

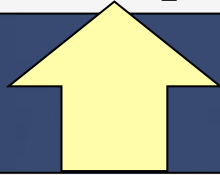
# **The Application of Attenuation Mechanisms in Mine-Related Environmental Assessments: The Spectrum from Conceptual Model to Numerical Implementation**

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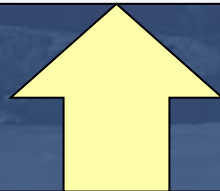
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# The Spectrum:

- Detailed understanding of attenuation process
- Quantifiable
- Included as numerical input in model



- Conceptual understanding of attenuation process
- Not quantified in model
- Presented as “model conservatism”



- Poor understanding of attenuation process
- Not quantified in model
- Possibly presented as “model conservatism”

Degree of Certainty

# Numerical Implementation:

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Attenuation can be expressed numerically in various ways:

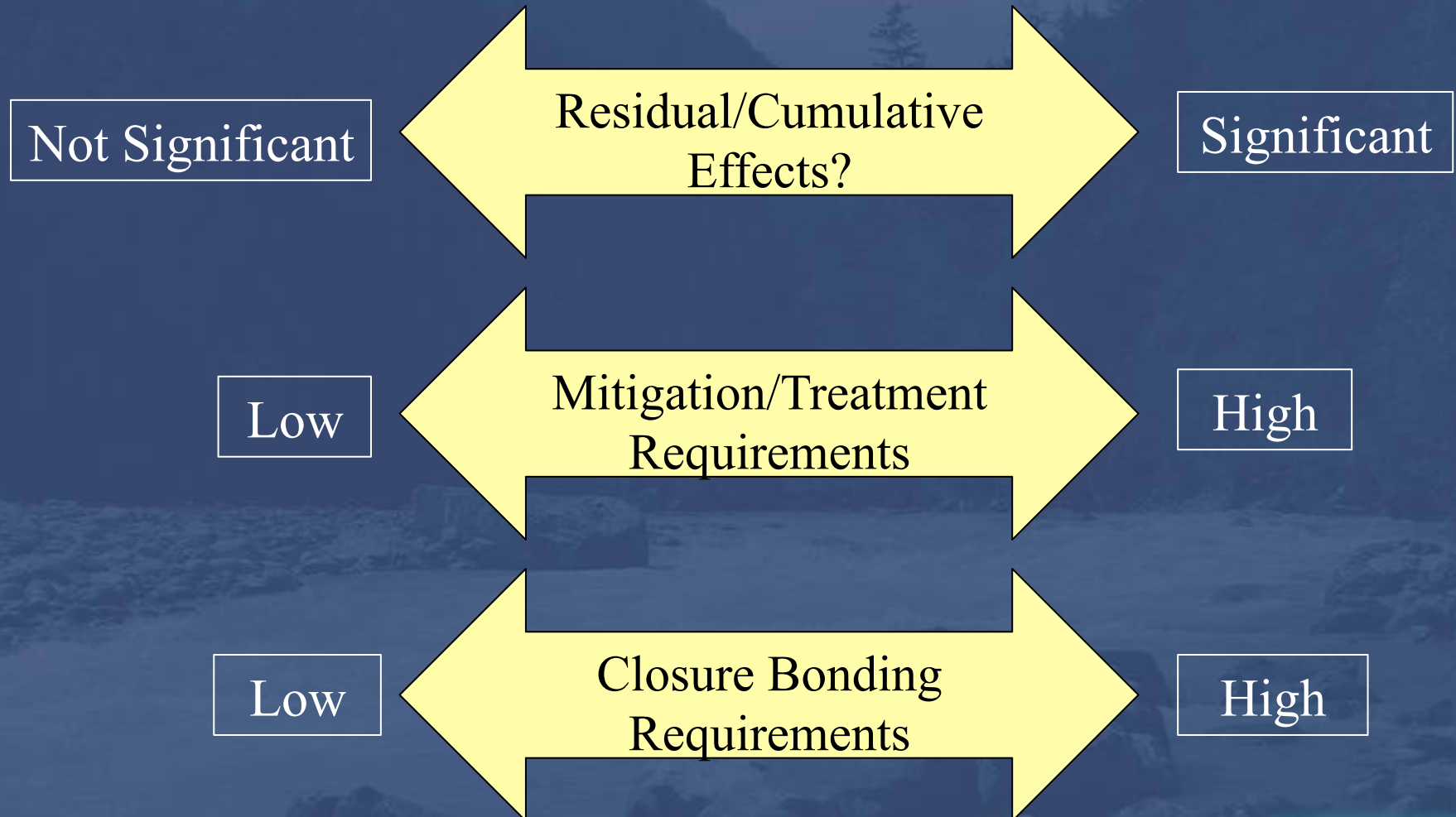
1. Concentration: fixed end-member concentration that is representative of attenuation process (e.g., concentration of Se reporting from saturated waste rock).
2. Reaction rate: rate of removal (mass or concentration) per unit time.
3. Distribution coefficient ( $K_d$ ): for some sorption processes, attenuation can be expressed as a constant ratio (e.g., metal sorption to Fe oxides):

$$K_d = \frac{\text{sorbed metal concentration (mg/kg)}}{\text{dissolved metal concentration (mg/L)}}$$

4. Reactive transport modelling: rate of removal along a flow path (mass or concentration loss per unit time and/or distance).

# Implications for the Inclusion of Attenuation are Significant

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# Attenuation Processes are Complex

- Multiple mechanisms
  - Mineral precipitation
  - Co-precipitation
  - Adsorption
  - Oxidation/Reduction
  - Biological uptake
  - Volatilization
  - Photo-chemical processes
- Complicated by:
  - Scale (bench top to facility)
  - Heterogeneity in mine systems (pathway, material properties)
  - Non-steady-state effects (kinetics, supersaturation)
  - Complexation/speciation
  - Microbial/biological processes
  - Finite processes (e.g., adsorption)

Processes that can vary in  
in time and in space.



# Attenuation Processes Mine Settings: Where Do They Occur?

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- Source zones:
  - Mine rock
  - Pits
  - Tailings
  - Underground workings
- Pathways:
  - Seepage collection systems
  - Sedimentation ponds
  - Saturated rock fills
  - Surface water pathways
  - Groundwater pathways
- Receptors:
  - Streams
  - Lakes
  - Wetlands

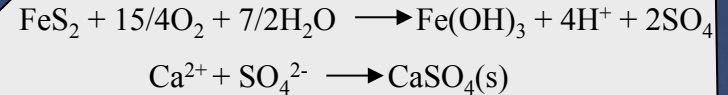
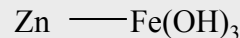


# Attenuation: Unsaturated Waste Rock & Tailings

- Precipitation of secondary phases associated with oxidative weathering:
  - Sulphate minerals (e.g., gypsum)
  - Fe/Mn oxides (e.g., goethite,  $\text{MnO}_2$ )
  - Al hydroxides:  $\text{Al}(\text{OH})_3$
  - Carbonate minerals (e.g., calcite, siderite)
  - Acidic minerals (e.g., jarosite, schwertmannite)

## Trace element adsorption/coprecipitation

- Sorption with hydrous ferric oxides
- Precipitation of oxide minerals

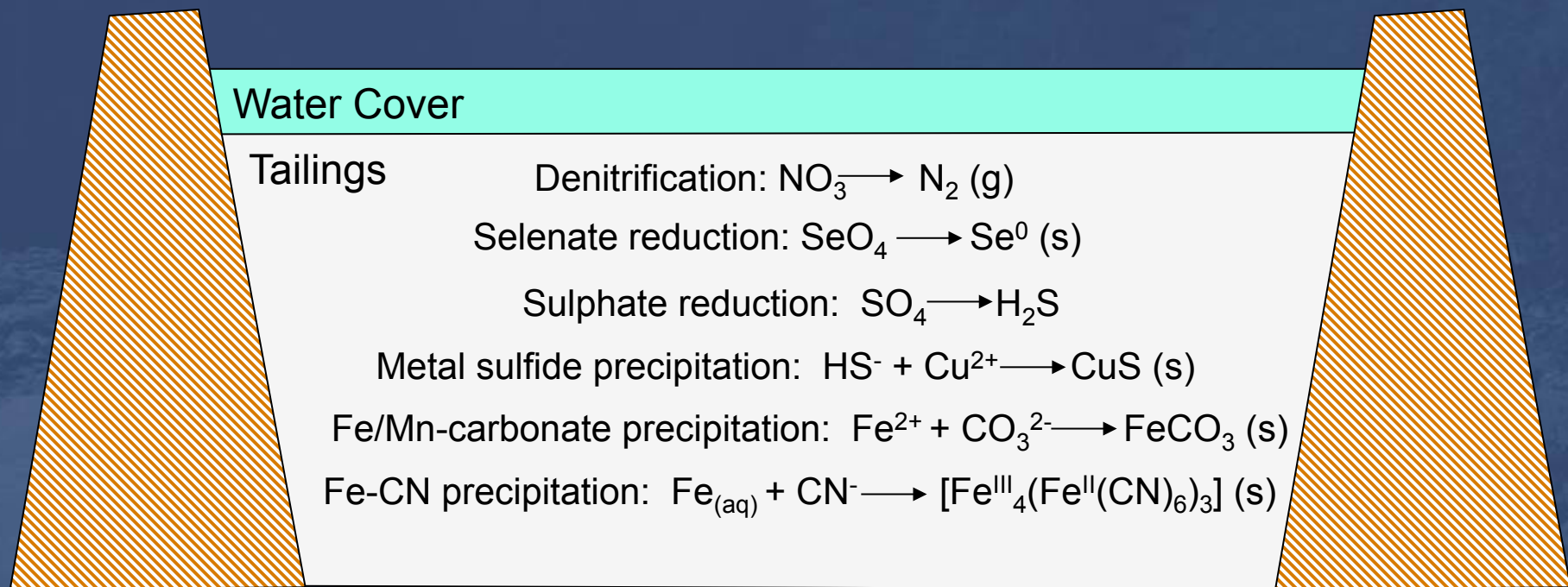


### Suboxic Zones

- Denitrification
- Selenium reduction

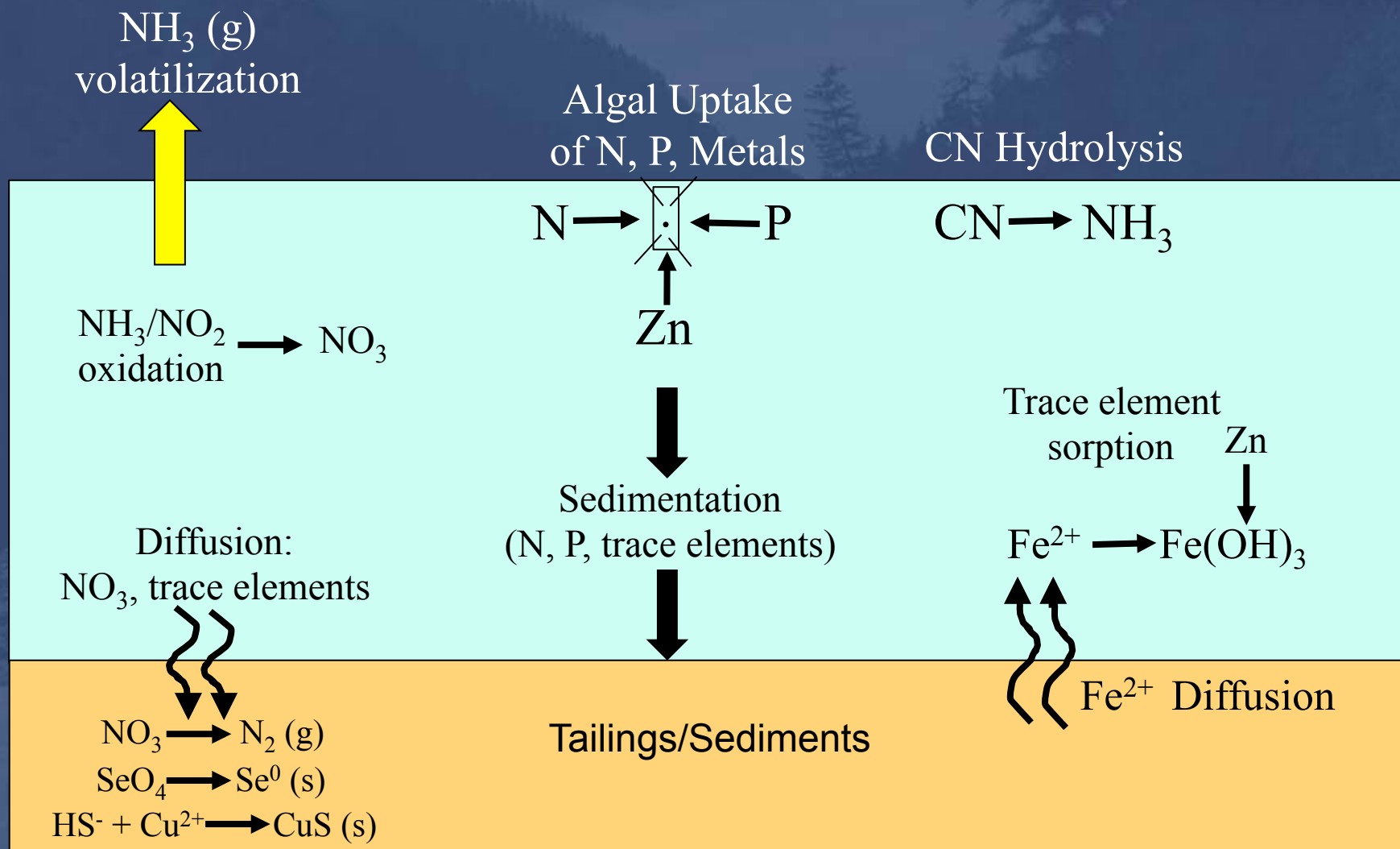
# Attenuation: Saturated Waste Rock and Tailings

- Suboxic redox processes are important.
- At the base of the impoundment where tailings and waste rock are in contact with remnant vegetation and/or organic soils, there is the potential for more reducing conditions to develop.

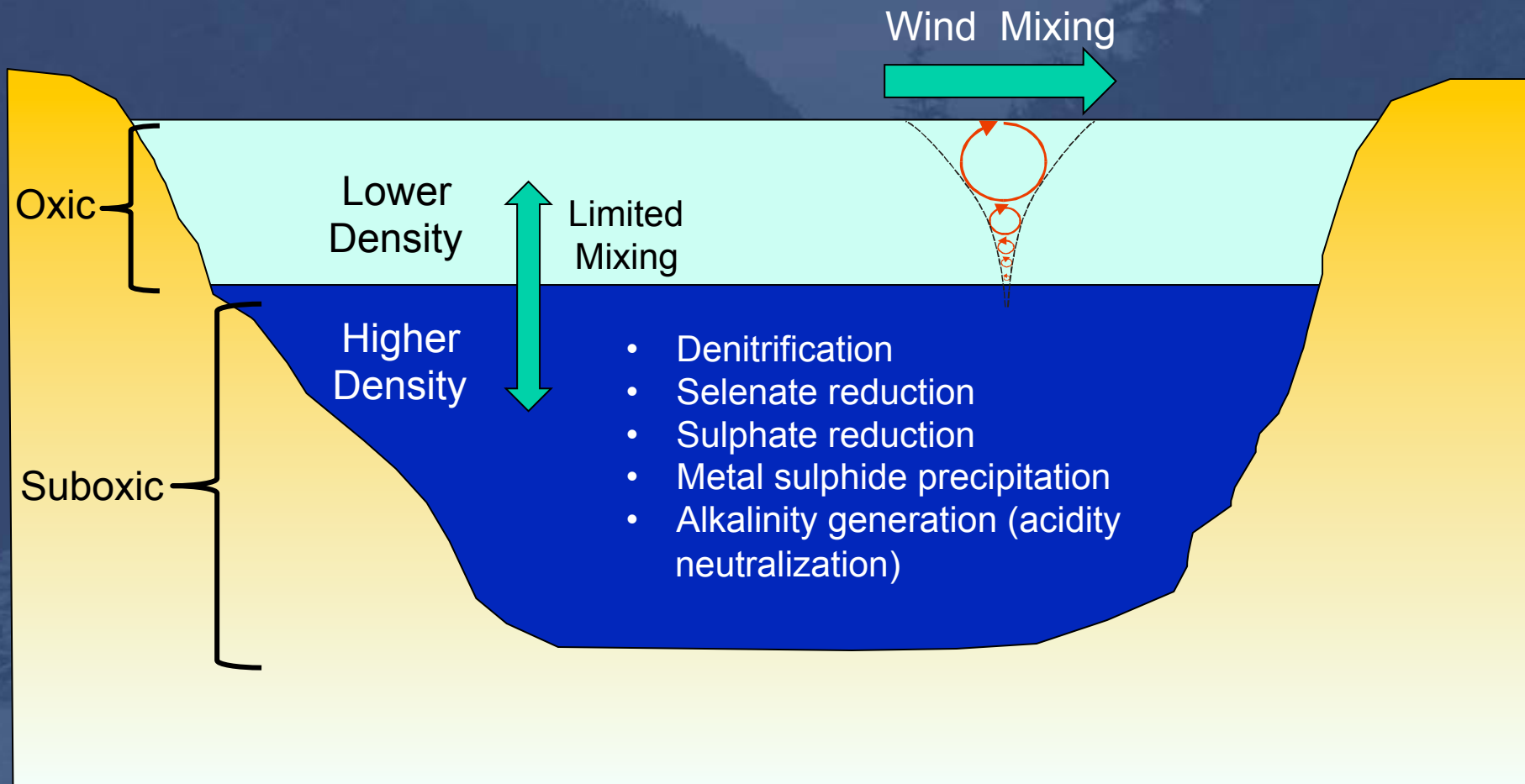




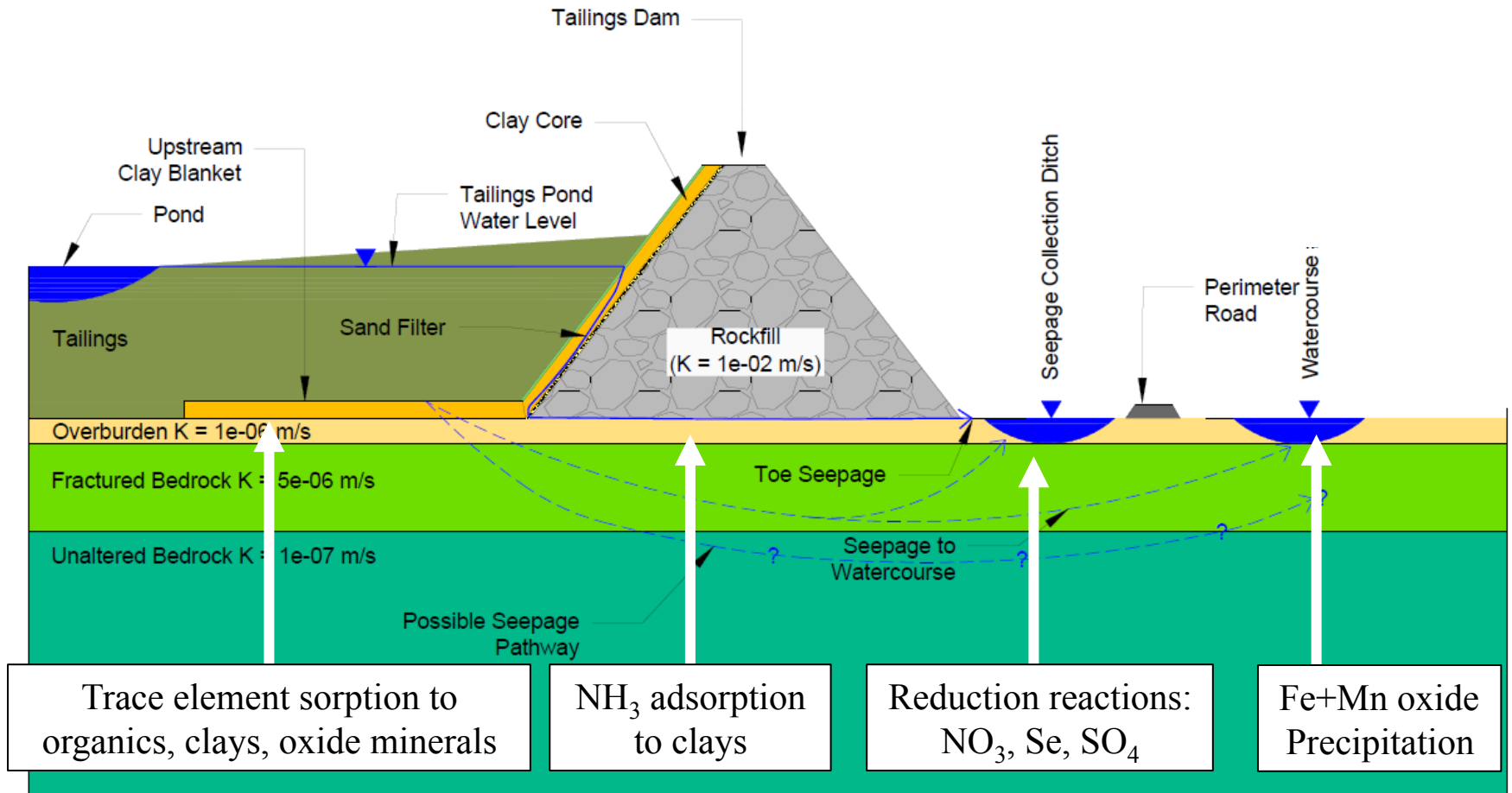
# Attenuation: Tailings Ponds, Sediment Ponds & Lakes



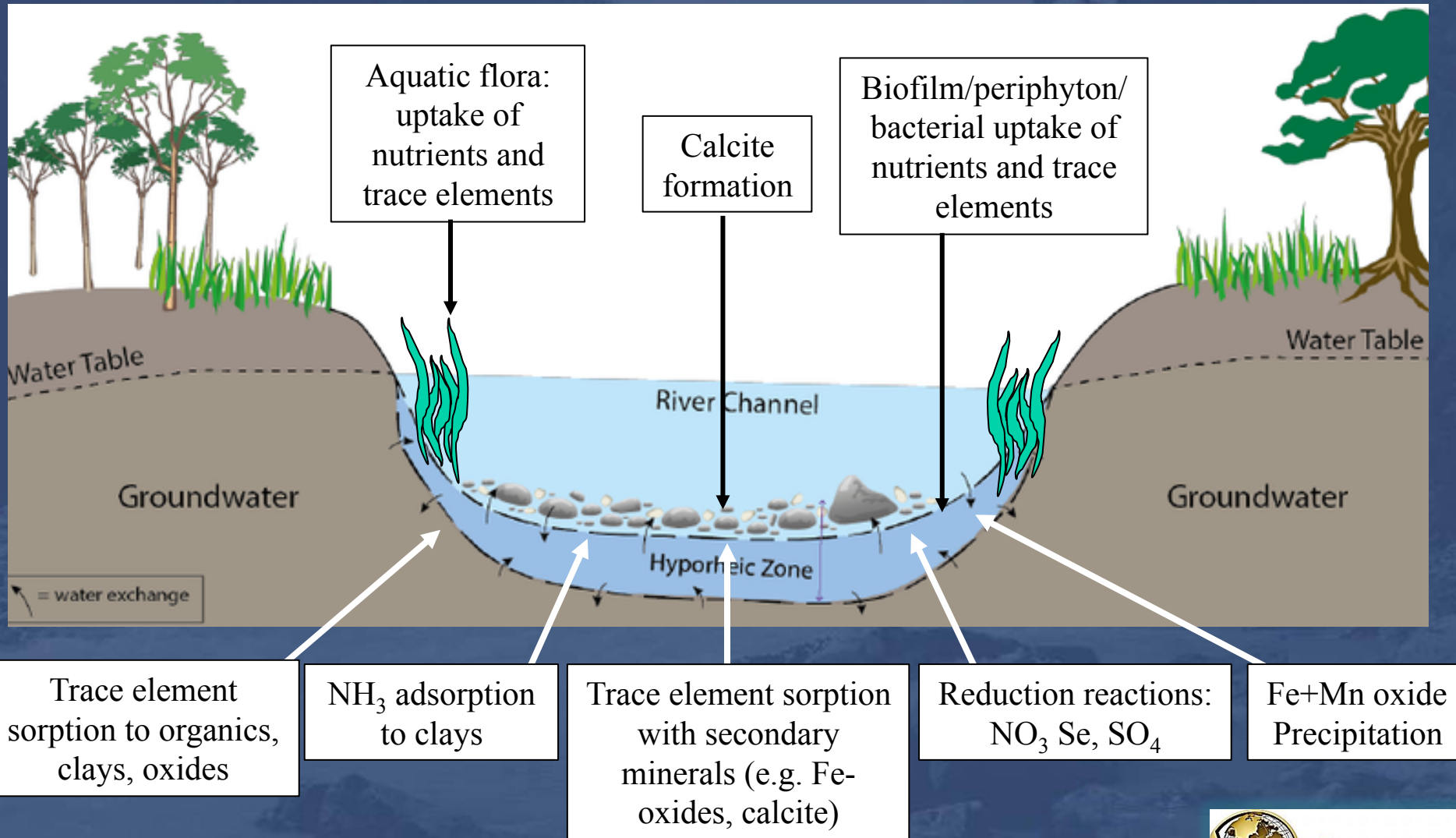
# Attenuation: Stratified Pit Lakes



# Attenuation: Subsurface Flow Pathways



# Attenuation: Streams



# Attenuation Process: 3 Groupings

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## High Confidence:

**Numerical implementation warranted based on First Principles and limited site-specific information**

## Moderate Confidence

**Numerical implementation requires some degree of site-specific evaluation**

## Low Confidence

**Numerical implementation requires high degree of site-specific evaluation**



# Some Subjectivity Involved: The banjo paradigm

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Maybe next  
year he'll learn  
to play to  
fiddle.....



# Attenuation Process – High Confidence

## Numerical implementation warranted based on First Principles and limited site-specific information

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- Particle settling (ponds, lakes)
- Phosphorus uptake: biological uptake in ponds, lakes, pit lakes.
- CN degradation (tailings ponds)
- $\text{Fe}^{2+}$  oxidation/hydrolysis
- $\text{H}_2\text{S}$  oxidation
- Mineral precipitation at circum-neutral pH for certain major mineral phases (e.g., anhydrite/gypsum, barite, Fe oxides)
- Denitrification: Saturated waste rock, saturated tailings, lake sediments
- Selenium bio-reduction: saturated waste rock, saturated tailings, lake/wetland sediments
- Lake surface waters: Trace element removal via biogenic scavenging (Cu, Cd, Ni, Zn).

rapid oxidation kinetics

# Attenuation Process – Moderate Confidence

## Numerical implementation requires some degree of site-specific evaluation

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- Ammonia oxidation
  - Nitrite oxidation
  - $\text{Mn}^{2+}$  oxidation
  - Ammonia biological uptake
  - Ammonia adsorption along clay-bearing groundwater pathways
  - CN removal along groundwater pathways
  - Metal sulphide precipitation in lake sediments, submerged tailings
  - Calcite formation in ditches, streams
  - Sulphate reduction: saturated tailings, lake/wetland sediments
  - Alkalinity generation and acid neutralization in suboxic sediments and pit lakes.
  - Co-precipitation of certain trace elements (radium with barite)
- Variable (slow) oxidation kinetics



# Attenuation Process – Low Confidence

## Numerical implementation requires high degree of site-specific evaluation

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- Metal attenuation in low-pH systems (e.g., acidic sulfate minerals)
- Trace element adsorption/co-precipitation along groundwater flow paths
- Stream attenuation processes: trace element adsorption, biological uptake, metal co-precipitation with secondary minerals
- Sulphate reduction along groundwater pathways
- Sulphate reduction in stratified pit lakes
- Metal sulphide precipitation in stratified pit lakes
- Volatilization from surface water systems (e.g., Se, Hg)

# What are the tools? – Scientific Method

- Environmental baseline investigations:
    - Geology, Hydrogeology, WQ, Hydrology, etc.
  - Kinetic test work (lab scale, pilot scale, field scale)
  - Adsorption testwork
  - Mineralogy: microscopy, extractions, etc.
  - Natural analogues
  - Mine site analogues
  - Geochemical modelling
  - Reactive transport modelling
- Pre-Mining
- Operational monitoring data (site specific)
    - Waste rock seepage/porewater
    - Tailings seepage/porewater
    - Groundwater quality and pathways
    - Surface water quality and pathways
    - Model refinement
- Operations

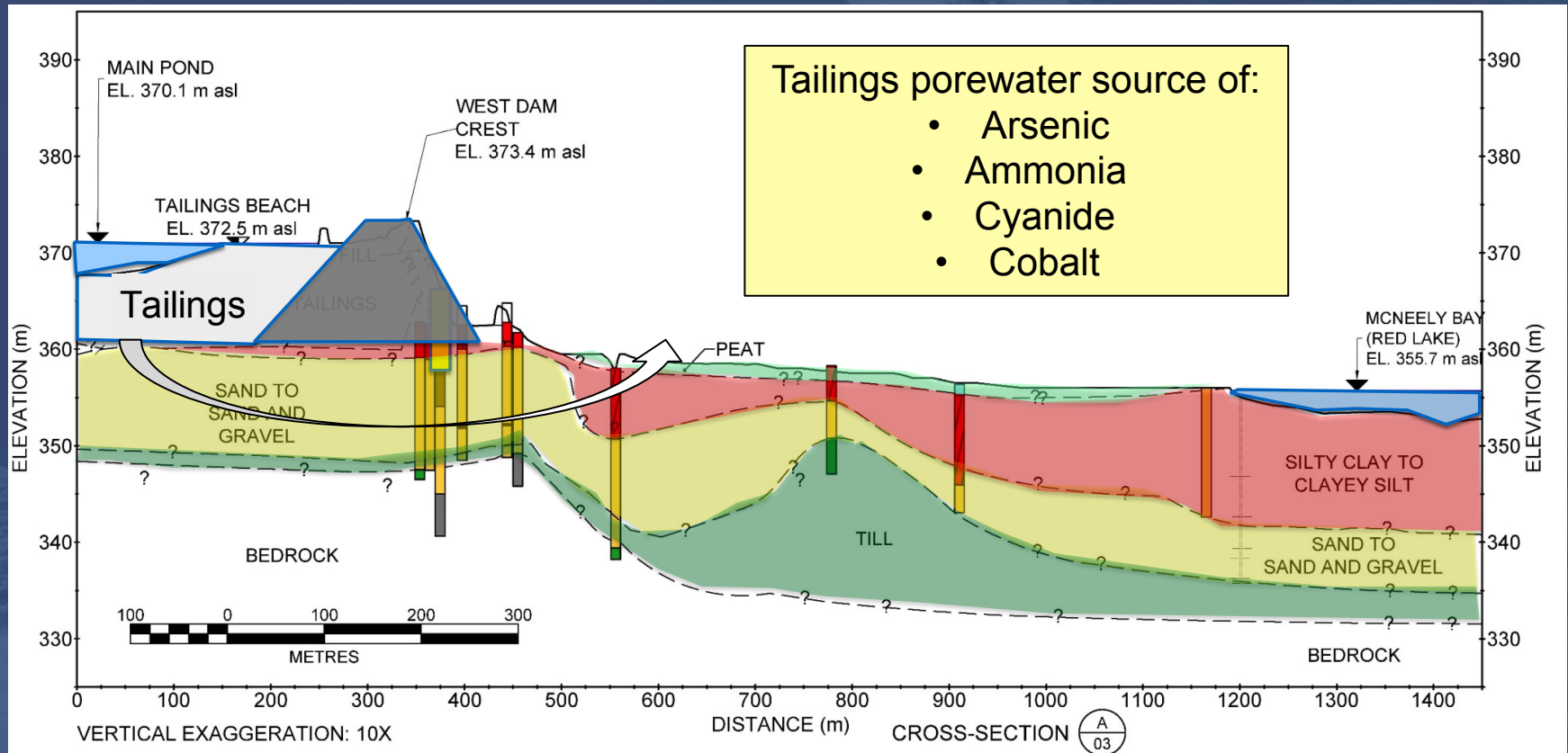


# A Cautionary Warning: Cobalt case study - Red Lake Gold Mines

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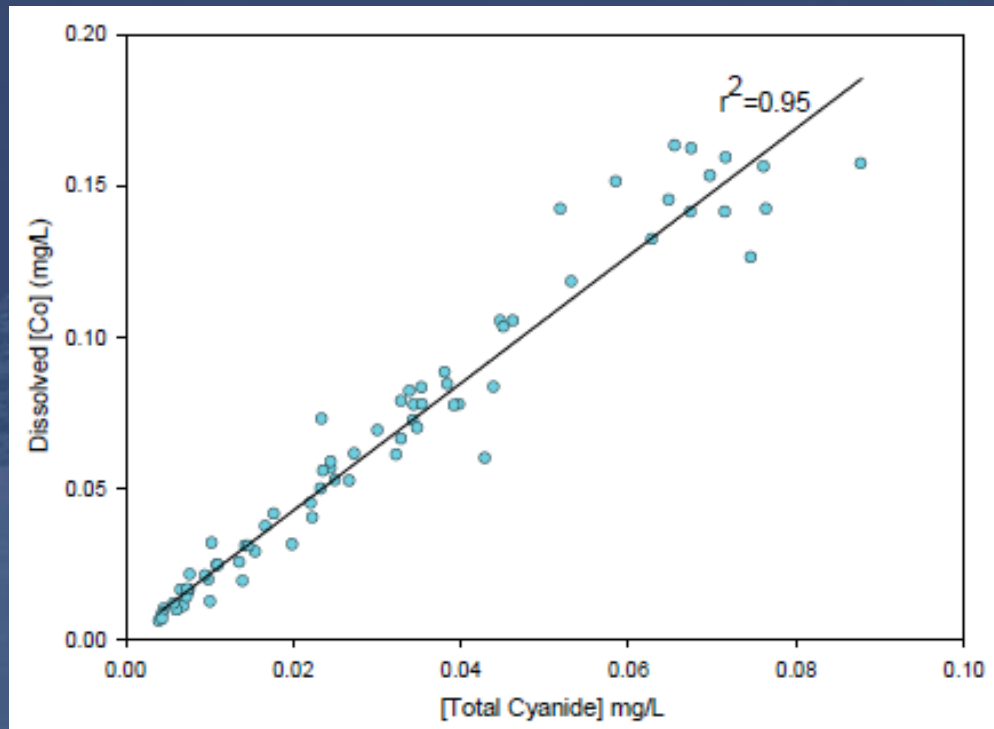


# A Cautionary Warning: Cobalt case study - Red Lake Gold Mines



# A Cautionary Warning: Cobalt case study - Red Lake Gold Mines

- Arsenic, ammonia and cyanide show pronounced attenuation along groundwater flow paths (interaction with peat and clayey-till).
- In contrast, cobalt shows conservative behaviour in groundwater.
- In bench-scale permeable reactive barrier (PRB) testwork, cobalt removal is limited, even in presence of abundant free sulphide.



- Unpredictable behaviour of Co can be explained by its presence as strong CN-complexes (as inferred by Co speciation testwork).

# Summary

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- Attenuation mechanisms are varied and complex.
- Conceptual models represent an effective way to illustrate pathways and attenuation processes.
- The inclusion of numerical attenuation factors in “base case” water quality prediction models will be highly parameter specific and site-specific.
- For many attenuation processes, a high-level of site-specific evaluation will be required to permit numerical application.
- Expect the unexpected.



# Acknowledgements

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# Thank You!

