2021 BC MEND ML/ARD Workshop

Characterization Studies of the Waste Rock Dumps at the Bingham Canyon Mine, Utah, USA

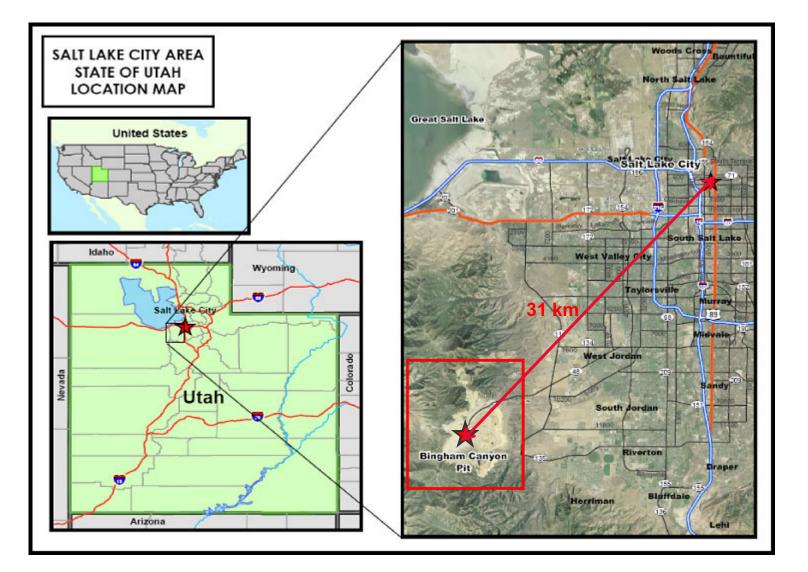
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Location



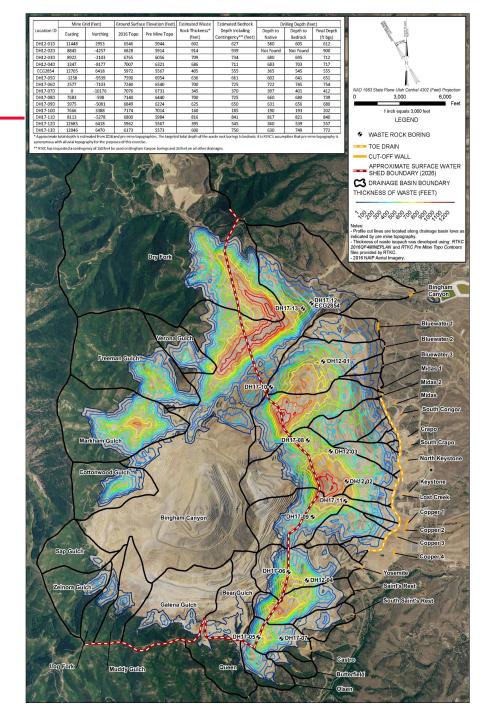
Bingham Canyon Mine



- Porphyry copper
- Deposit discovered in 1846
- Mining commenced in 1864 and open pit mining in 1903
- 2,000 km of underground workings
- Waste rock dumped using rail and then trucks
- Conventional Cu recovery, but also Mo, Au, Ag and PGM
- Waste rock dumps on east, south and west sides and large Bingham Canyon / Dry Fork dump

Waste Rock Dumps

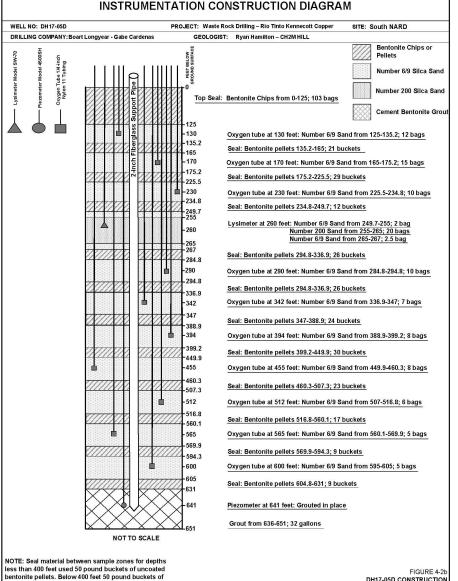
- Greater than 2000 ha
- 5.4 billion tonnes
- Selected dumps actively leached from the 1930s until 2000
- 12 paired and one single borehole (25 total) placed in six dumps
- Pairs of boreholes placed in dumps one near valley facing crest and the other much further back near water boundary
- Acidity arises from oxidation of sulfide minerals (pyrite) and moderated by jarosite
- Some neutralisation via calcite and aluminosilicate minerals
- Collection system used to intercept waste rock seepage that is neutralised in tailings line



RioTintc

Installation of Boreholes

- One deep borehole (to bedrock up to 275 m) and one shallow hole (30 m) at each location, except one
- Instrumentation included suction lysimeters, ٠ vibrating wire piezometers, gas (oxygen and carbon dioxide) tubes, direct temperature sensing fibre optic cables
- Temperature measured at a spacing of 1 m ٠ substantial improvement on measuring using thermistors
- Gas measurements made at multiple depths in ٠ both deep and shallow boreholes with one measurement at same depth in the deep and shallow borehole at the same location
- Water guality measurements taken on multiple occasions
- Core analysed visually for lithology, state of oxidation and moisture content. Core composites analysed for mineralogy using QEMSCAN. All core was also analysed using CoreScan



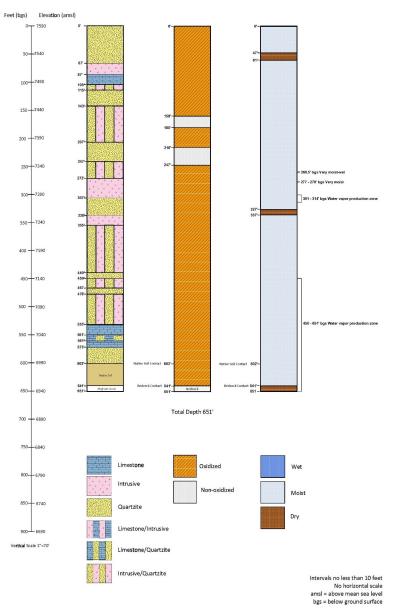
RTKC CLOSURE PFS GEOCHEMICAL FIELD DATA COLLECTION REPORT

RIO TINTO KENNECOTT COPPER, UTAH

bentonite nellets. Below 400 feet 50 nound buckets of coated bentonite pellets were used. Filter pack sand used 50 pound bags.

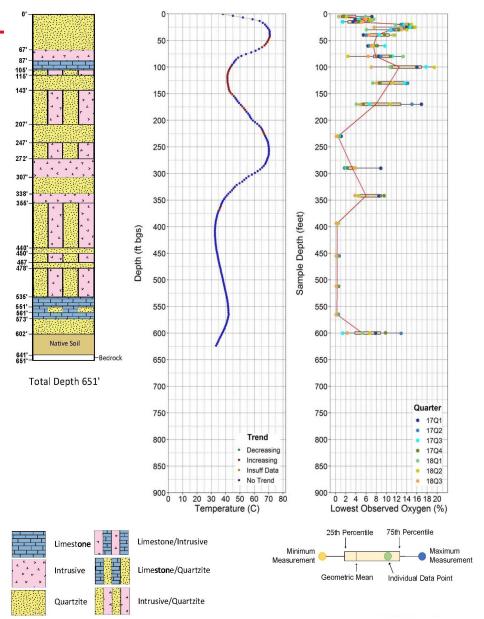
Visual Analysis of Borehole Core

- Core often contained each lithology although leached dump cores contained little limestone (not placed in these dumps), sometimes as individual and sometimes mixed sequences
- Extent of oxidation was quite variable across the boreholes
- There were sections of the core that were moist and some that were dry
- Almost no sections of the core was wet



Characterisation of Drill Cores

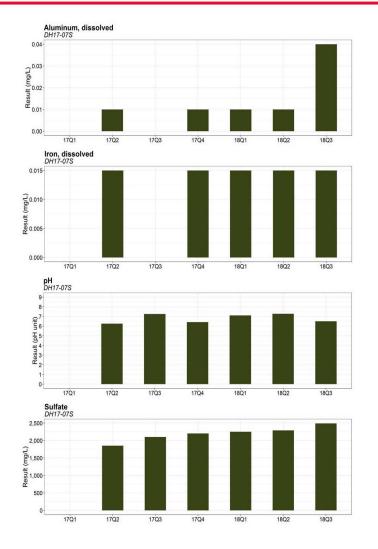
- Three main lithologies at the Bingham Canyon mine – quartzite, intrusive (monzonite), limestone
- Temperature measurements
 recorded using fibre optic sensors
- Significantly elevated temperatures within the dumps (up to 80 °C)
- Large number of oxygen measurements taken at each location and depth
- Oxygen supplied into the dumps by both convective and diffusive mechanisms
- Significant oxygen concentrations observed deep into the waste rock dumps



In Situ Water Quality

Aluminum, dissolved DH17-06S 600 Result (mg/L) 200 17Q1 17Q2 17Q3 17Q4 18Q1 18Q2 18Q3 Iron, dissolved DH17-06S 400 Result (mg/L) 000 100 0-17Q1 17Q2 17Q3 17Q4 18Q1 18Q2 18Q3 **pH** DH17-06S 9-8-Result (pH unit) 1-0-17Q1 17Q2 17Q3 17Q4 18Q1 18Q2 18Q3 Sulfate DH17-06S Result (mg/L) 2,000 17Q2 17Q3 17Q4 18Q3 17Q1 18Q1 18Q2

Temporal Change in Water Quality



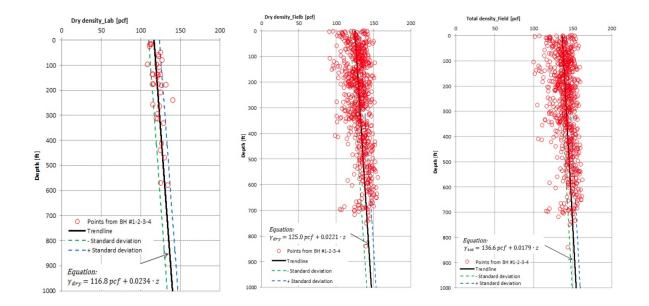
Minimal Change in Water Quality

Mineralogy of Drill Core

Mineral	Median Observed Mineralogy of Unleached Waste Rock Dumps (wt%)	Median Observed Mineralogy of Leached Waste Rock Dumps (wt%)			
K-feldspar	11.74	11.07			
Anorthite	1.41	1.09			
Albite	3.82	2.94			
Muscovite	3.99	4.90			
Kaolinite	0.78	1.24			
Chlorite	1.94	2.46			
Pyroxene (diopside)	0.72	1.11			
Amphibole (tremolite)	1.78	0.55			
Talc	1.45	1.06			
Biotite (phlogopite)	2.81	1.09			
Garnet (grossular)	0.30	0			
Calcite	0.93	0.21			
Gypsum	0.06	0.40			
Pyrite	3.62	3.86			
Quartz	63.1	64.8			
Wollastonite	0	0			
Jarosite	0.69	0.59			
Iron oxides (Fe(OH) ₃)	1.03	0.88			

- Dump mineralogy divided into that characteristic of an unleached dump and of a leached dump
- Mineralogy of the two characteristic dumps is quite similar
 - Lesser amounts of more reactive neutralising minerals
- Mineralogy can be utilised to estimate an initial (as dumped) mineralogy
 - Ore contains minimal amounts of iron oxide (Fe(OH)₃) and jarosite which are products from the oxidation of sulfide minerals (pyrite)

Bulk Density and Specific Gravity



Bulk density: At dump surface = 2.002 g/cm^3 Average = 2.069 g/cm^3 Specific gravity (Gs20°C) = 2.78 g/cm^3

Specific gravity (mineralogy) = 2.76 g/cm³

Measurement of Oxidation Rate and Oxygen Flux

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Field Measurement of Gas Flux

- O₂ consumption
- CO₂ production

Laboratory Measurement of Oxidation Rate

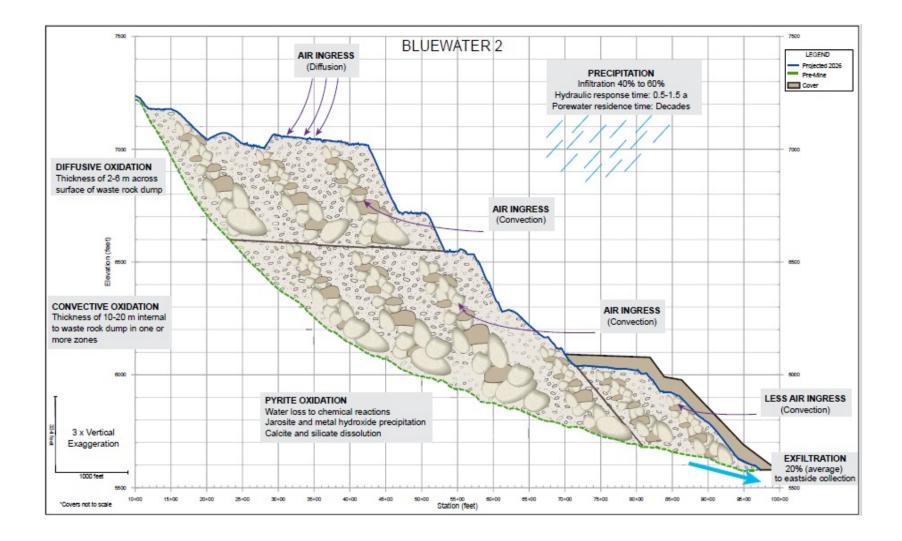


Measurement of Pyrite Oxidation Rate

Lithology	Measurement	Oxidation		Std		Maximum	Minimum	Sulfate Release
or Waste Rock	Туре	Rate kg(O₂)/m³/s	log(Rate)	Dev.	Count	kg(O ₂)/m³/s		Rate mg(SO₄)/kg/a
Quartzite	Laboratory	3.98 x 10 ⁻⁸	-7.40	0.40	31	2.24 x 10 ⁻⁷	7.28 x 10 ⁻⁹	1051
Intrusives	Laboratory	3.46 x 10 ⁻⁸	-7.46	0.54	21	5.38 x 10 ⁻⁷	5.43 x 10 ⁻⁹	914
Limestone	Laboratory	3.25 x 10 ⁻⁸	-7.49	0.85	11	1.49 x 10 ⁻⁶	4.26 x 10 ⁻⁹	858
Waste Rock	Laboratory	4.42 x 10 ⁻⁸	-7.35	0.43	13	3.10 x 10 ⁻⁷	1.39 x 10 ⁻⁸	1168
Waste Rock	O ₂ Concentration	5.12 x 10 ⁻⁸	-7.29	0.80	13	8.59 x 10 ⁻⁷	8.03 x 10 ⁻¹⁰	1353
Waste Rock	O ₂ Flux	1.23 x 10 ⁻⁸	-7.91	0.49	24	5.57 x 10 ⁻⁸	3.58 x 10 ⁻¹⁰	326
Waste Rock	O ₂ Flux	2.66 x 10 ⁻⁸	-7.57	1.07	7	4.60 x 10 ⁻⁷	8.20 x 10 ⁻¹⁰	703
Waste Rock	Natural Convection	2.37 x 10 ⁻⁸	-7.63	1.04	9	4.25 x 10 ⁻⁷	6.11 x 10 ⁻¹⁰	625
All	All	2.93 x 10 ⁻⁸	-7.53	0.64	129	1.49 x 10 ⁻⁶	3.58 x 10 ⁻¹⁰	773

- Rate determined using multiple techniques
 - o Laboratory measurement of all three major lithologies
 - Analysis of depth of oxygen ingress from surface
 - \circ $\,$ Oxygen consumption measurements in chambers placed on dump surfaces
- Measurement values ranged over nearly four orders of magnitude by all techniques and for all lithologies
- Geometric average similar for all techniques and for all lithologies: overall geometric average of 2.93 x 10⁻⁸ kg(O₂)/m³/s

Conceptual Gas Transport Model



Coarse / Fine Zone Segregation

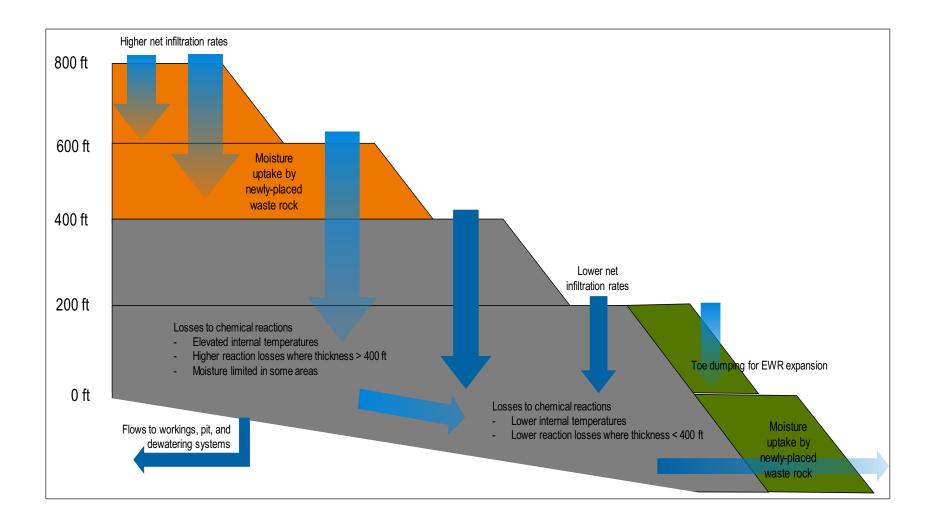


• Coarse zones (clast supported) bounded by fine zones (matrix supported)

Convective Basal Air Flow

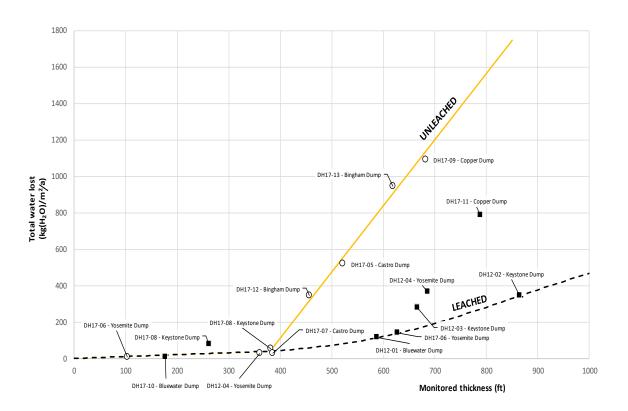


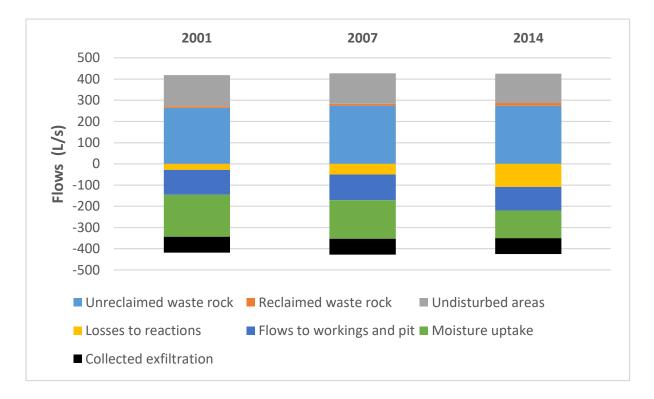
Hydrogeological Conceptual Model



Water Loss to Chemical Reactions

- Differing amount of water loss to chemical reactions depending on whether or not dumps were leached
- Leached dumps are cooler due to significant flux of leach water through the dumps
- Cooler temperatures results in reduced amount of oxidation which leads to reduced consumption of water





- Significant losses to chemical reactions
- Also substantial moisture uptake in fresh waste rock placed in dumps
- Some flows from south and west dumps intercepted within underground workings and pit
- Collected exfiltration equals about 75 L/s

Conclusions

- Paired borehole placement in waste rock dumps and associated instrumentation enabled:
 - Characterisation of geochemical and hydrogeological mechanisms occurring in the dumps that lead to generation of acidity.
 - Demonstration that gas (oxygen) supply to the dumps occurs by both convective (predominant) and diffusive mechanisms.
 - High temperatures (up to 80 °C) leads to considerable water consumption within the dumps.
- Requirement to develop long-term understanding of seepage evolution from the Bingham Canyon mine dumps:
 - Development of long-term management strategies during operation.
 - Program of work undertaken has involved a significant number of innovative characterisation aspects aiding the development of the understanding.
 - Highlights what is needed to be undertaken at mine sites with long operational histories.