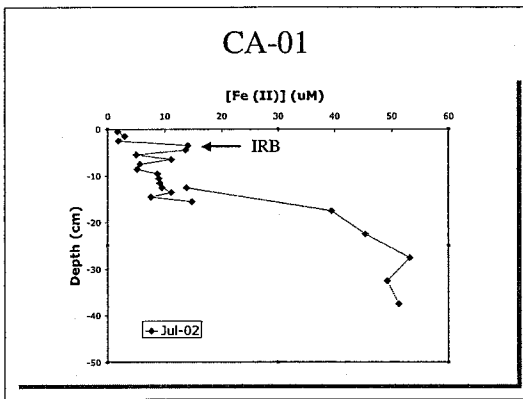
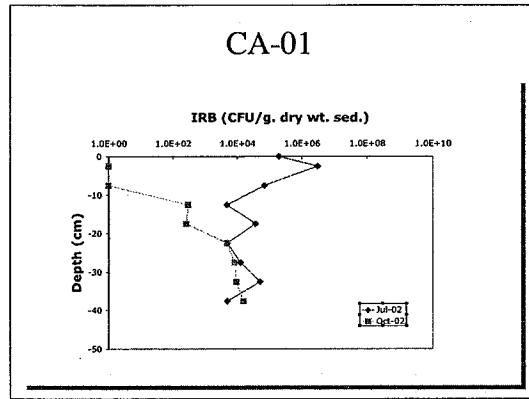
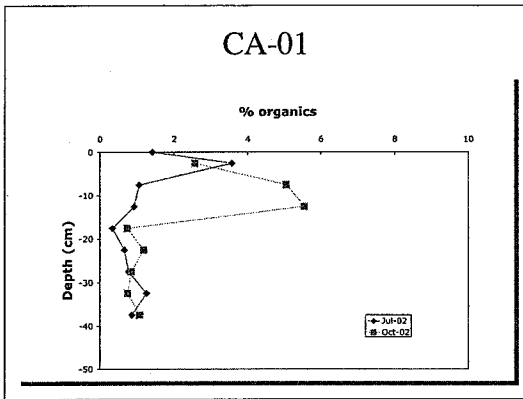
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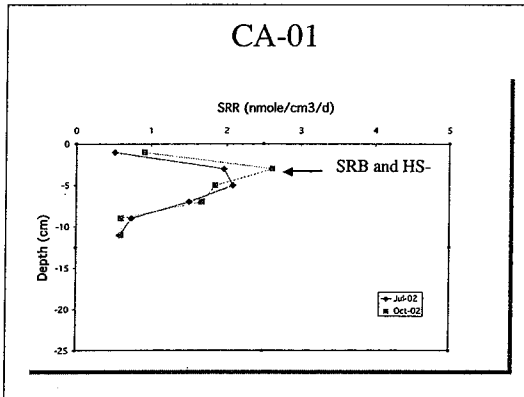
## PO-01







## CA-01



## Discussion

- IRB and SRB present and active at both sites
- SRR higher in acidic tailings containing large amounts of sulfide minerals
- Active zones of sulfate reduction vary with depth and seasons in the alkaline tailings
- Sulfide and Fe(II) concentrations both controlled by microbial activity and chemical reactions (abiotic reduction of Fe(III)-rich minerals by HS<sup>-</sup>)
- Microbial Fe(III) and sulfate reduction partially responsible for pH increase in sub-surface

## Acknowledgements

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