Mechanisms of Arsenic Attenuation in a Gold Mill Tailings Impoundment

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Overview

- Background and Site Setting
- Objectives
- Arsenic Attenuation Mechanisms
 - Diffusive pathway
 - > Advective pathway
- Implications for tailings/water management





Site Setting – Cochenour, ON



Cochenour Wilanour Mine (Goldcorp)

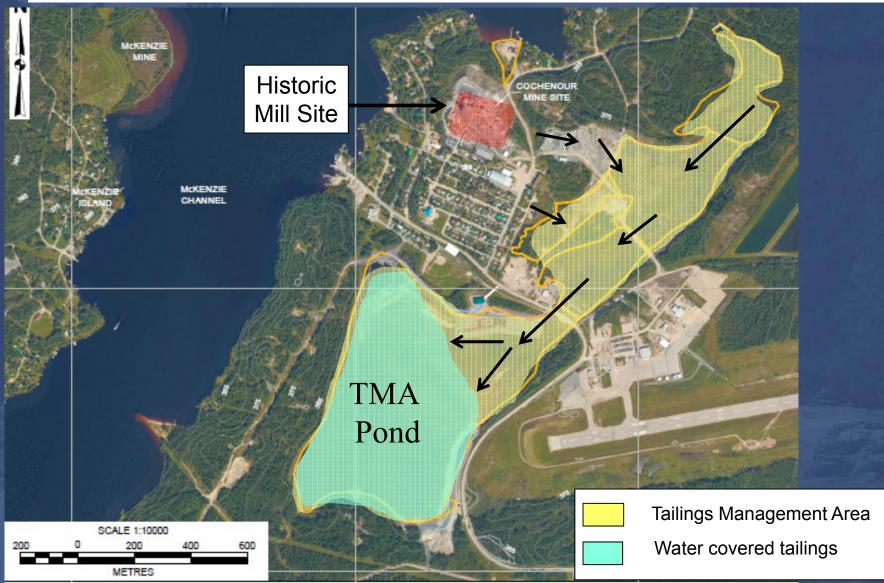


Cochenour Wilanour Mine - Site History

- The milling of gold ores was conducted continuously at the site between 1939 and 1975, and for a brief period in 1982.
- Approximately 2.3 million tonnes of tailings were generated, with ores subjected to roasting and cyanidation.
- Site employs a ferric sulphate treatment system to reduce soluble arsenic in surface waters prior to discharge (Red Lake).
- Project re-activated in 2009, with production scheduled to commence in 2014 (20 year mine life). To supply ore to Campbell Mine.



Historic Tailings Deposition





Study Objectives

- Quantify the various sources of arsenic to the Pond Area (arsenic mass balance); and
- Define the controls governing the remobilization and attenuation of arsenic and other tailings-related parameters.

Water/waste management strategies to:

- Reduce arsenic loadings to Pond Area
- Reduce volume/loading to treatment plant



Study Components

Water Balance

- Climate
- Hydrology
- Water management

Hydrogeology

- Hydrostratigraphy
- Water levels
- Hydraulic conductivity
- Water and As Flux
- 2-D Modelling (Seep/W)

Geochemistry

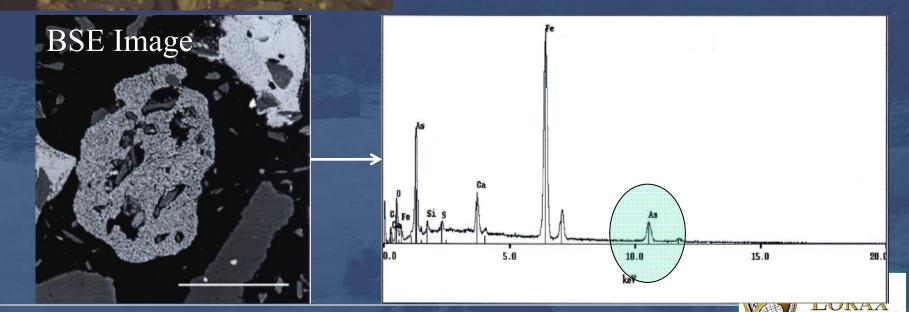
- Surface water quality
- Porewater chemistry
- Mineralogy (Petrography, XRD, SEM, XAS)



Arsenic Source – reductive dissolution of roaster products

Roaster Fe oxides (porous texture hematite)

VIRONME



Arsenic Attenuation Mechanism #1

Diffusive pathway: removal of dissolved As within porewaters of submerged tailings in Pond Area (natural bioremediation).



Pond Area

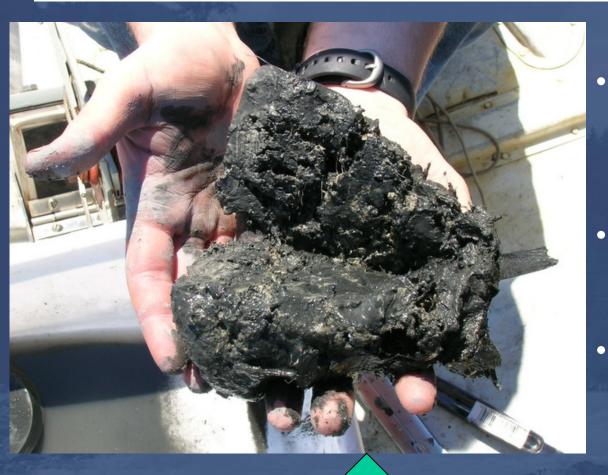
- Submerged tailings deposits
- Passive reclamation since 1975 (35 plus years)
- Cattail, horsetail, sedge
 - Extensive benthic macrophytes

Pond Area – abundant benthic flora





Pond Area - Tailings



quartz, plagioclase, muscovite, chlorite, and dolomite

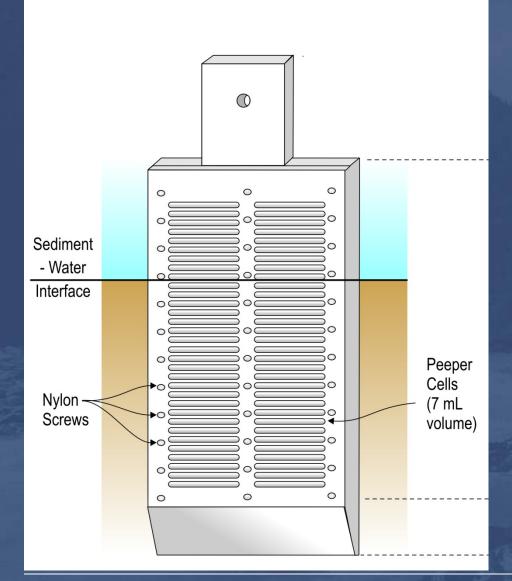
• Sulfides: arsenopyrite, pyrrhotite

 Arsenic rich: 1,000-11,000 mg/kg

Split core showing pervasive root masses in tailings



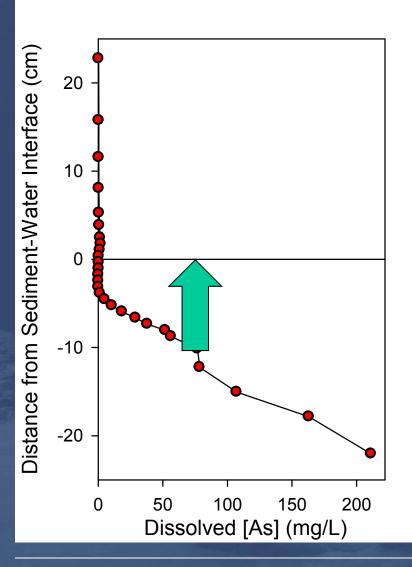
Porewater - Methods



- Dialysis Arrays (peepers)
- 7 mm resolution profiling of dissolved species
- Parameters:
 - Metals/metalloids
 - ➢ pH, alkalinity
 - Redox sensitive species (Fe, Mn, nitrate, H2S)
- Data used to define biogeochemical mechanisms and flux



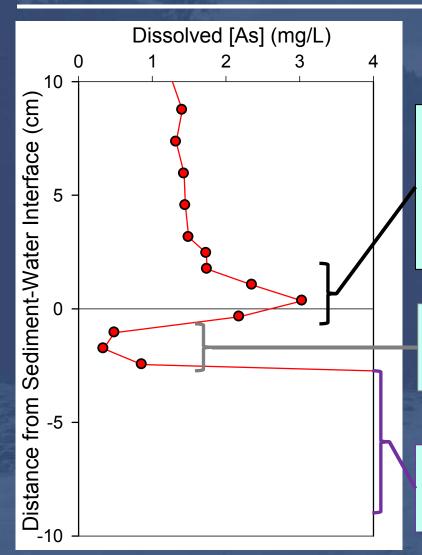
Pond Area – As Porewater Profiles



- Elevated concentration of dissolved As in tailings porewaters (100-300 mg/L)
- Concentration gradient drives large upward diffusive flux towards benthic boundary.



Pond Area – As Porewater Profiles



Three principal mechanisms controlling the mobility of As:

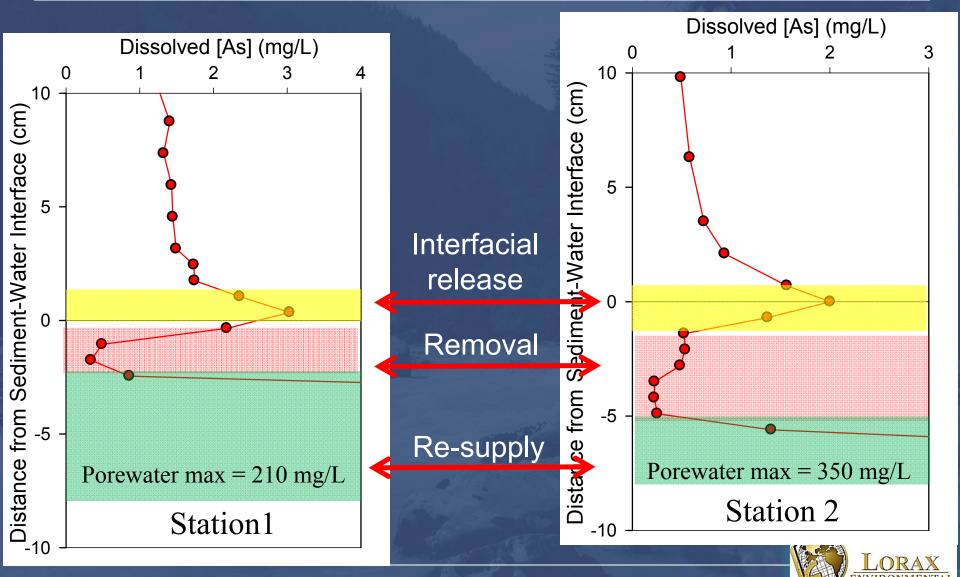
Remobilization via reductive dissolution of Fe-oxides at or near the sediment-water interface

2. Removal from pore solution via *in situ* precipitation.

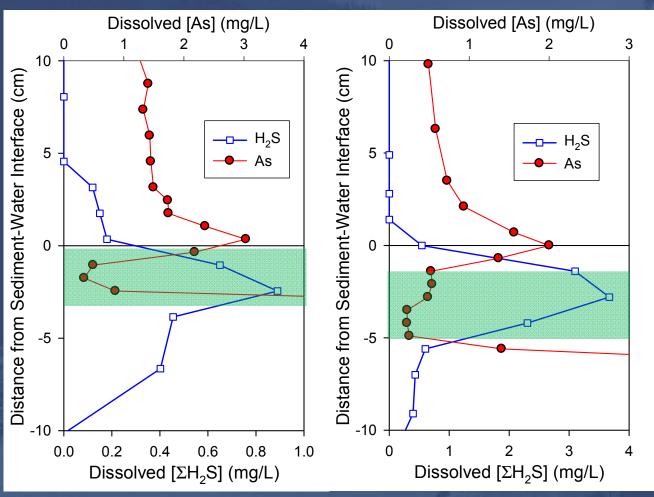
3. Re-supply of dissolved arsenic from deeper tailings horizons.



Pond Area – As Porewater Profiles Spatial homogeneity



Pond Area – Diffusive Sink Mechanism: <u>Precipitation of secondary sulfides</u>



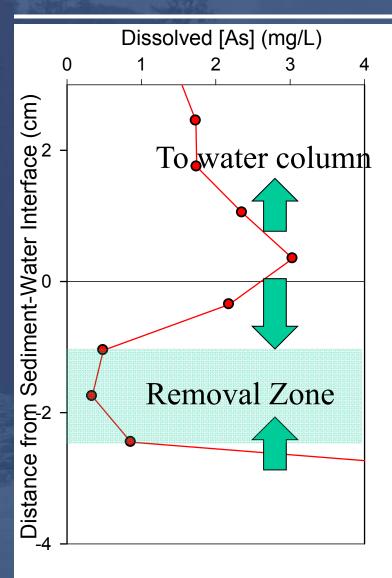
• Secondary phase could not be identified.

Possibilities:

- Discrete As sulfides
- Co-precip with Fesulfides.



Pond Area – Diffusive Sink



- Bi-directional diffusion of dissolved As to localized zone of removal.
- Diffusive sink acts acts to chemically isolate deeper As-rich porewater (200-300 mg/L) from the water column.



Pond Area – As Removal Rate

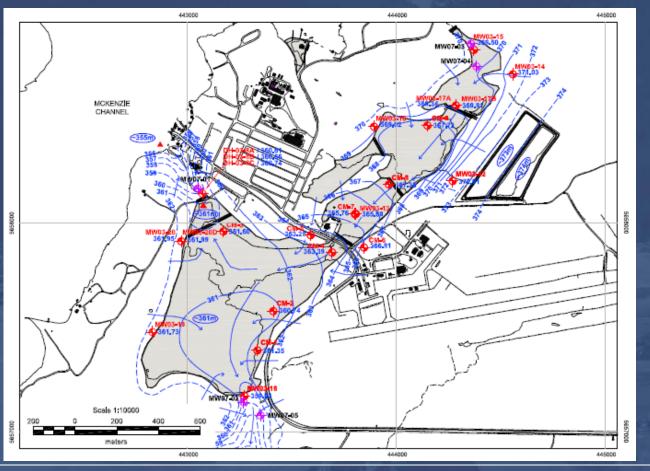
- Per unit area = $13 \text{ mg/m}^2/\text{day}$ (based on mean of 4 profiles)
- Area of Pond = $150,000 \text{ m}^2$
- Annual removal rate = $\sim 700 \text{ kg/year}$
- Current treatment system (ferric salt precipitation):
 - $> ~800,000 \text{ m}^{3}/\text{year} (~25 \text{ L/s})$
 - > Mean [As] feed = 0.7 mg/L
 - > Mean [As] effluent = 0.03 mg/L
 - > Annual removal rate = $\sim 540 \text{ kg/year}$

Passive removal comparable to active treatment



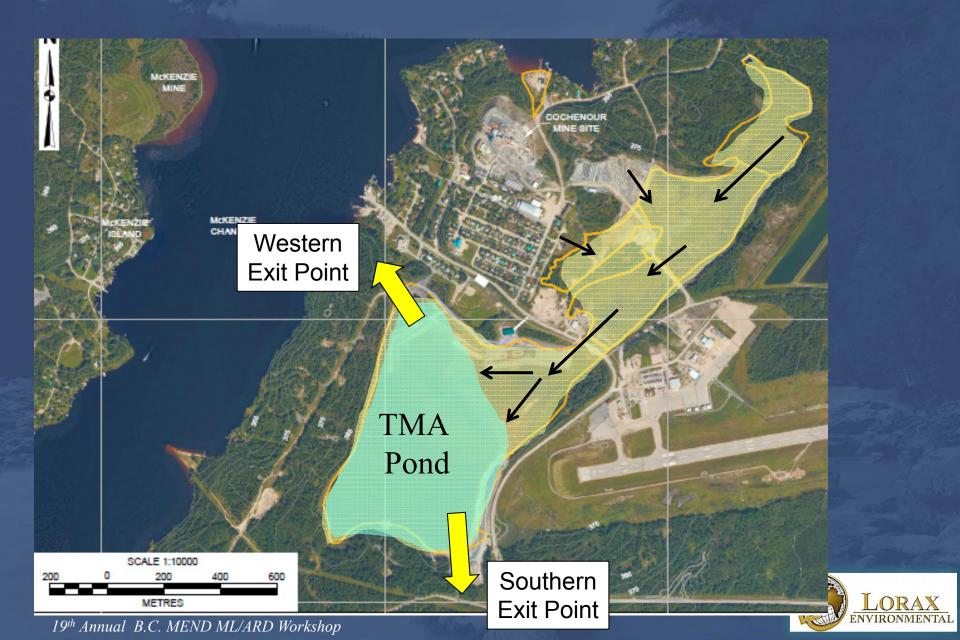
Arsenic Attenuation Mechanism #2

Advective pathway: removal of dissolved As along groundwater flow paths at exit points from the facility (adsorption/precipitation)

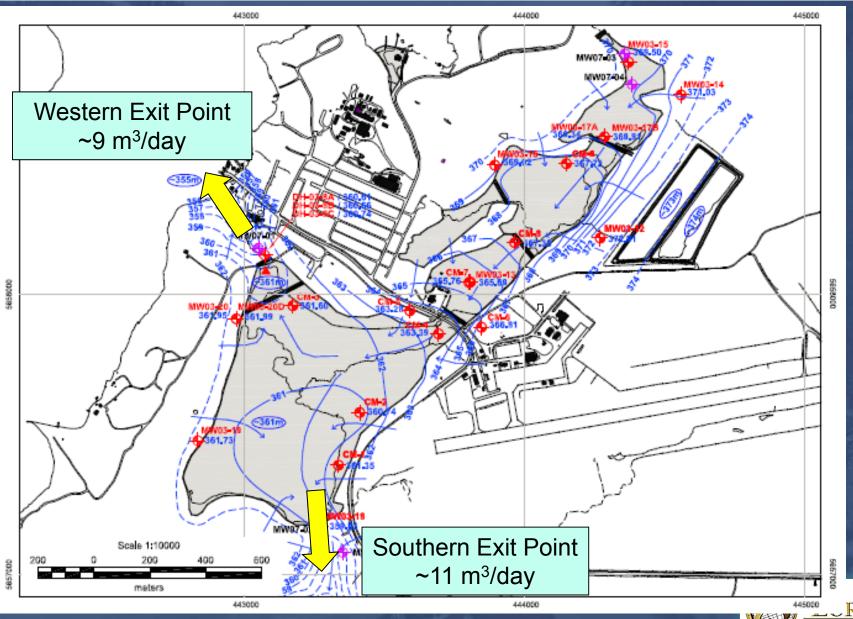




Groundwater Exit Points



Groundwater Elevations and Flux



Attenuation along Groundwater Flow Paths

Exit of tailings-related groundwater based on:

- Hydraulic gradients, permeability and hydrostratigraphy
- 2-D seepage analysis
- Major ion signatures (TDS, Cl, SO_4)

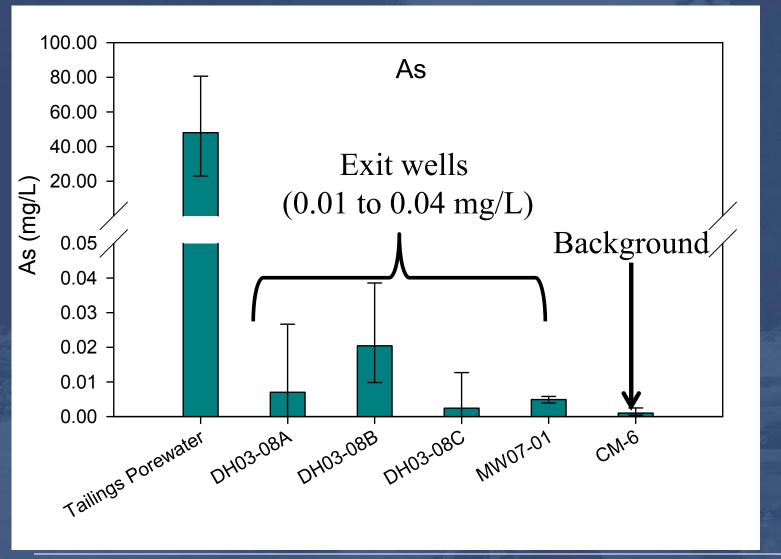
Evidence for attenuation of several parameters along flow path:

- Arsenic
- Ammonia
- Other trace elements

All MOE compliance criteria are being met



Attenuation along Groundwater Flow Paths - Arsenic





Attenuation along Groundwater Flow Paths – Arsenic Removal

Based on:

- Total groundwater flow of 20 m³/day
- Average [As] in tailings porewater = 50 mg/L
- Average [As] in tailings-influenced flow path downgradient of facility= 0.03 mg/L

Annual removal rate = $\sim 370 \text{ kg/year}$



Groundwater Attenuation Mechanism? Clues from Hydrostratigraphy

Tailings Unit (up to 10 m)

Strong likelihood attenuation is associated with either Peat and/or Clay+Silt Unit.

Mechanism not known:

- Adsorption of As(III)?
- Sulfide mineral precipitation?

Peat Unit (1.5 m) Clay + Silt Unit (>1 m)

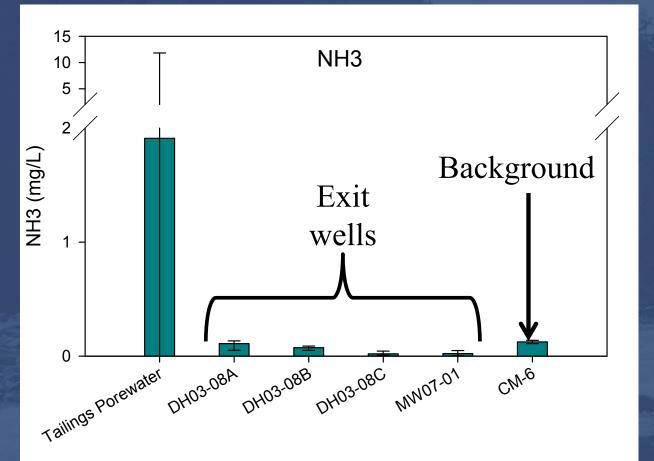
Sand and Silt-Sand Unit (~ 4 m)

Till Unit (> 10m)

Bedrock Unit



Attenuation along Groundwater Flow Paths - Ammonia



NH₄⁺ is strongly adsorbed by clay minerals through cation exchange. Suggests adsorption is important.



Summary & Implications for Tailings Management

Three forms of treatment at Cochenour Wilanour Mine:

- 1. Active treatment (~540 kg As/year)
- 2. Passive bioremediation diffusive pathway (~700 kg As/year)
- 3. Passive removal groundwater pathway (~370 kg As/year)
- Passive removal mechanisms contribute significantly to reducing loadings to surface water receptors, and contribute to lower active treatment costs.



Implications for Tailings Management: Bioengineering for As Stability



- Arsenic mobility is limited under strongly oxidizing and strongly reducing conditions.
- Mobility strongly enhanced under suboxic conditions.
- Practitioners must carefully consider risks associated with bioremediation efforts.



Thank you! Questions?

