

# **Evolution of Mine Drainage Chemistry Red Dog Mine, Alaska**

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# Outline

- ⇒ Mine background
- ⇒ Geological overview of Red Dog
- ⇒ Static geochemistry
- ⇒ Kinetic geochemistry
- ⇒ Waste rock seepage chemistry
- ⇒ Mechanisms controlling seepage chemistry

# Red Dog Mine



- ➔ Located in NW Alaska  
55 miles from Chukchi  
Sea
- ➔ Largest zinc deposit  
in the world.
- ➔ Started production in  
1989.
- ➔ Open pit mine
- ➔ Conventional sulphide  
flotation

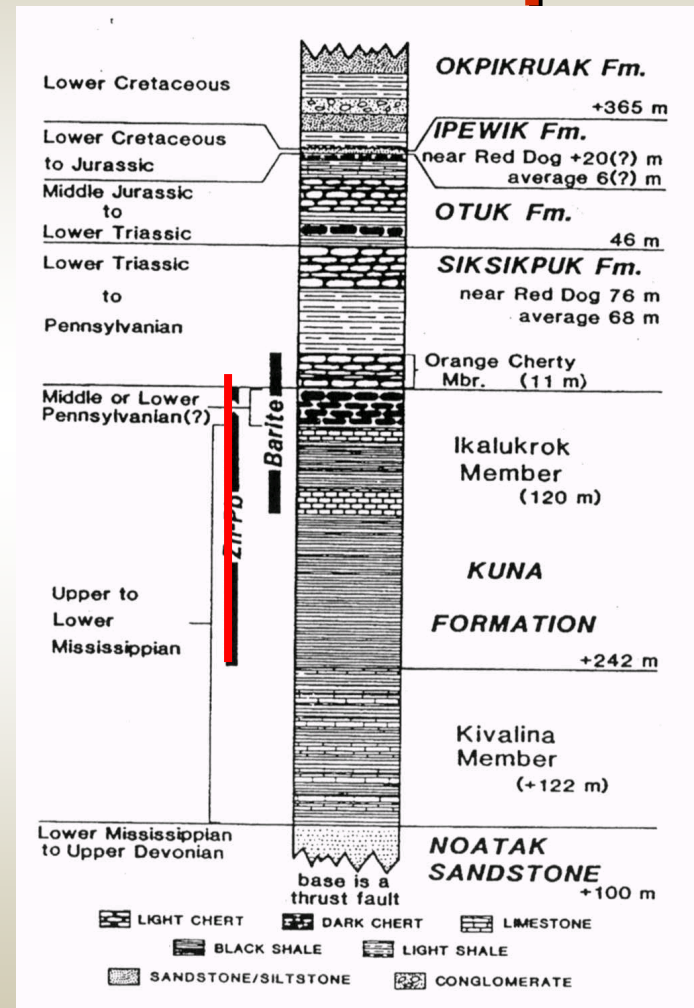




# Black Shale-Hosted Pb-Zn Deposit

## ⇒ Geology

- Friable rocks
- Generally low carbonate content.
- Wide range of sulphide mineral occurrences in ore and host rock

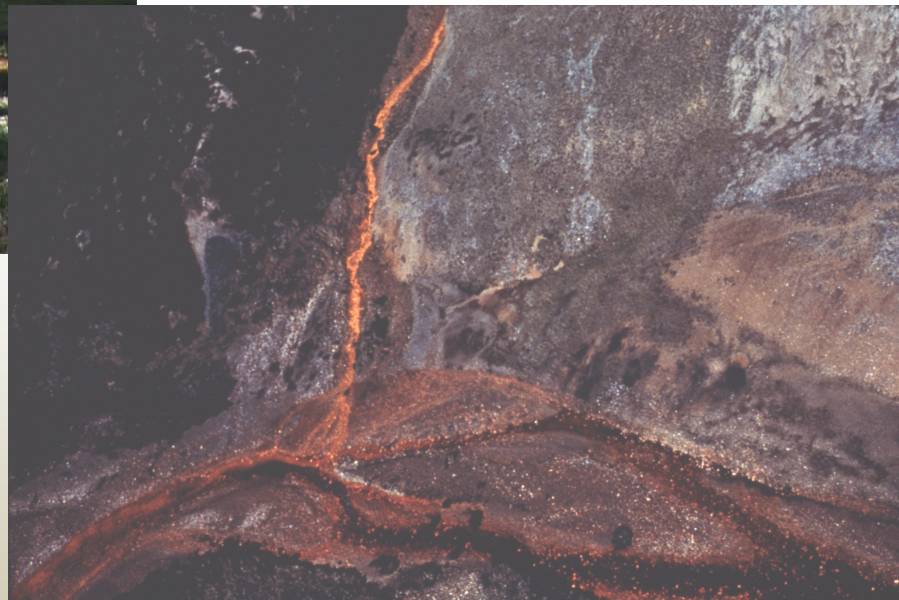
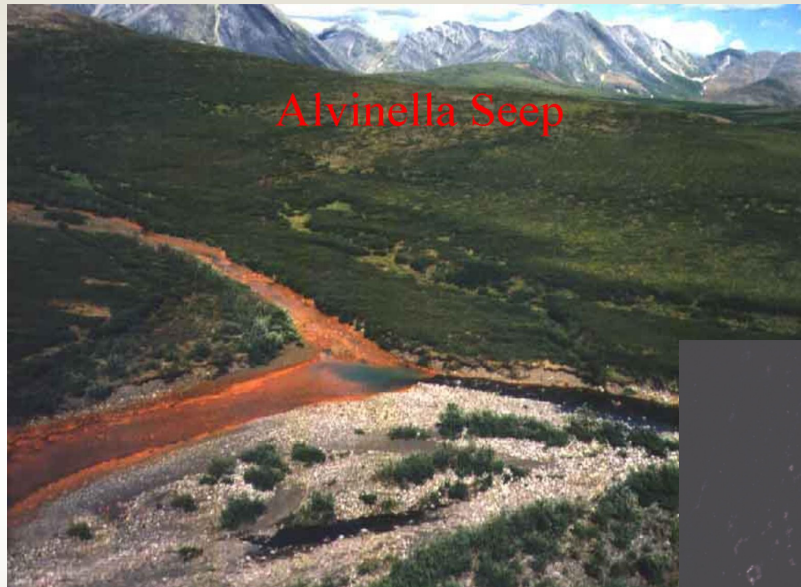




# Natural Weathering Effects

- ⇒ Deposit discovered by oxidized cap (“gossan”)
- ⇒ Natural oxidation has produced a large number of iron, zinc, lead and copper oxides and sulphates.
- ⇒ Acidic salts and natural seepage.

# Natural Weathering Effects



# Oxidation Salts





# Static Geochemistry: Sulphur Speciation

## ⇒ Complex sulphur speciation

- Conventional static testing is meaningless.
- Sulphur speciation calculated from dominant forms:

$$\text{S(Fe Sulphide)} = \text{Total S} - \{ \text{S(Ba)} + \text{S(PbS)} + \text{S(PbSO}_4\text{)} + \text{S(ZnS)} + \text{S(ZnSO}_4\text{)} \}$$

- Acid Potential from S(Fe Sulphide)

## ⇒ Day et al (2000 – Denver ICARD)

# **Static Geochemistry: Neutralization Potential**

- ⇒ Carbonate content of most rock types is very low.**
- ⇒ Mostly calcite.**
- ⇒ Iron and barium carbonates occur.**

# Geochemical Database

## ⇒ Waste Rock Characterization

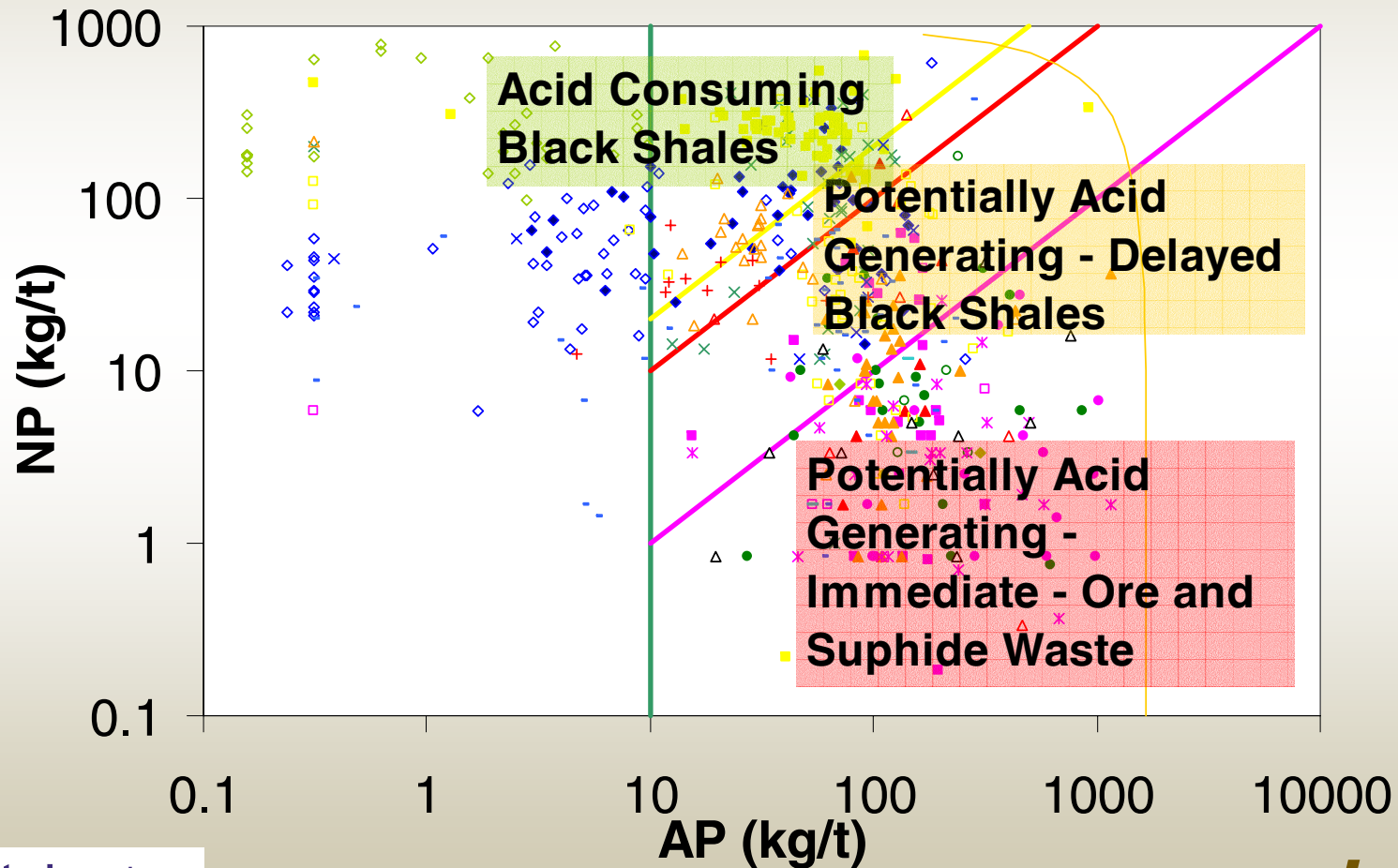
- 600 static tests
- Eight humidity cells (up to 4 years)
- Seepage monitoring for waste rock (7 years)

## ⇒ Tailings

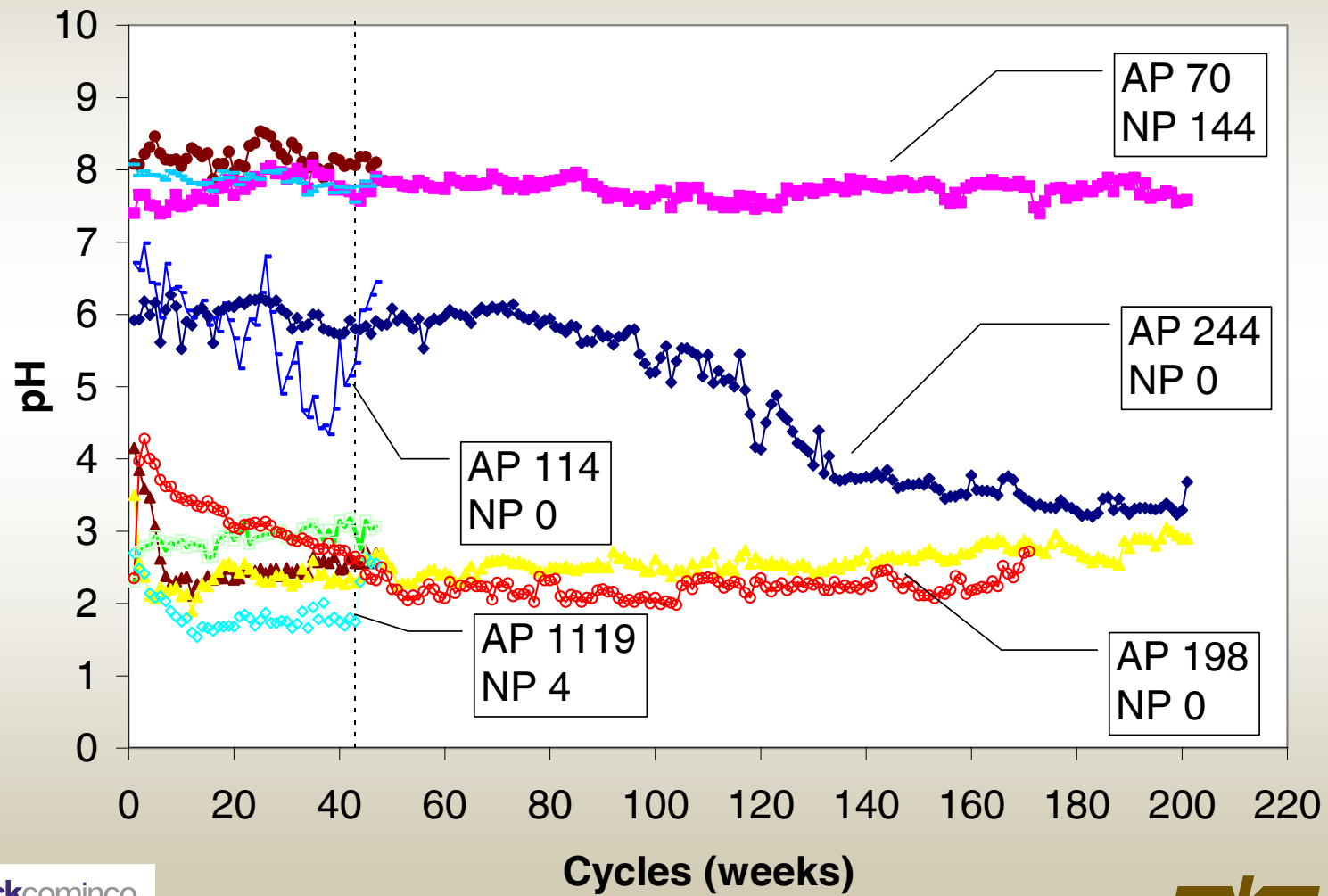
- Humidity cells and subaqueous columns on one sample (3 years)
- Seepage monitoring (5 years)



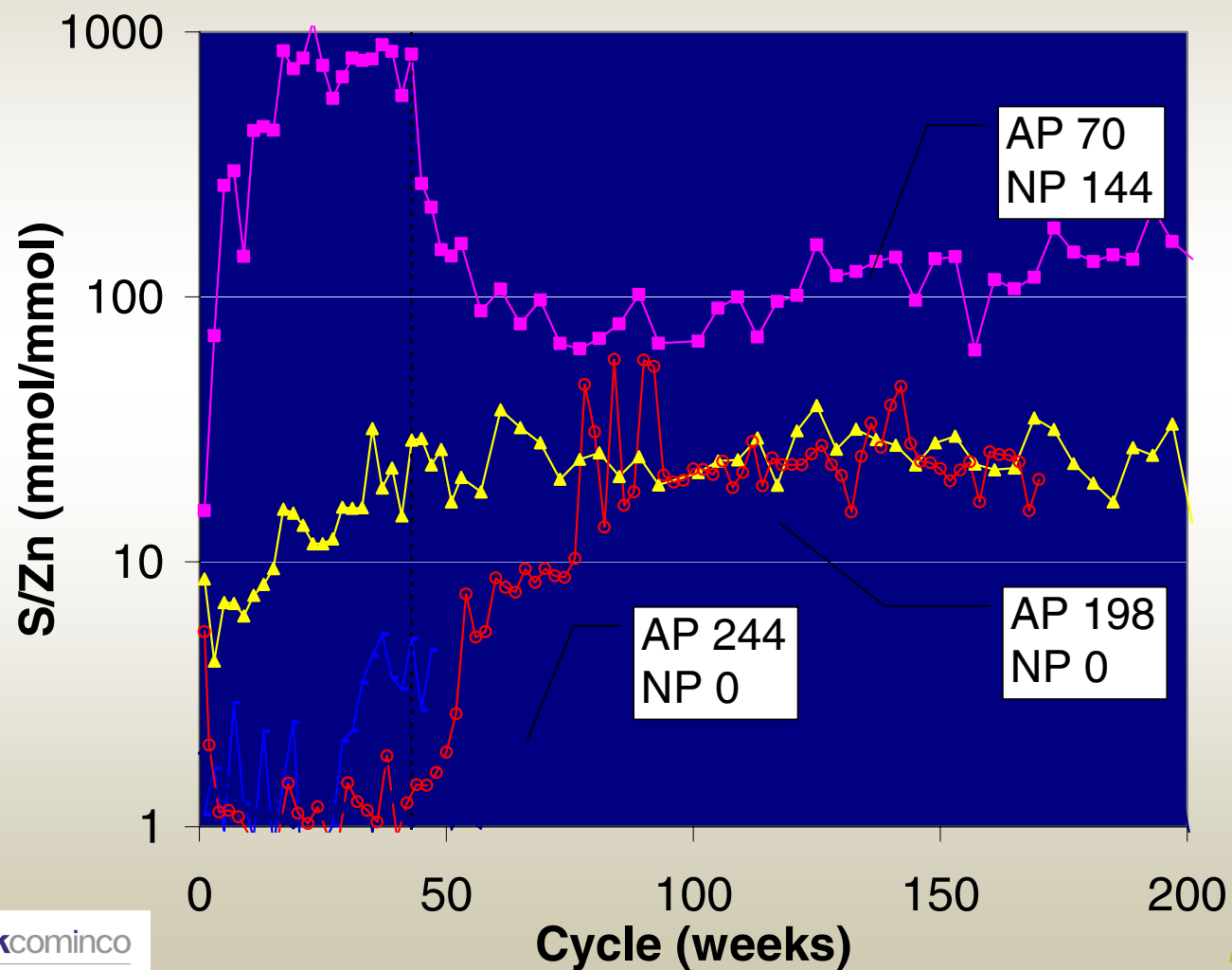
# Acid Generation Potential



# Humidity Cell Results

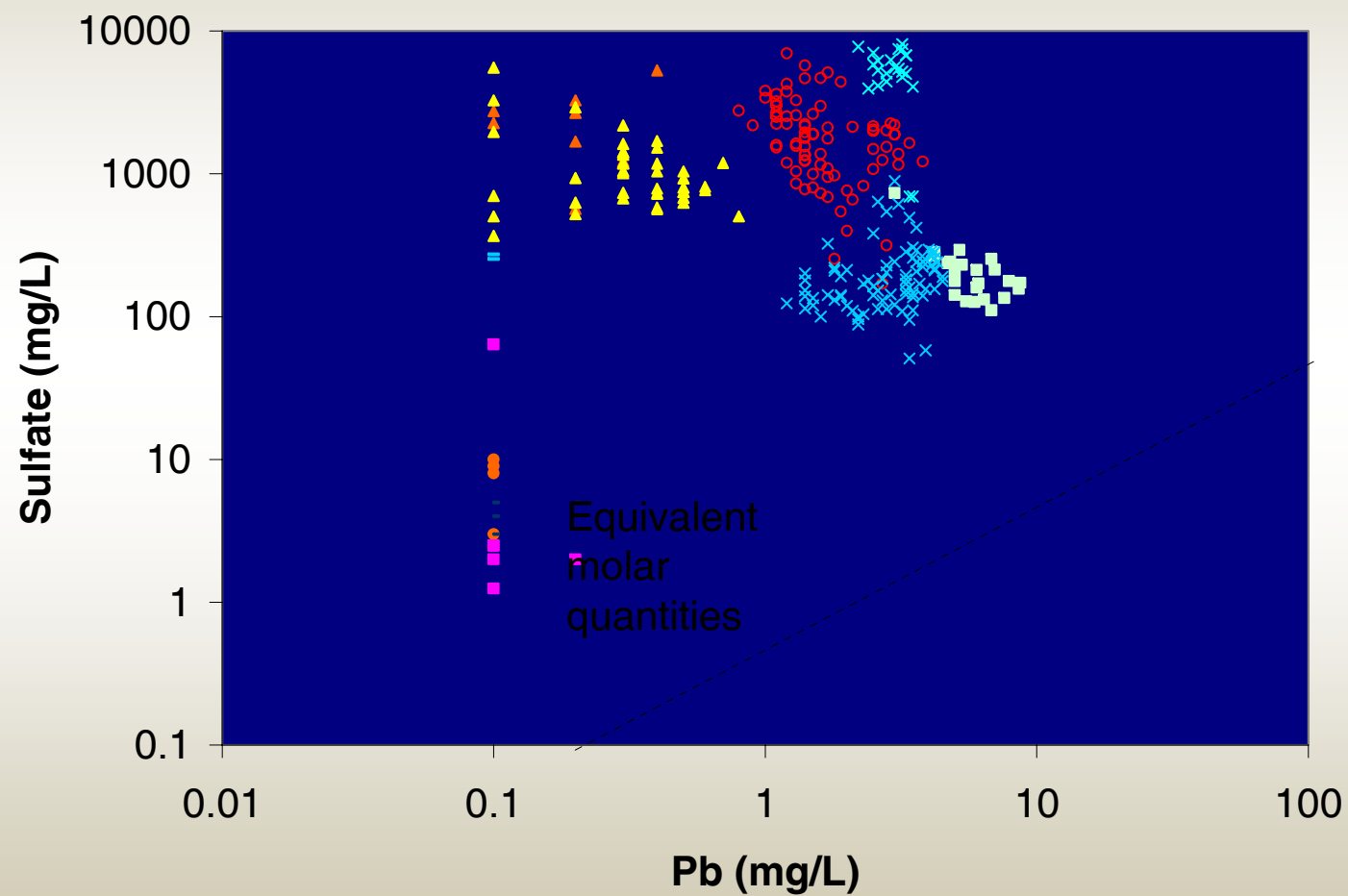


# Humidity Cell Results





# Humidity Cell Results

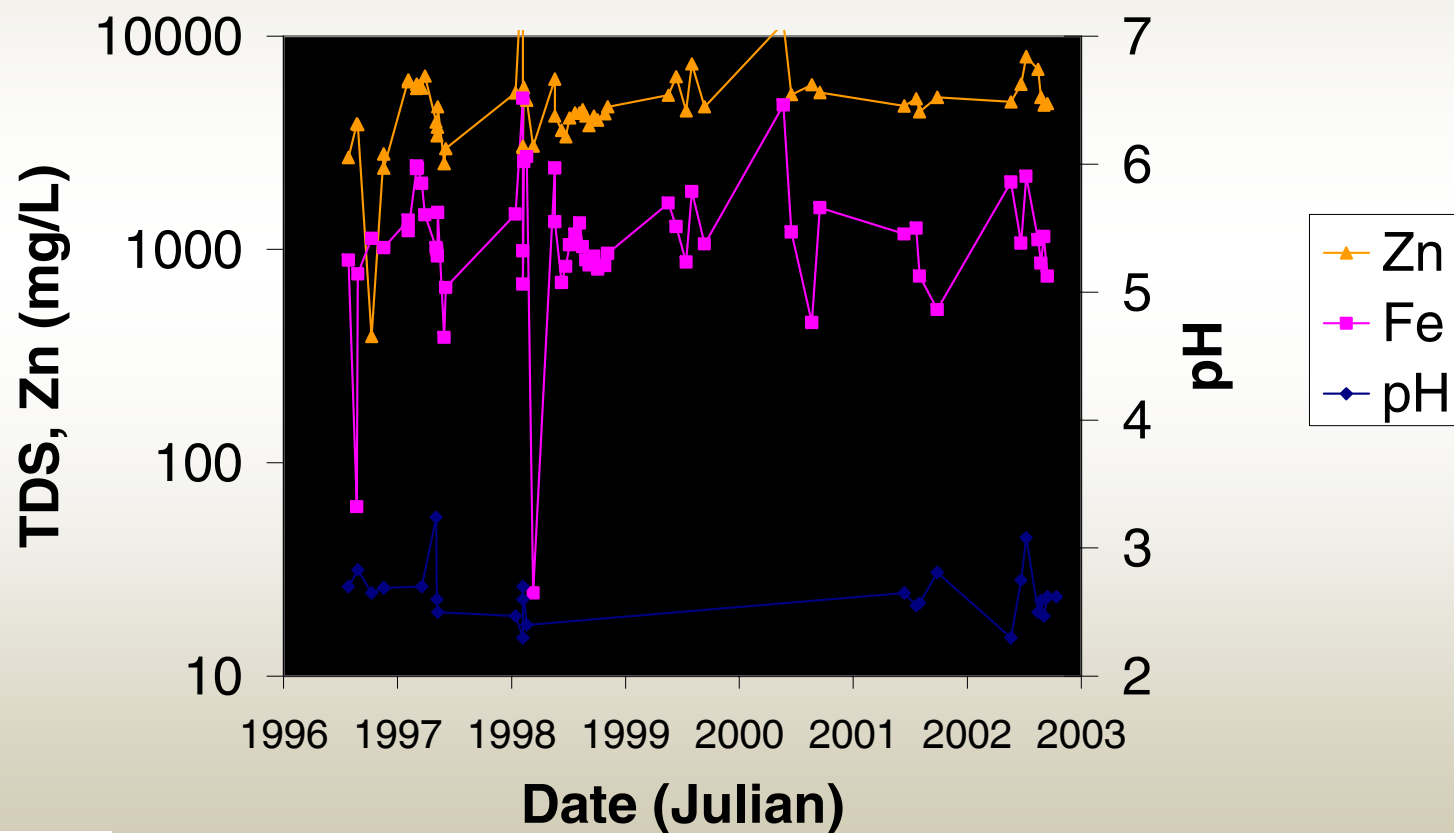


# Waste Rock Characteristics

Major Unit	Sub-Unit	Overall Classification	
		Primary	Secondary
Mélange (8%)	Basal and Upper	Non-AG, metals	AG
Kivalina Shale (5%)		Non-AG, metals	
Kogruk Limestone (0%)		Non-AG	
Ikalukrok	Ore host (55%)	Rapid AG	
	Shale (6%)	Rapid AG	Non-AG, metals
Siksikpuk (21%)	Baritic	Delayed AG	
	Chert	Uncertain AG	
	Shale	Rapid AG	Non-AG, metals
Okpikruak (4%)		Non-AG	

# Waste Rock Stockpile Seepage

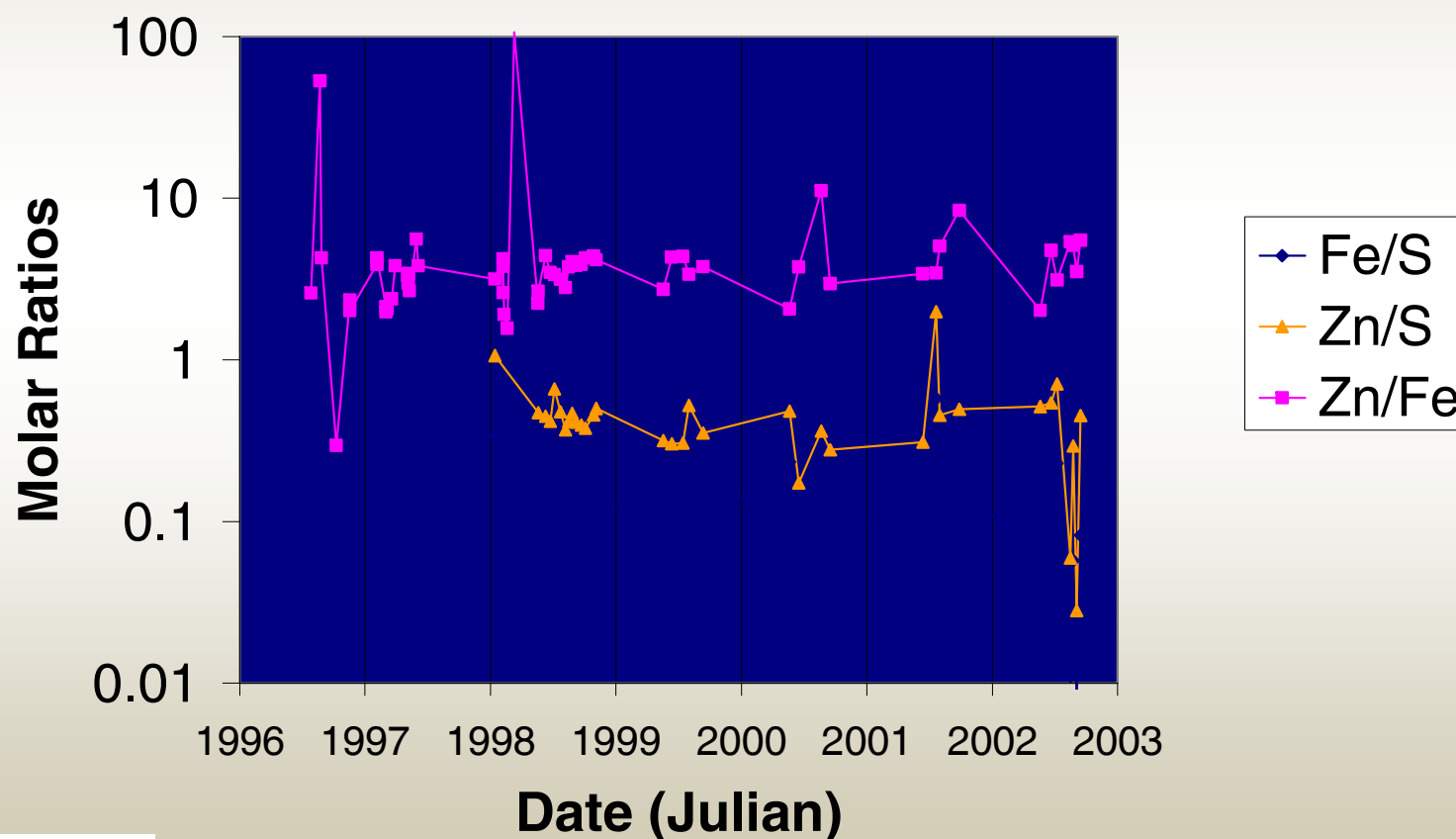
MWD-18





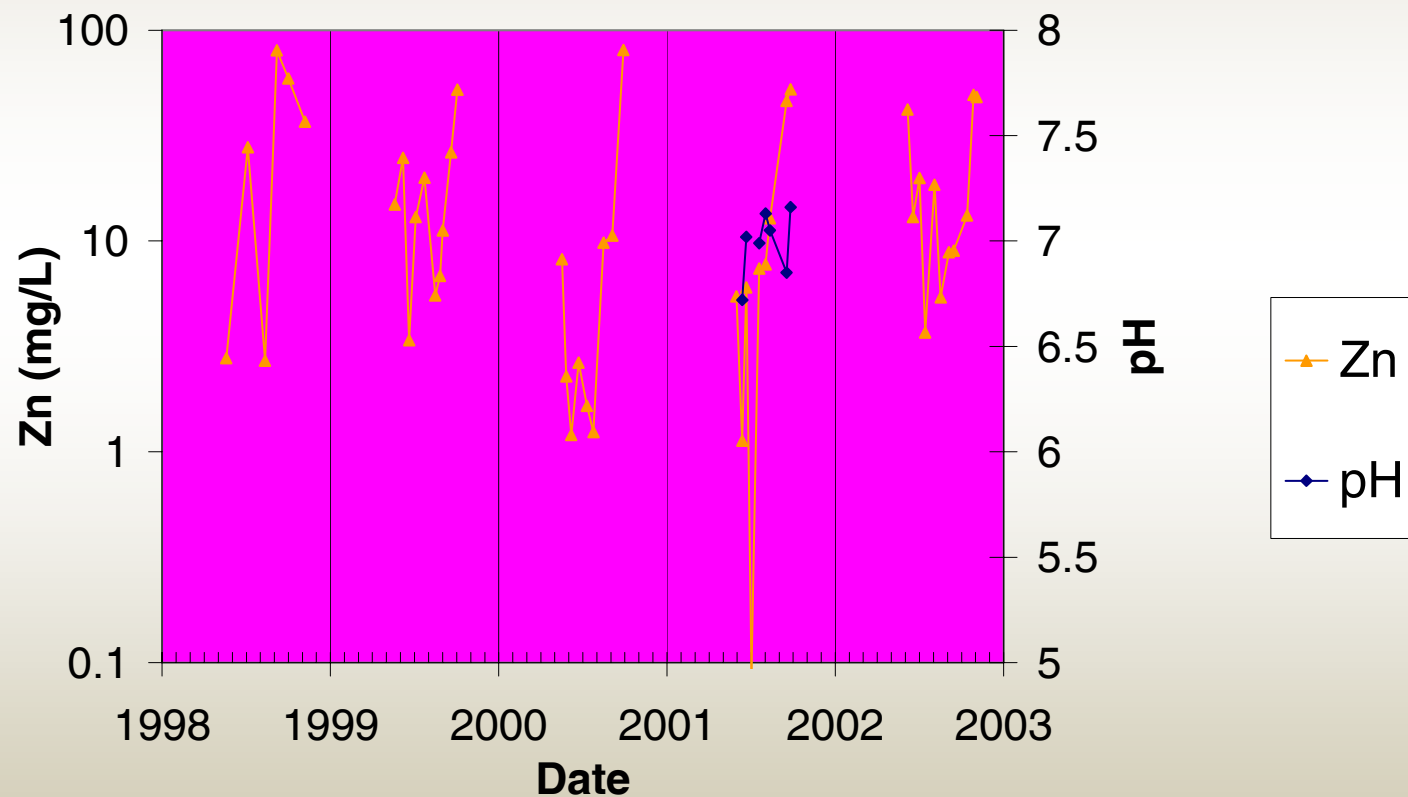
# Waste Rock Stockpile Seepage

## MWD-18



# Overburden Stockpile Seepage

## East Sump



# Observations on Trends

## ⇒ Humidity Cells

- Initial flush of (highly acidic) salts.
- Conventional delay of pH depression by carbonates.
- Other delay effect
  - Possible galvanic effect (ZnS oxidizes first)
- Lead release increases late in test as  $\text{SO}_4$  decreases.



# Observations on Trends

## ⇒ Waste Rock Seepage

- Main waste seeps are strongly acidic.
- Seeps monitored for several years indicate flat or slightly decreasing trends.
  - Dominance of zinc over iron
- Overburden waste pile has non-acidic seepage
  - No trend.

# Conclusion – Evolution of Drainage Chemistry

## ⇒ Immediate to Short to Medium Term Effects (Zn, Cd, mildly acidic to alkaline)

### – Immediate to Short Term

- Leaching of natural (original) soluble salts containing mostly zinc, cadmium ( $\pm$  iron).

### – Short to Medium Term

- Oxidation of zinc sulphides.
- Galvanic protection of iron sulphides.
- Carbonate buffering (if carbonates present).
- Leaching of soluble salts produced by oxidation

# **Conclusion – Evolution of Drainage Chemistry**

## **⇒ Longer Term Effects (acid rock drainage)**

### **– Medium to Long Term**

- **Zinc sulphide oxidation continues**
- **Oxidation of iron sulphides**
- **Leaching of soluble salts produced by oxidation**

### **– Long Term**

- **Leaching of lead sulphate (increase in lead)**

# Thank you!